



哈爾濱工業大學
HARBIN INSTITUTE OF TECHNOLOGY

High-temperature geopolymer & C_f reinforced composites and their derived CMC

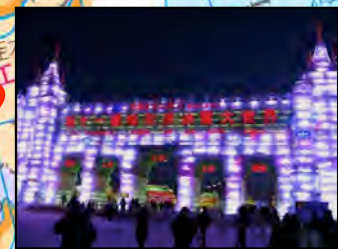
Dechang Jia, Peigang He



Institute for Advanced Ceramics

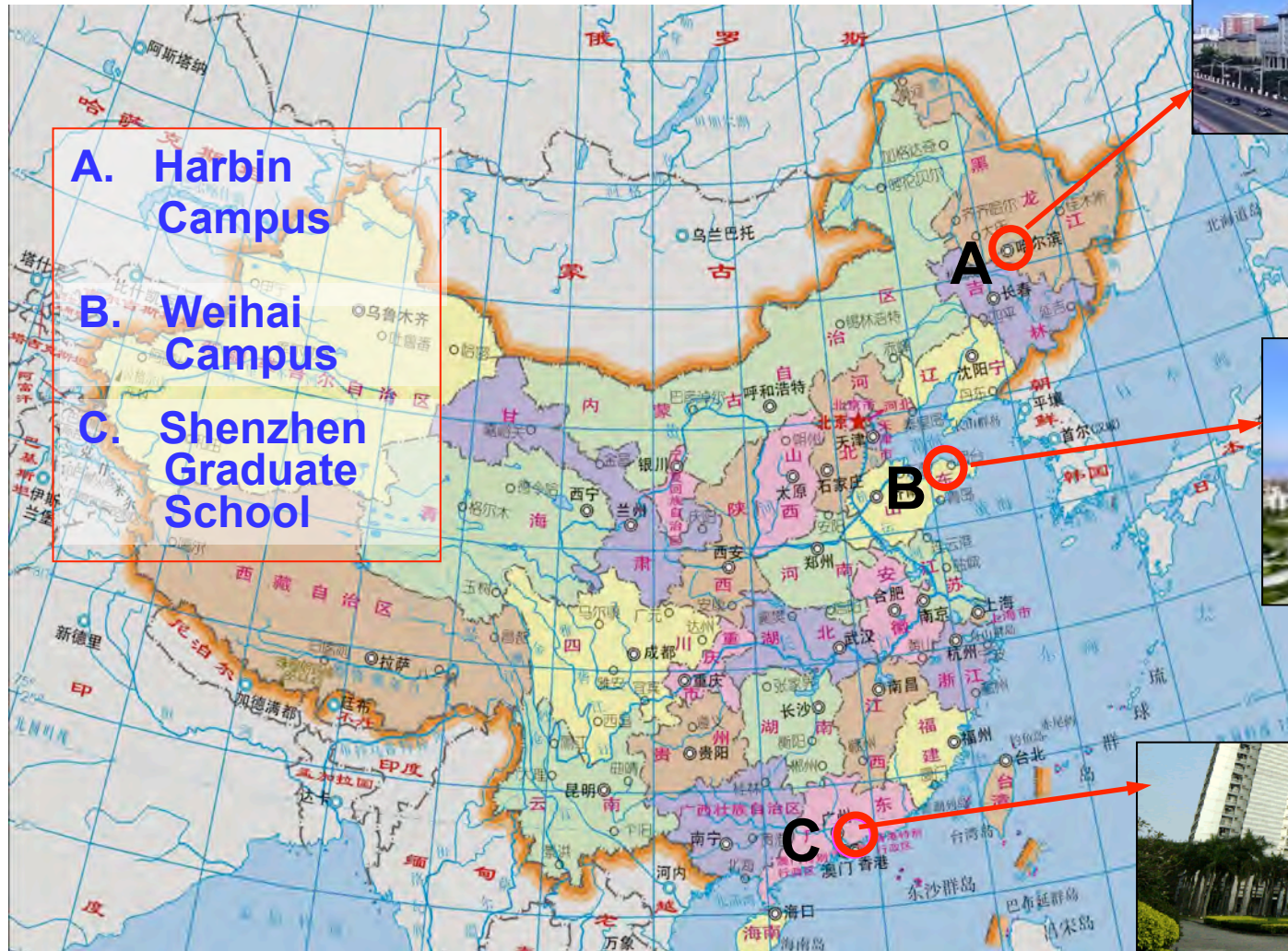
Harbin Institute of Technology, Harbin, P.R. China

Saint-Quentin, France • July 4~6th, 2011

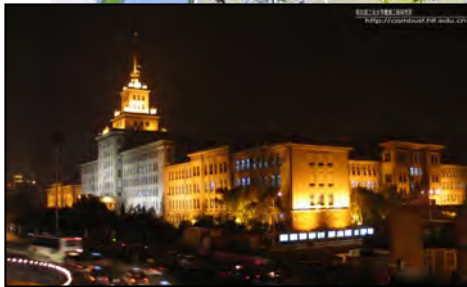
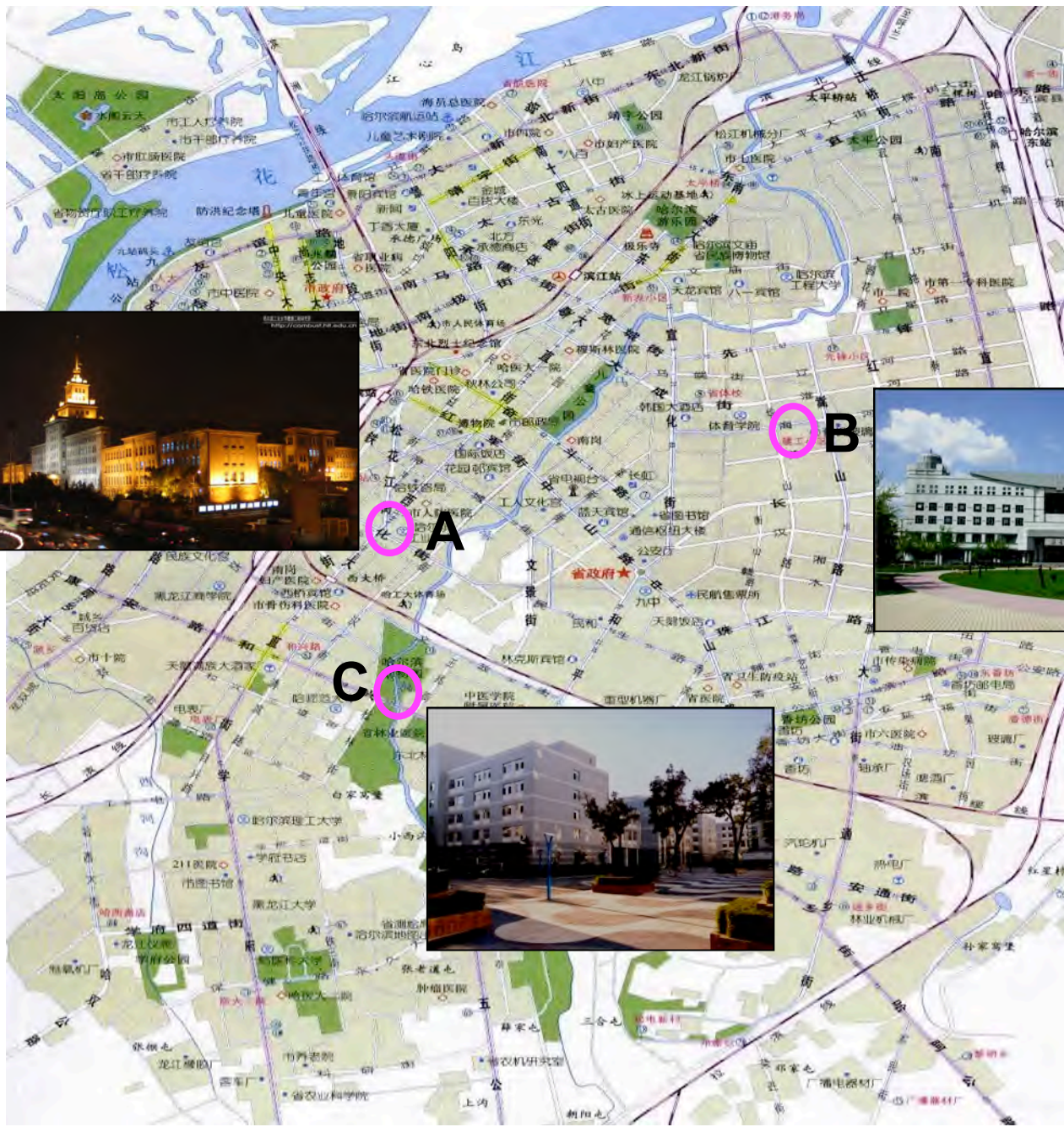


Three Locations in China

Where are we?



Three Locations in Harbin



- A. Campus 1
- B. Campus 2
- C. Science Park

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1. Introduction

Geopolymer

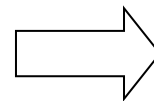
Characteristics

- Environmental friendly
 - No CO₂ emission
- Low temperature fabrication
 - RT~150 °C
- Ceramic formation ability upon heating
 - improved properties
- Adjustable thermal properties
 - Melting point、CTE

Applications and potential applications

- Geopolymer cement and concrete
- Refractory items
- Preparation of ceramics, i.e. Leucite
- Fireproof high-tech applications:
aircraft interior, thermal insulators for racing car engine, etc. as an alternative to organic polymers

Low strength
brittleness
low reliability



**Geopolymer matrix
composites**

{ particle
short fiber
continuous fiber



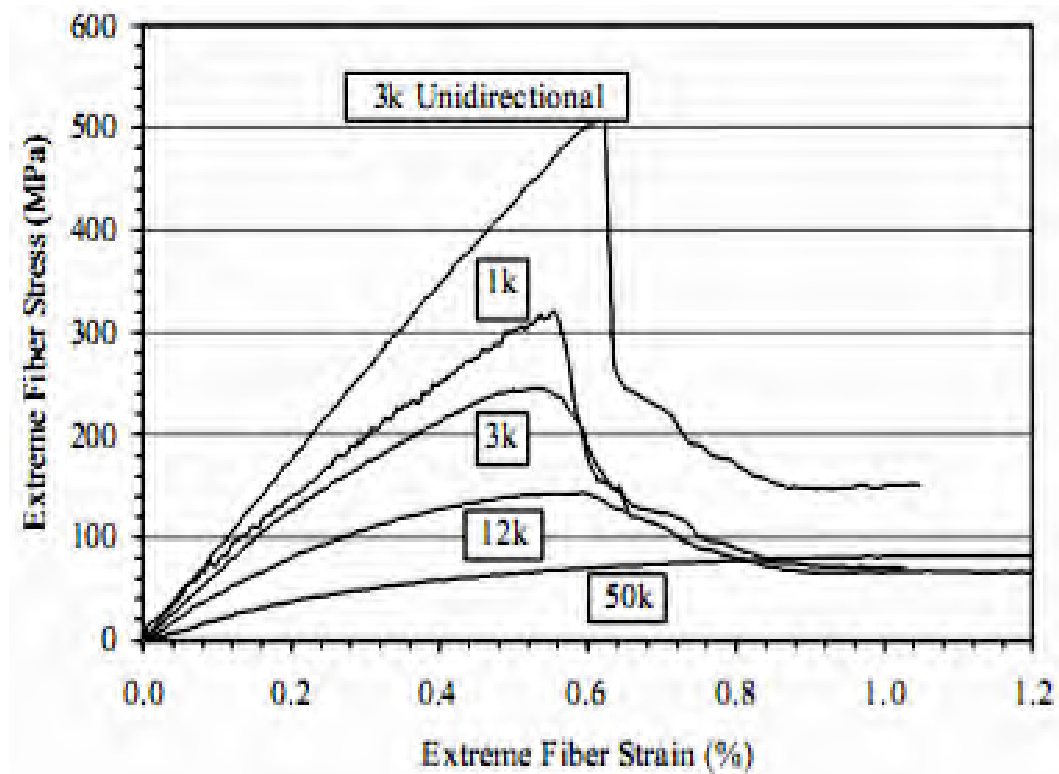


Fig.1 Stress-displacement curves of bidirectional carbon fiber reinforced geopolymer (Si/Al=27) composites. Fiber volume is 50%

---- J. Giancaspro, P. N. Balaguru, R. E. Lyon. *Proceedings of the Thirteenth International Offshore and Polar Engineering Conference*, Honolulu, Hawaoo, USA. 2003



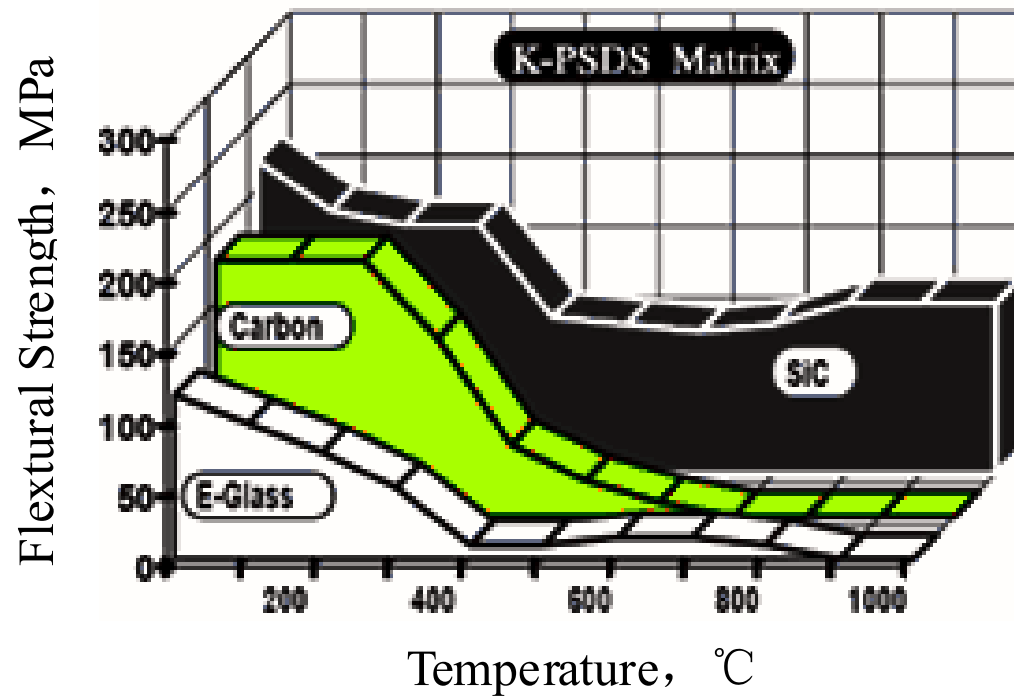


Fig.2 3-points flexural strength of geopolymer (Si/Al=3) composites reinforced by different kind of fibers (fiber contents is about 40vol%).

---- J. Davidovits, M. Davidovits. *36th Annual SAMPE Symposium*, 1991, 1939~1949

2. Thermal evolution and crystallization of geopolymer

2.1 Thermal evolution and crystallization mechanism of KGP

2.2 Effect of Cesium substitution on the crystallization of geopolymer

2.3 Effects of Cesium substitution on the phase transition of leucite



2.1 Thermal evolution and crystallization of KGP

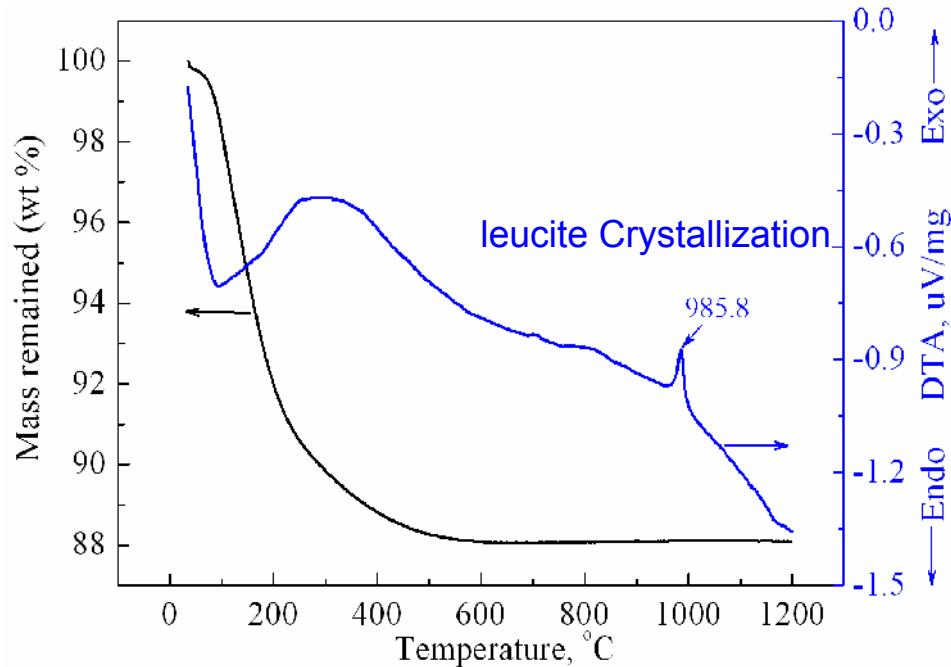


Fig.2-1 TG/DTA of KGP

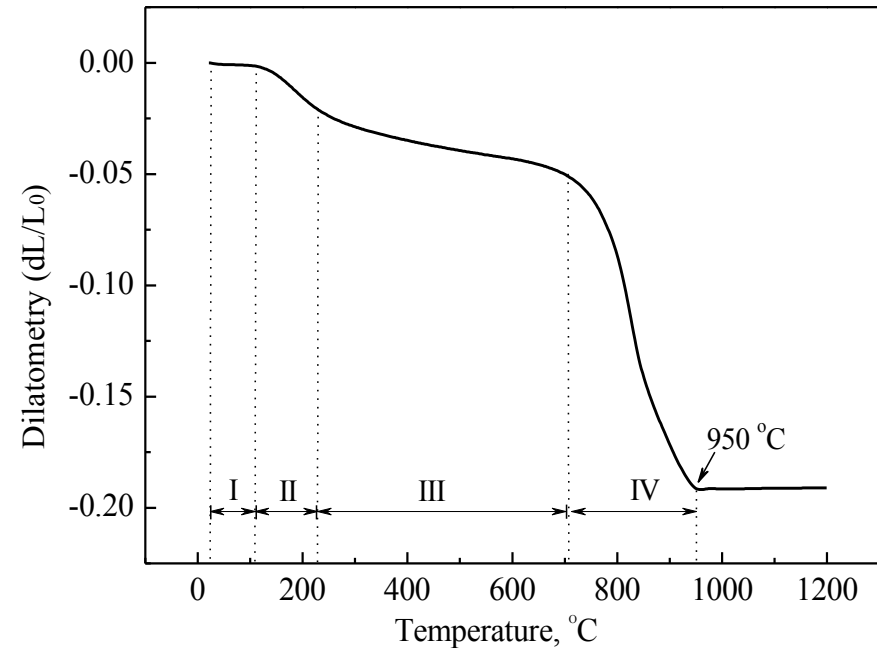


Fig.2-2 Thermal shrinkage of KGP

---- Peigang He, Dechang Jia. *Ceramics International*. 2011, 37 (1), 59-63.



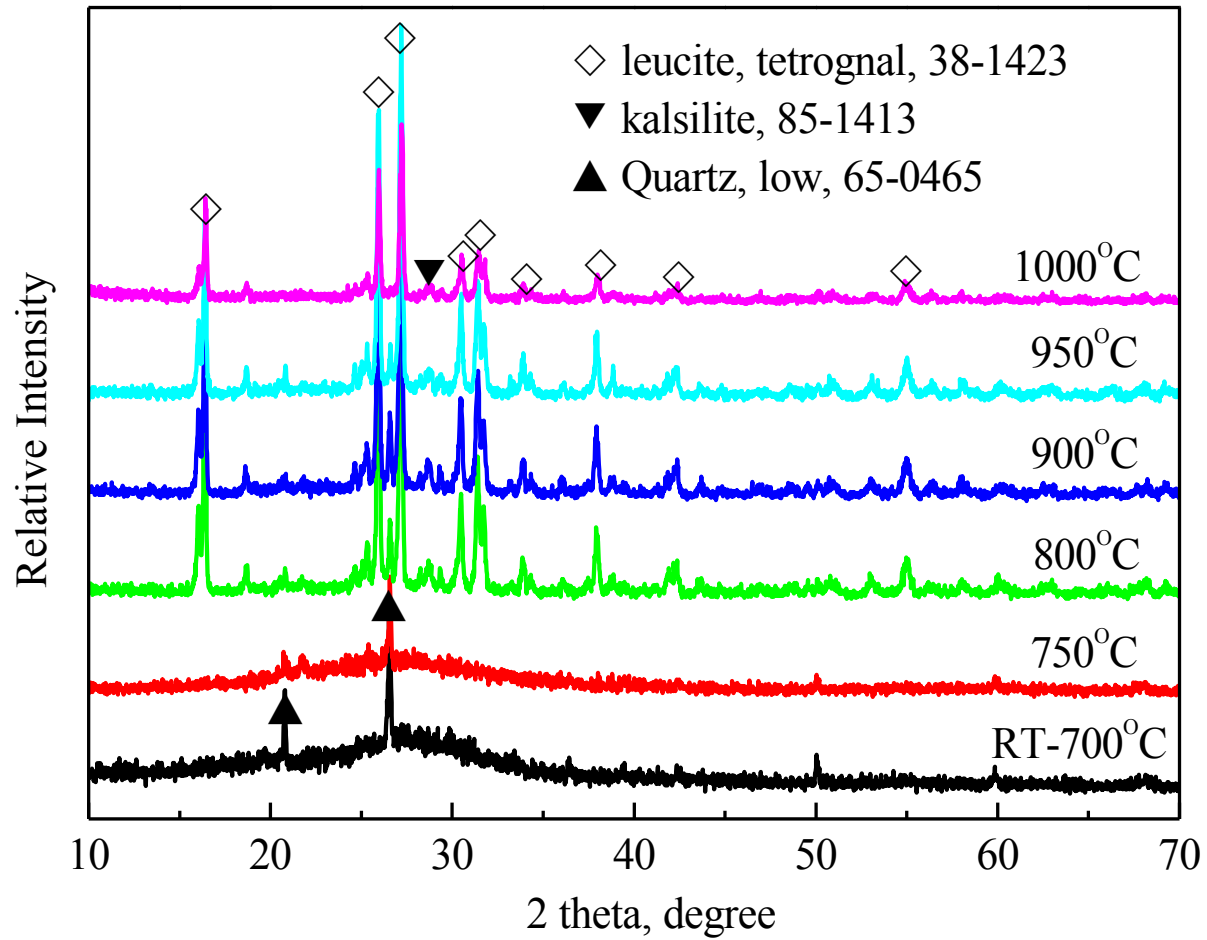
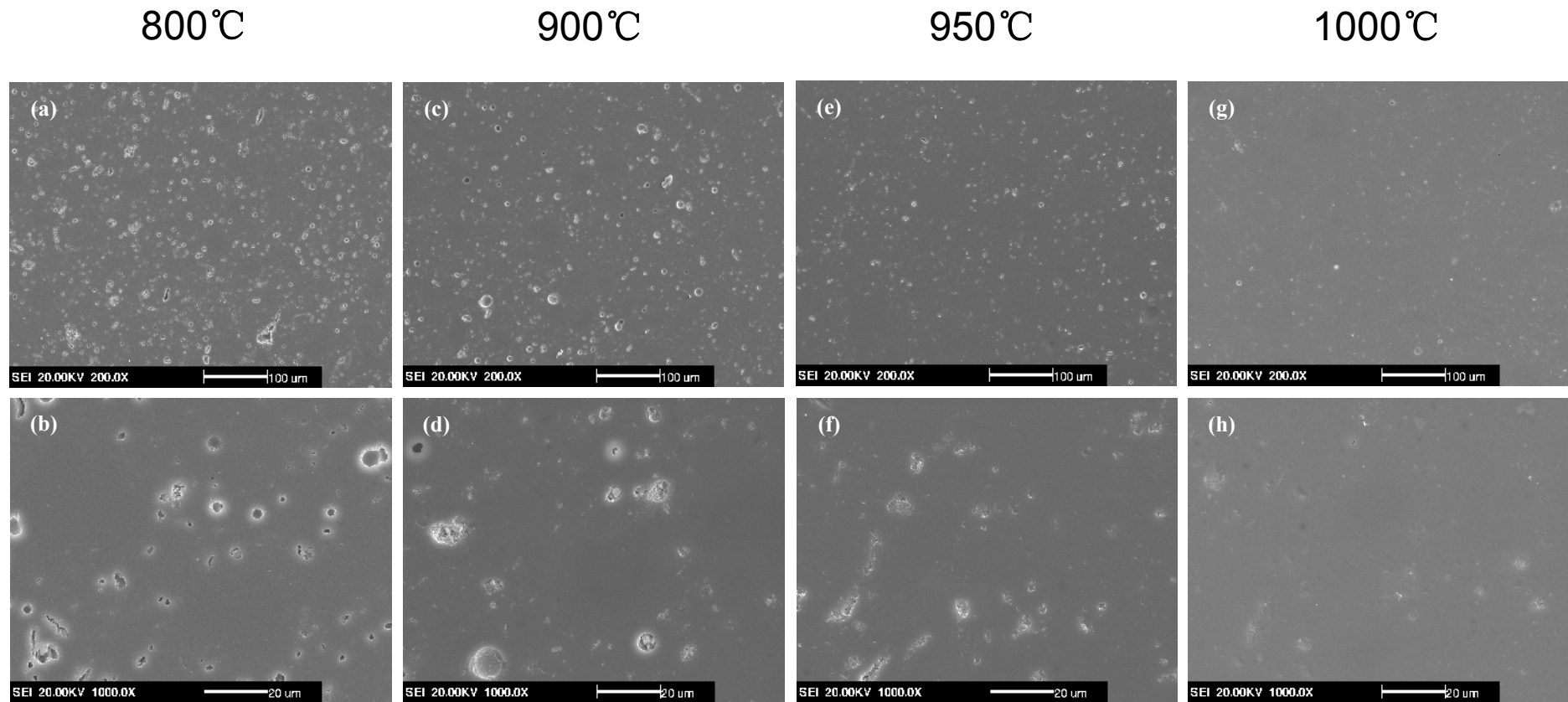
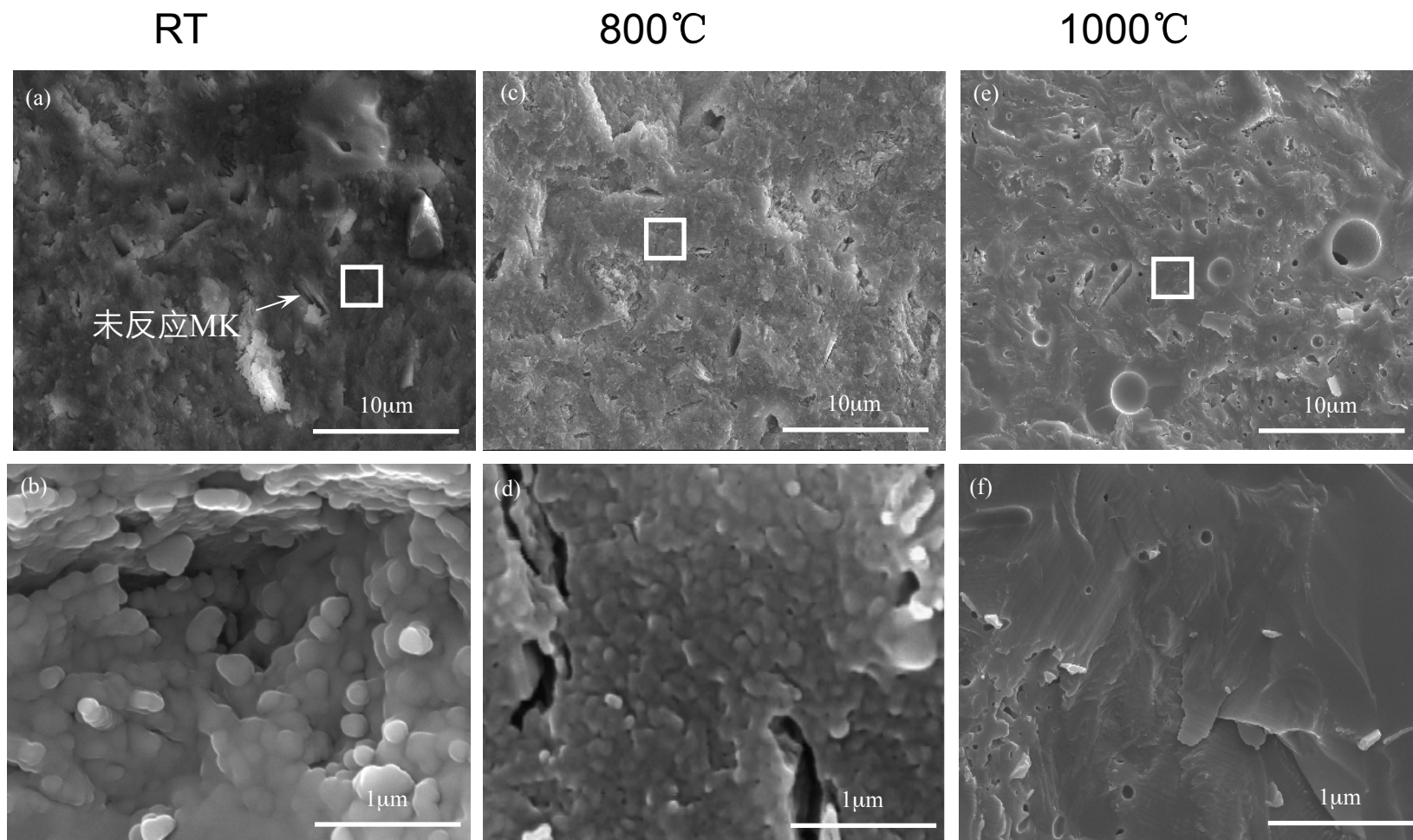


Fig.2-3 XRD pattern of KGP soaked at different temperature for 2h



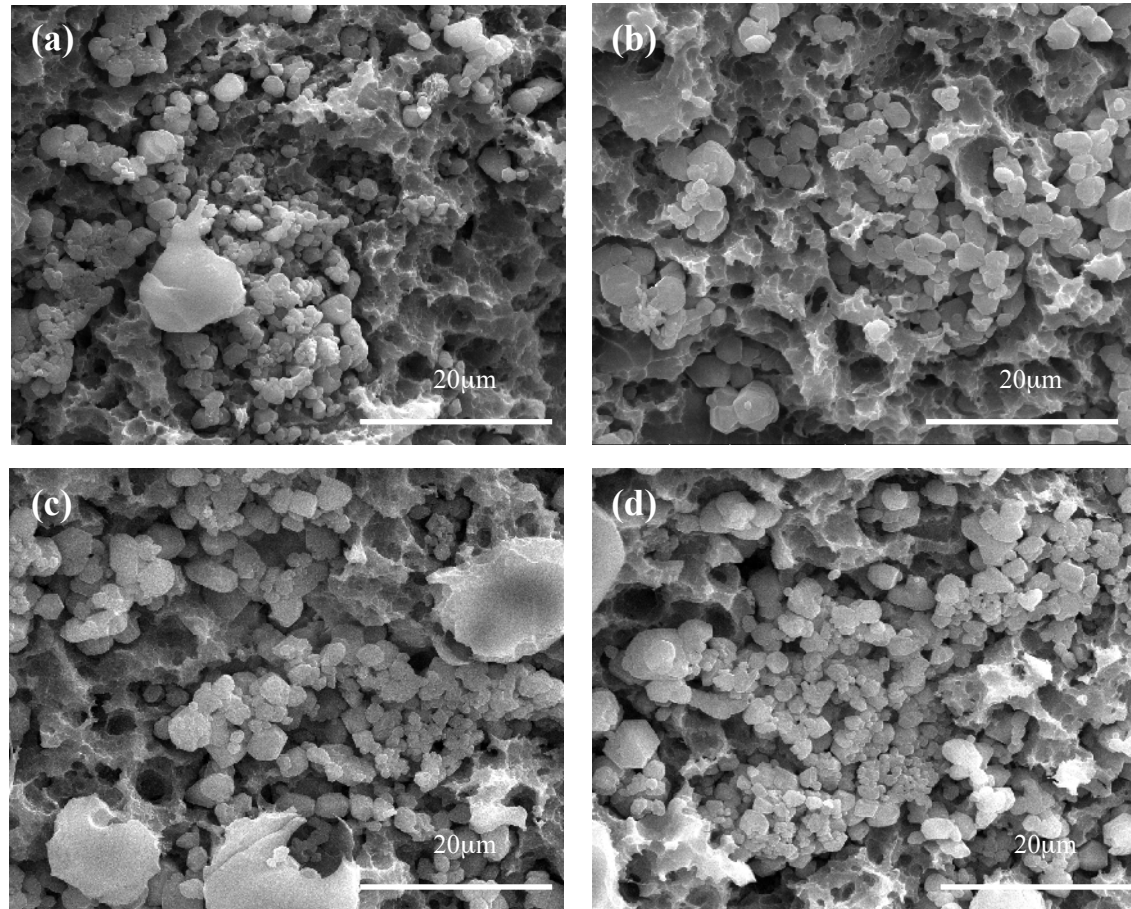
**Fig.2-4 Microstructure of KGP soaked at different temperature for 2h.
(a), (b) 800 °C; (c), (d) 900 °C; (e), (f) 950 °C; (g), (h) 1000 °C.**



**Fig.2-5 Fractographs of KGP soaked at different temperature for 2h
(a), (b) RT; (c), (d) 800°C; (e), (f) 1000°C**

---- Peigang He, Dechang Jia. *Ceramics International*. 2011, 37 (1), 59-63.





**Fig.2-6 Fracture surface morphology of the KGP treated at different temperature and etched in 3wt.% HF at room temperature for 30s:
(a) 800 °C, (b) 900 °C, (c) 950 °C, (d) 1000 °C**

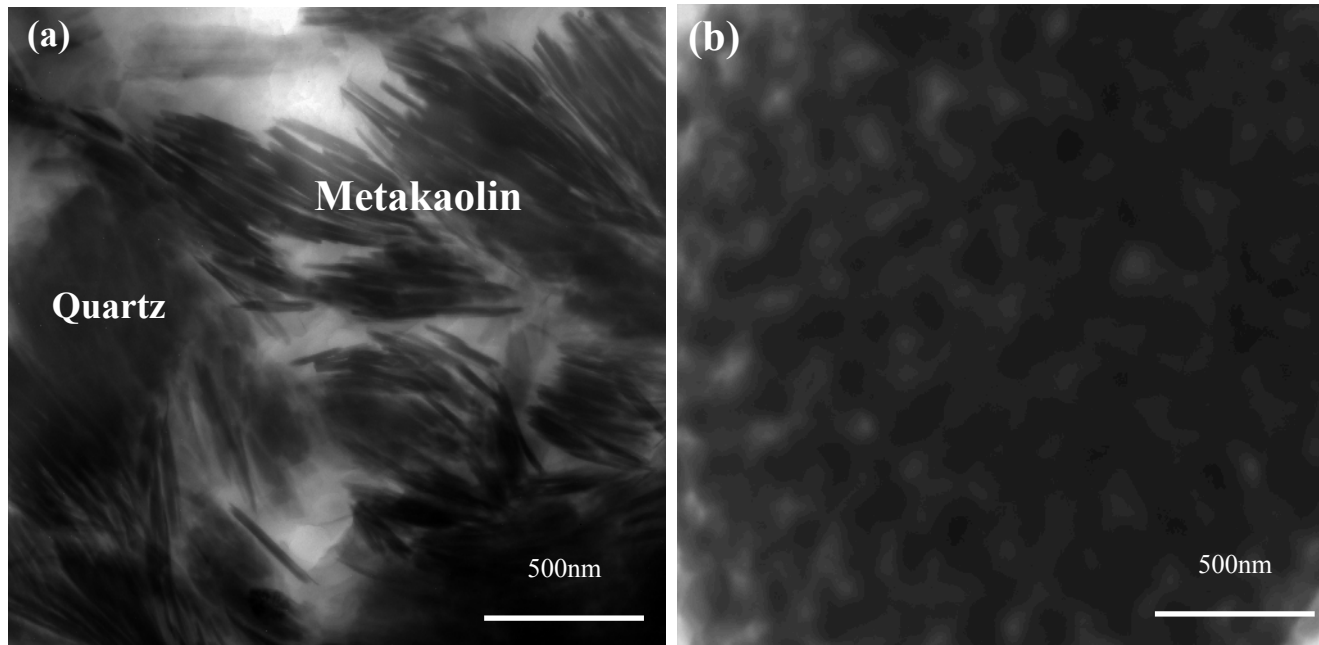


Fig.2-7 TEM images of: metakaolin (a) and the resulted geopolymer (b)

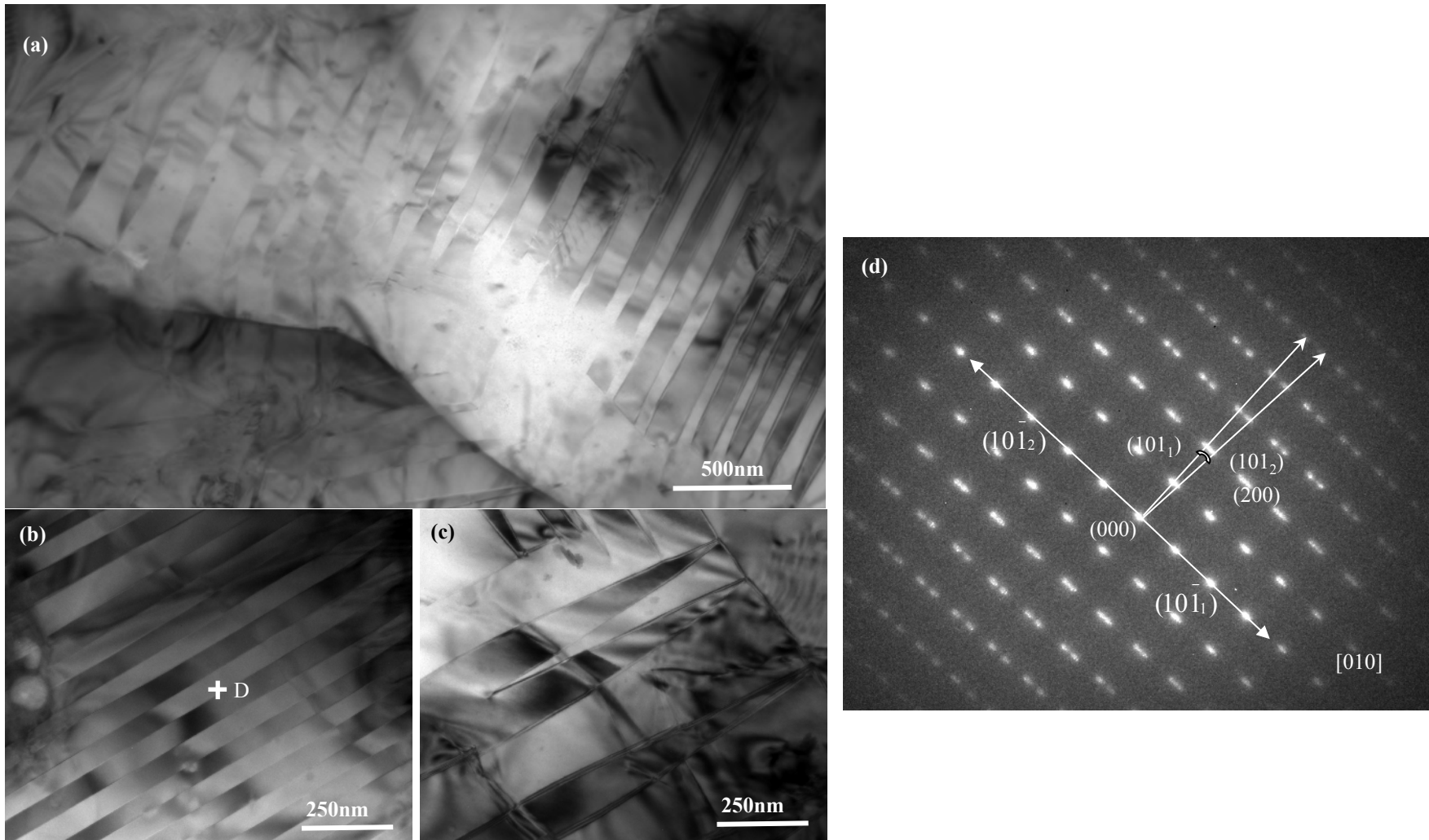


Fig. 2-8 TEM observation of geopolymer treated at 1000°C for 2h:
 (a) low magnification, (b) and (c) high magnification, (d) SAD pattern of area D

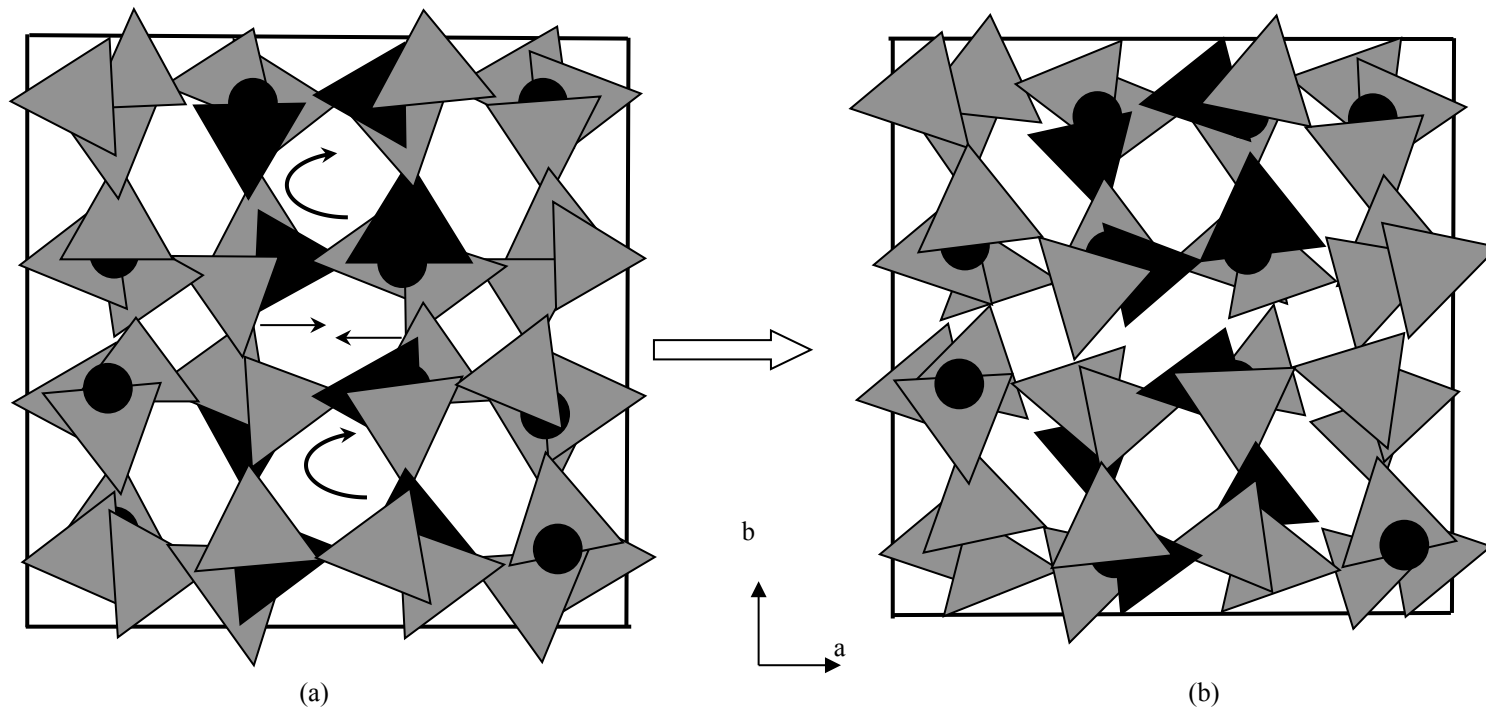
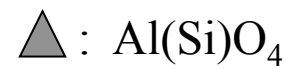


Fig. 2-9 Crystal structures of leucite: (a) cubic phase, (b) tetrahedral phase



Black tetrahedron : The fourfold rings

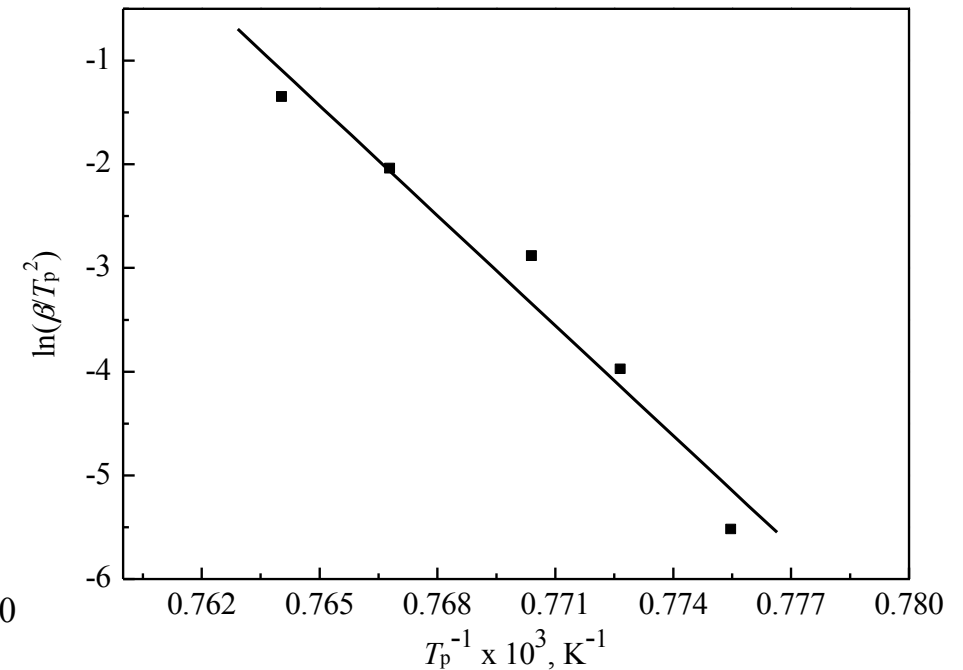
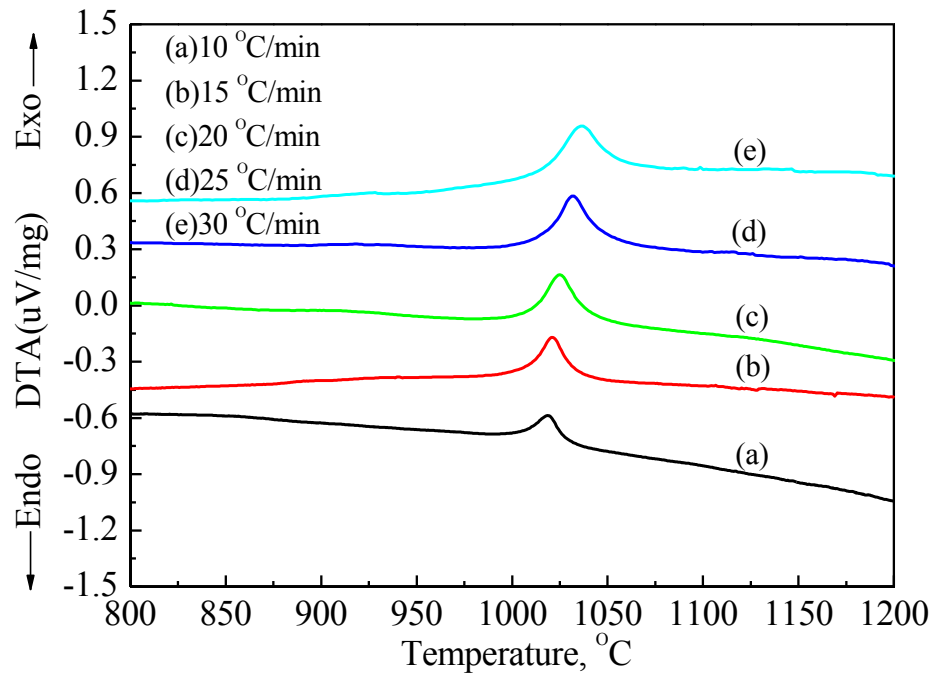


Fig. 2-10 DTA of KGP using different heating rates Fig. 2-11 Kissinger plot of $\ln(\beta/T_p^2)$ vs. T_p^{-1}

Kissinger method:
$$\ln\left(\frac{T_p^2}{\beta}\right) = -\frac{E_c}{RT_p} + \ln\frac{E}{Rv}$$

Augis-Bennett:
$$n = \frac{2.5RT_p^2}{\Delta TE}$$

$n=3.89, E_a=455.9 \text{ kJ/mol}$

Three dimensional crystal growth mechanism

---- Peigang He, Dechang Jia. *Ceramics International*. 2011, 37 (1), 59-63.



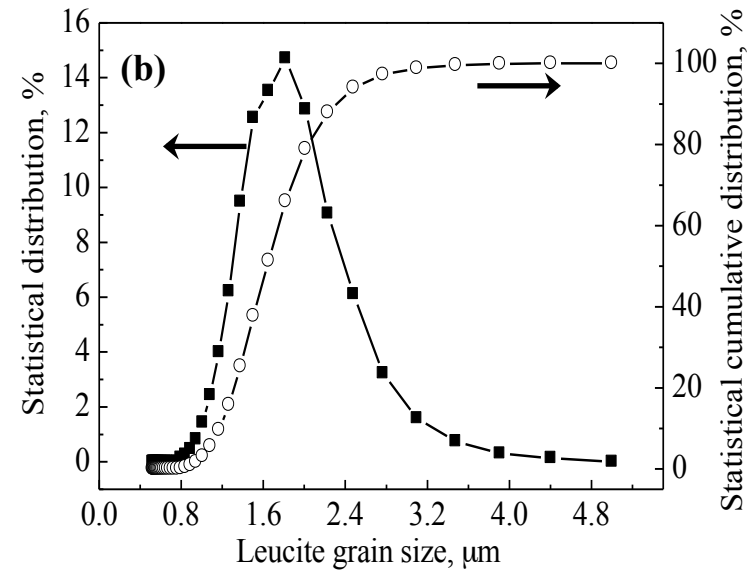
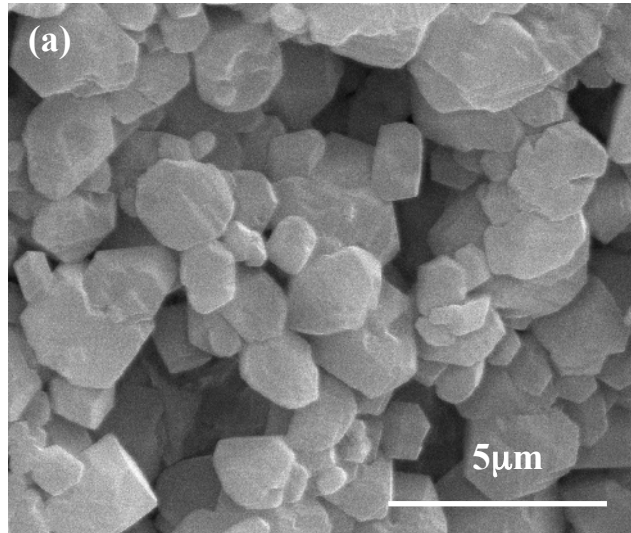


Fig. 2-12 SEM micrographs of leucite grains (a) and size distribution (b)

Table 2-1 Mechanical properties of the geopolymer and the resulted leucite ceramic

| Specimen | Flexural strength (MPa) | Young's modulus (GPa) | Fracture toughness (MPa · m ^{1/2}) | Vickers hardness (GPa) |
|-----------------|-------------------------|-----------------------|--|------------------------|
| KGP | 12.3±1.2 | 10.3±1.2 | 0.2±0.04 | 0.68±0.04 |
| Leucite ceramic | 70.0±6.8 | 65.0±6.3 | 1.3±0.16 | 7.39±0.24 |

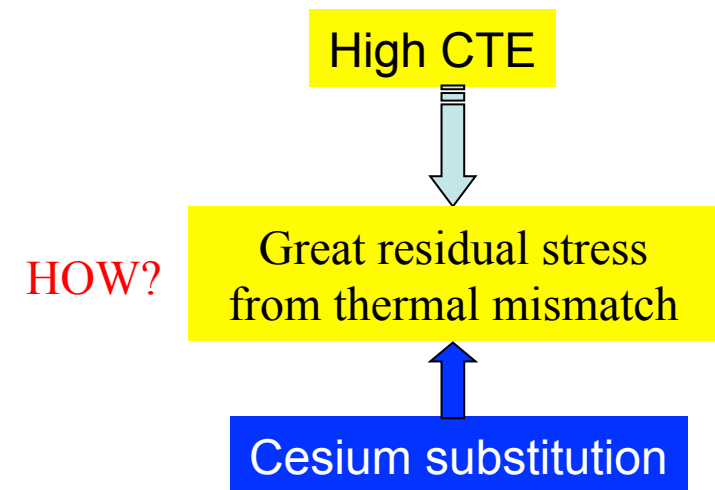
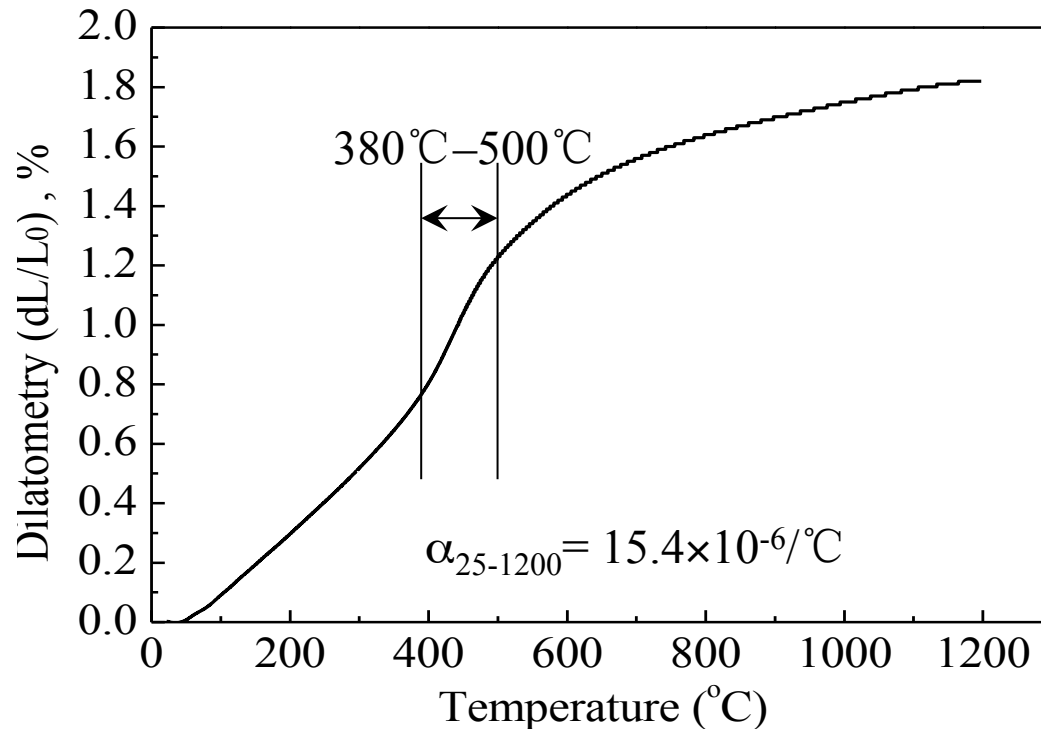


Fig.2-13 Thermal expansion of the ceramics derived from KGP

2.2 Effect of Cesium substitution on the crystallization of geopolymer

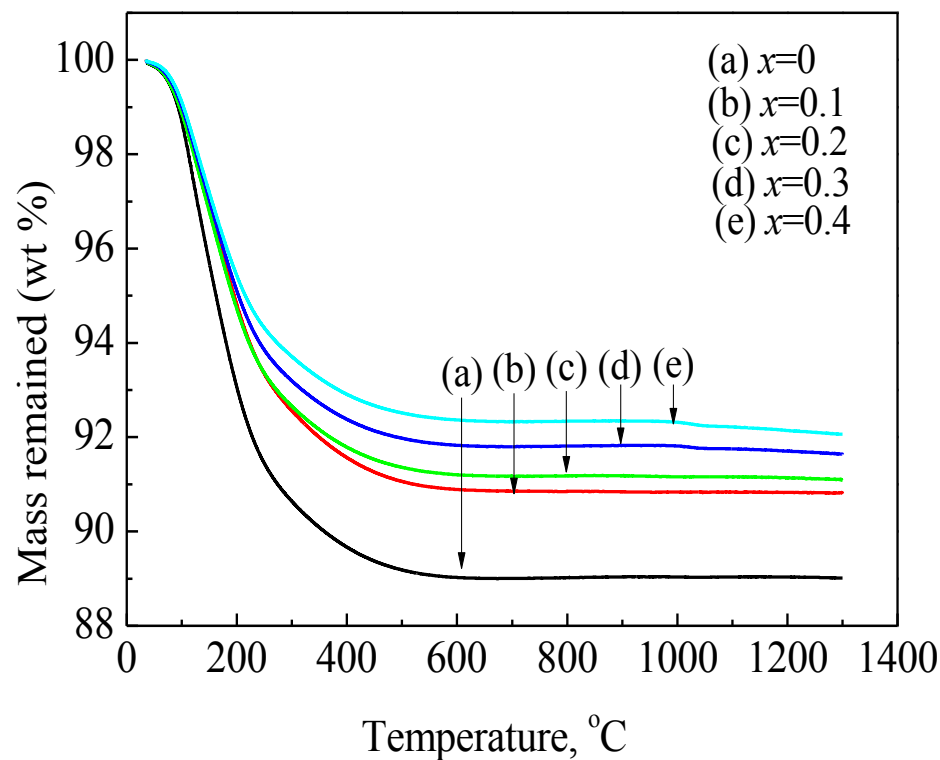


Fig. 2-14 Weight loss of $\text{Cs}_x\text{K}_{(1-x)}\text{GP}$

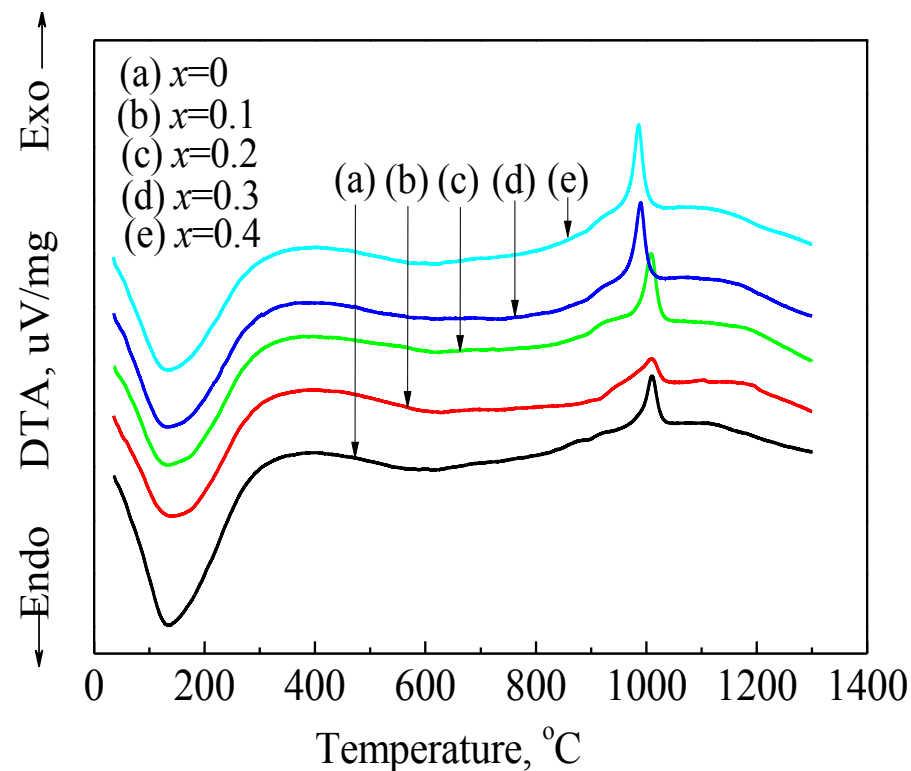


Fig. 2-15 DTA of $\text{Cs}_x\text{K}_{(1-x)}\text{GP}$

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (8), 2395-2400.



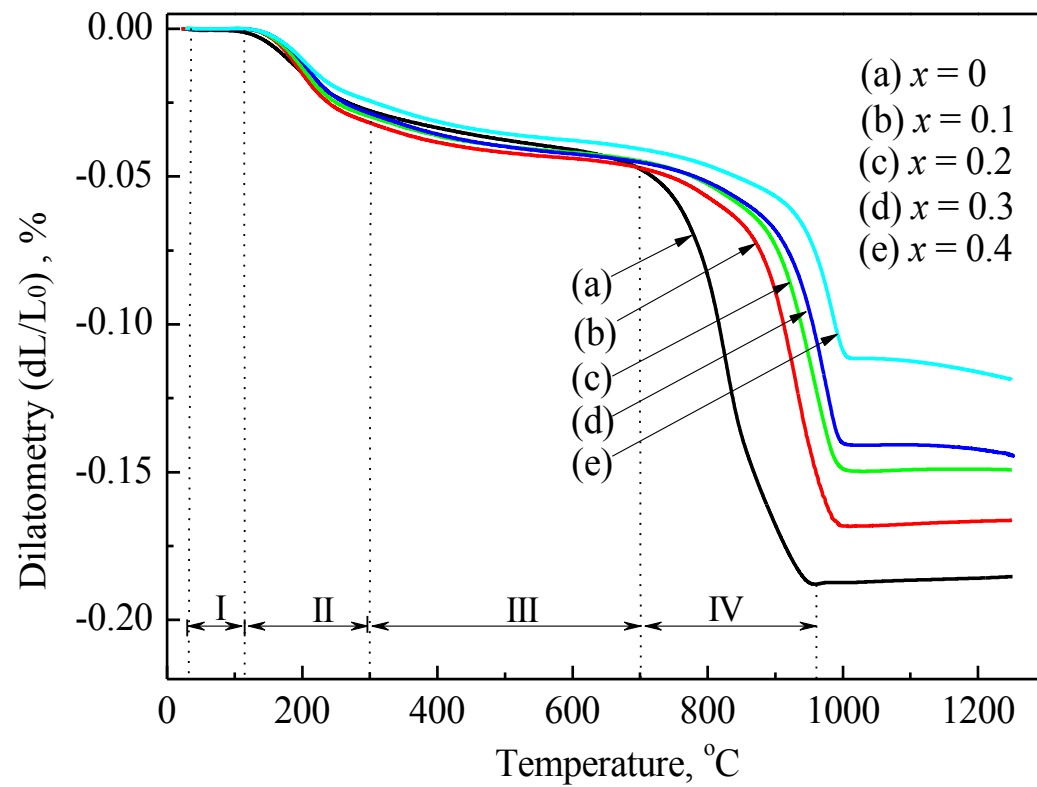


Fig.2-16 Thermal shrinkage of $\text{Cs}_x\text{K}_{(1-x)}\text{GP}$

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (8), 2395-2400.



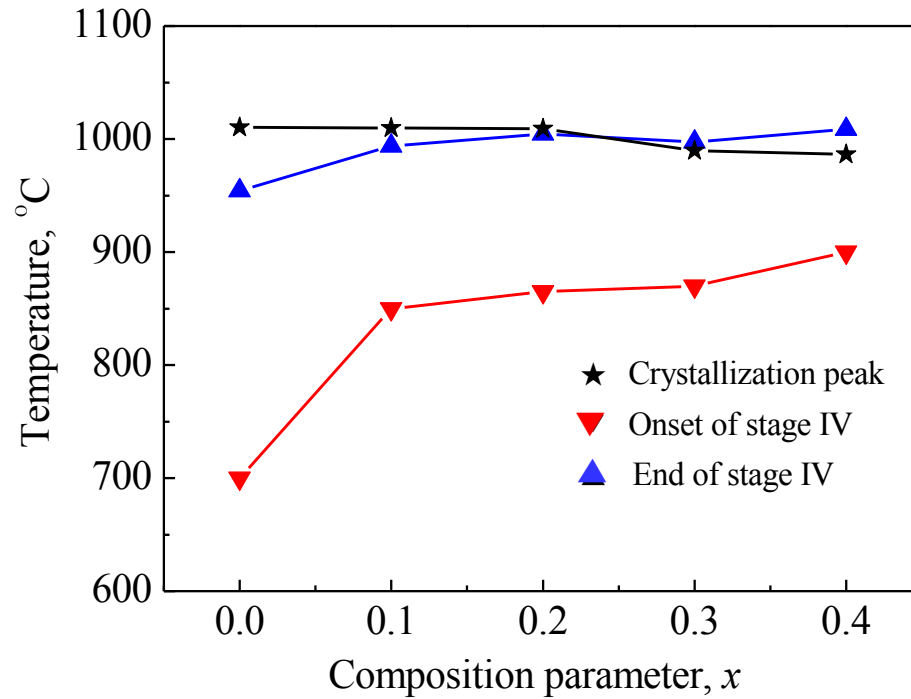


Fig. 2-17 Onset and end temperature of stage IV and crystallization temperature of $\text{Cs}_x\text{K}_{(1-x)}\text{GP}$

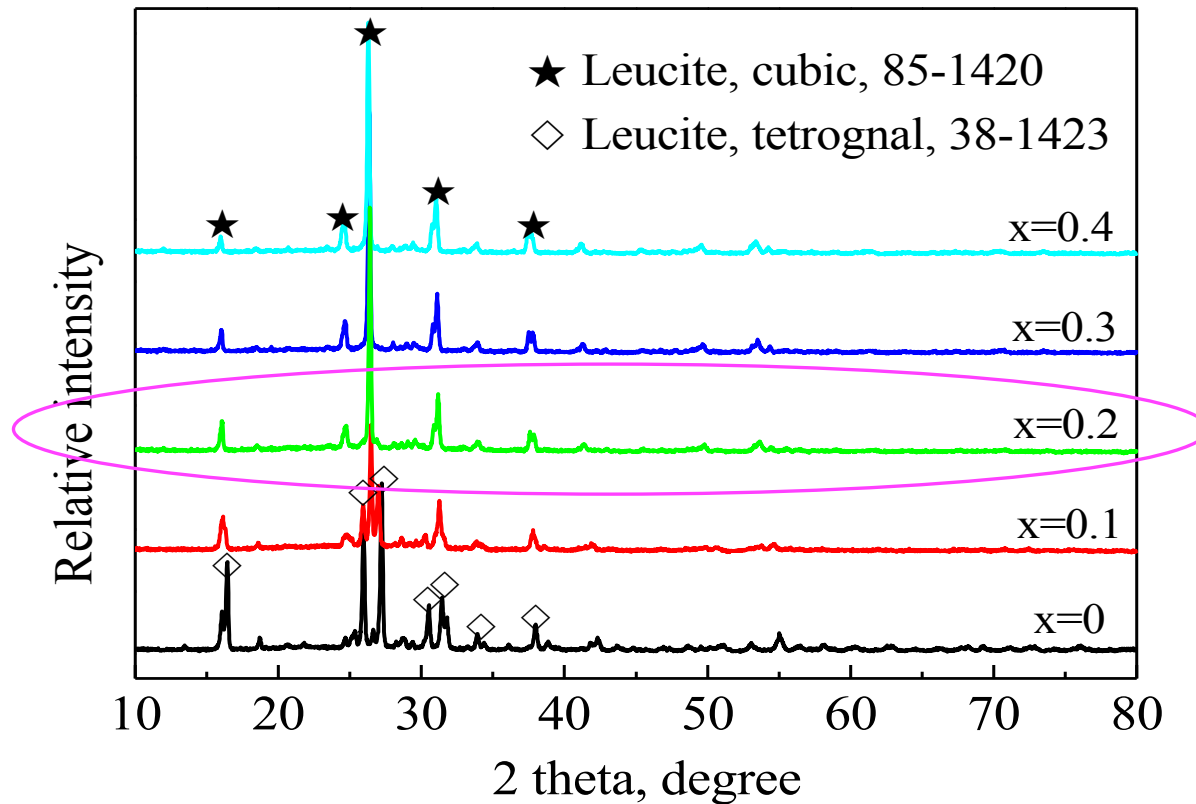


Fig. 2-18 XRD patterns of $\text{Cs}_x\text{K}_{(1-x)}\text{GP}$ ceramics heat treated at 1200°C for 2h

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (8), 2395-2400.



2.3 Effects of Cesium substitution on the phase transition of leucite

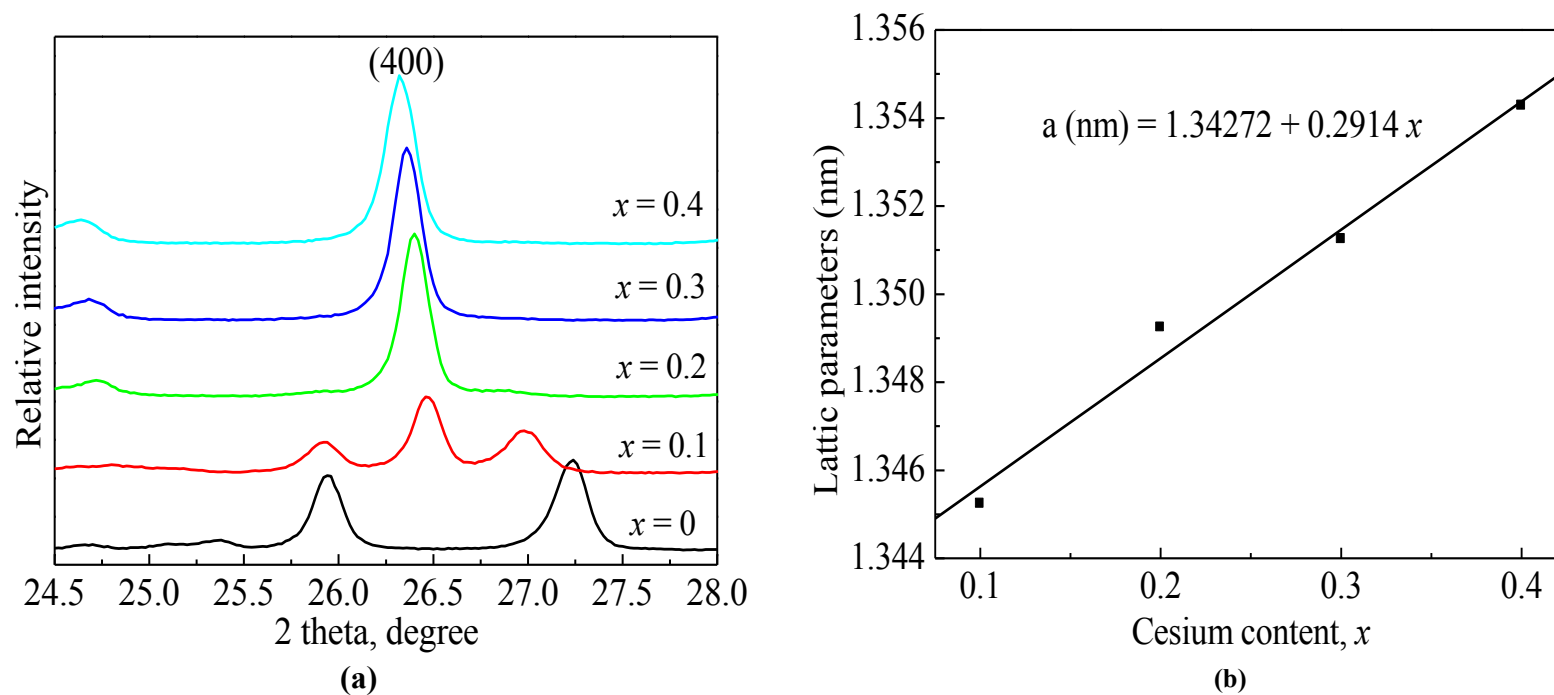


Fig. 2-19 Slow step-scan XRD patterns in a 2θ range of $24.5\sim 28^\circ$ (a) and the calculated lattice parameters (b) of leucite

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (8), 2395-2400.



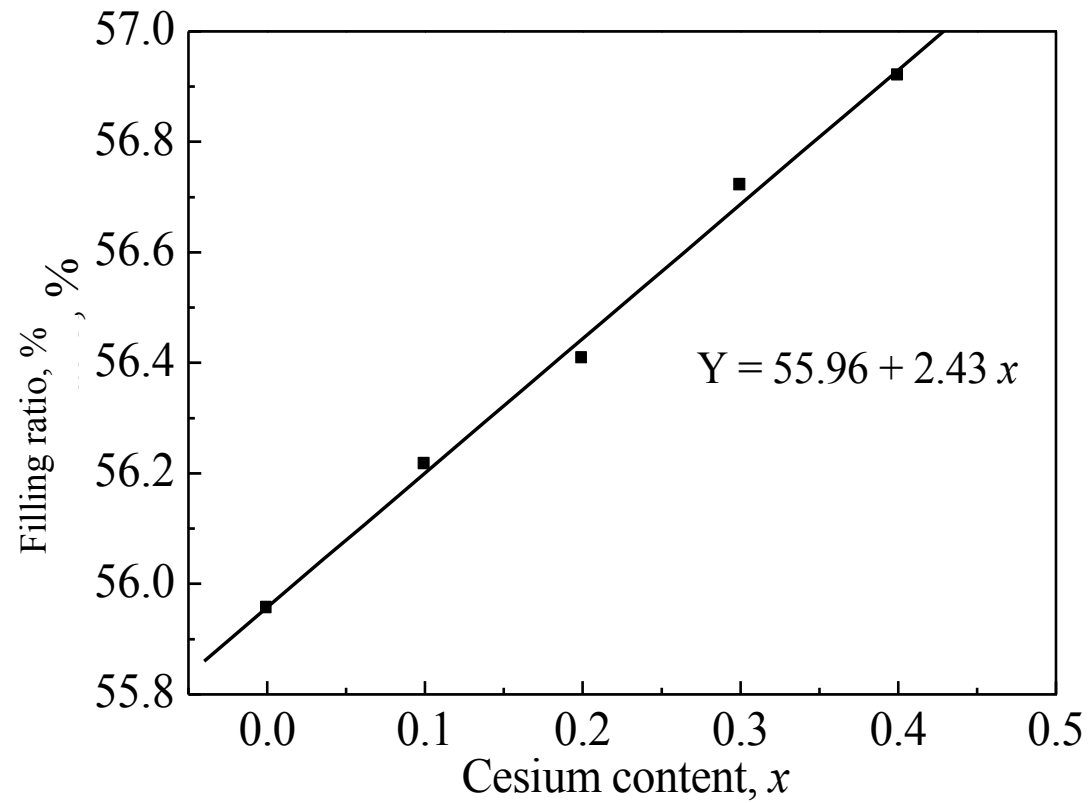


Fig. 2-20 The filling ratio of leucite crystal cell derived from $\text{Cs}_x\text{K}_{(1-x)}\text{GP}$

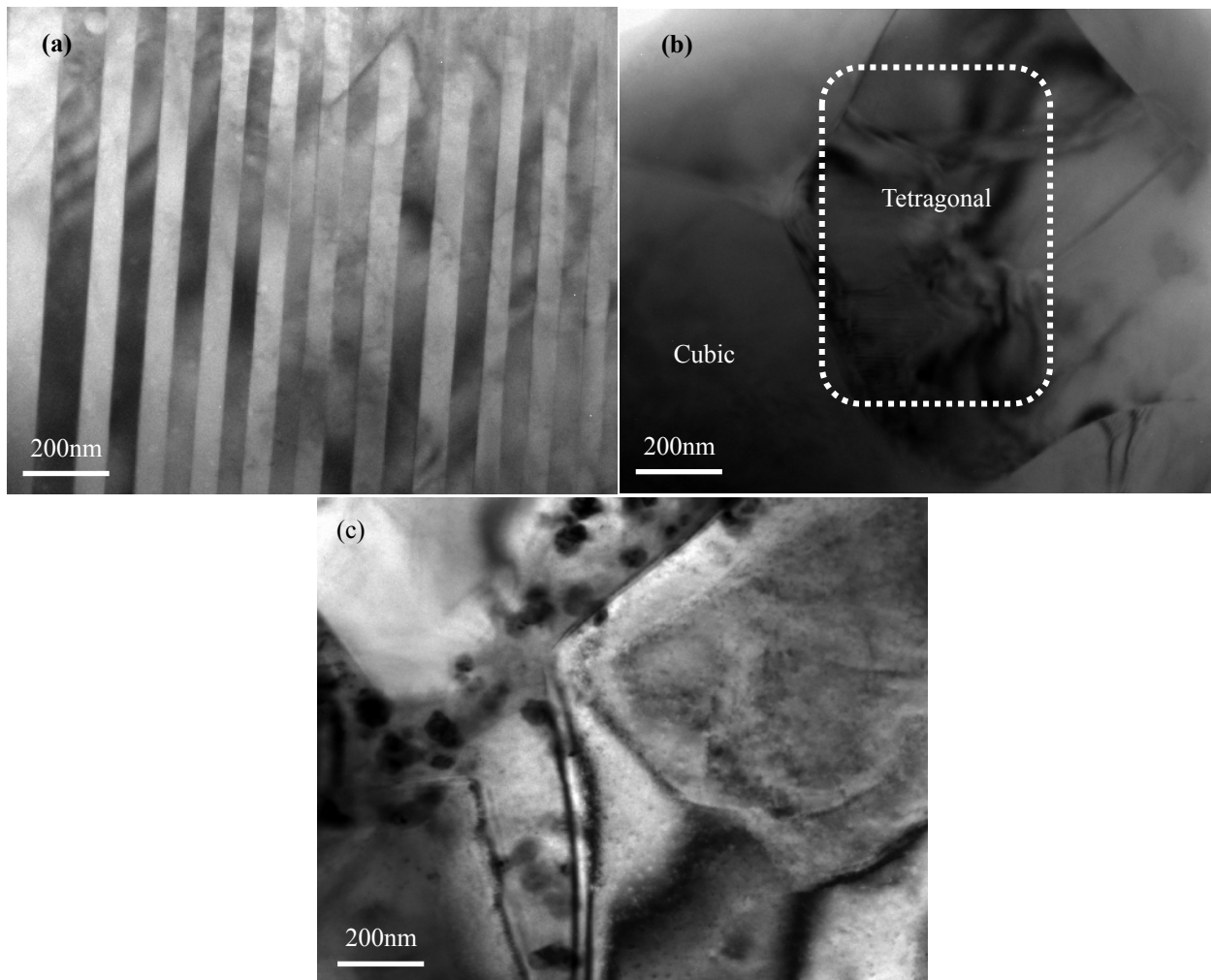


Fig. 2-21 TEM micrographs of the ceramics derived from $\text{Cs}_x\text{K}_{(1-x)}\text{GP}$:
(a) $x=0$, (b) $x=0.1$, (c) $x=0.2$.

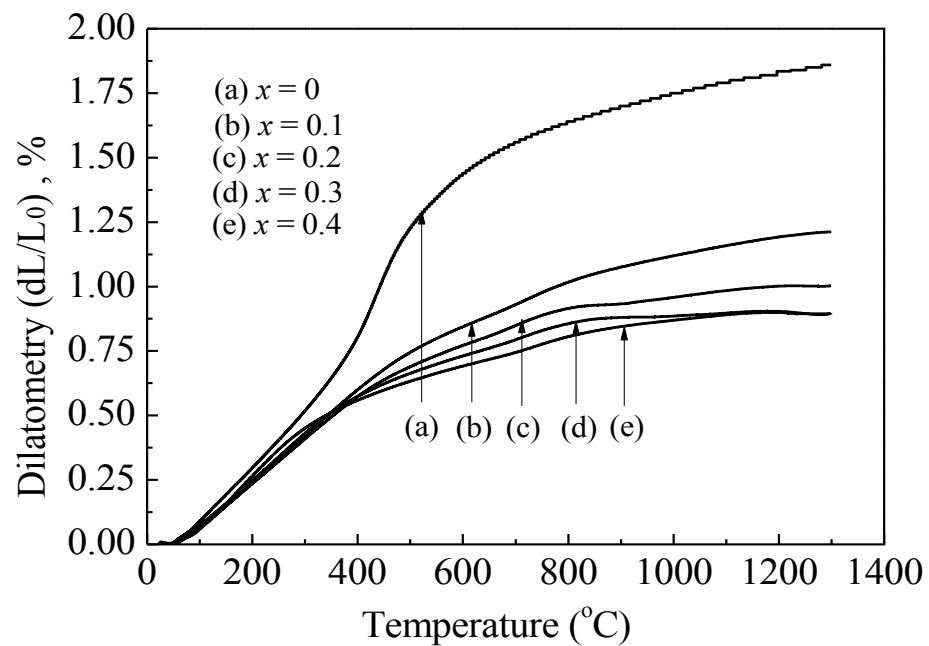


Fig. 2-22 Thermal expansion curves of $\text{Cs}_x\text{K}_{(1-x)}\text{GP}$ ceramic

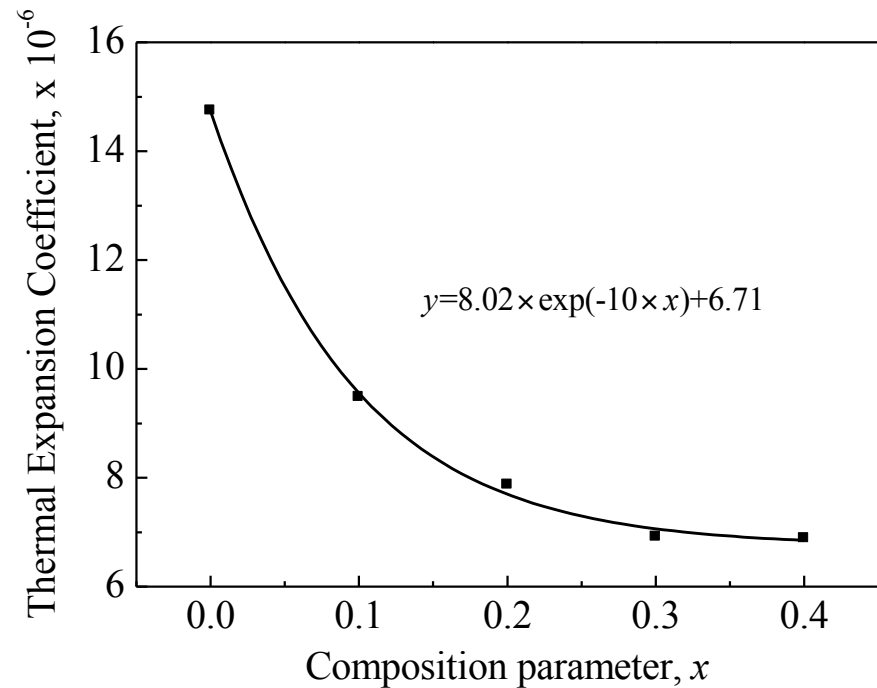


Fig. 2-23 Linear coefficient of thermal expansion of $\text{Cs}_x\text{K}_{(1-x)}\text{GP}$ ceramic

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (8), 2395-2400.



2.4 Summary

- ◆ Geopolymer technology provides a novel method to fabricate leucite ceramic with greatly improved mechanical properties.
- ◆ Thermal shrinkage of geopolymer during heat treatment can be divided into structural resilience, dehydration, dehydroxylation and sintering.
- ◆ For the K-based geopolymer, leucite crystallization appeared after the sintering stage and the Avrami parameter indicates the three-dimensional crystal growth mechanism.
- ◆ Leucite ceramic possesses an adjustable thermal expansion coefficient by doping cesium ions.



3. Effects of high-temperature heat treatment on the mechanical properties of C_f/geopolymer composites

3.1 Effects of heat treatment on the properties of C_f/KGP

3.2 Effects of heat treatment on the mechanical properties of C_f/CsKGP

3.3 Summary



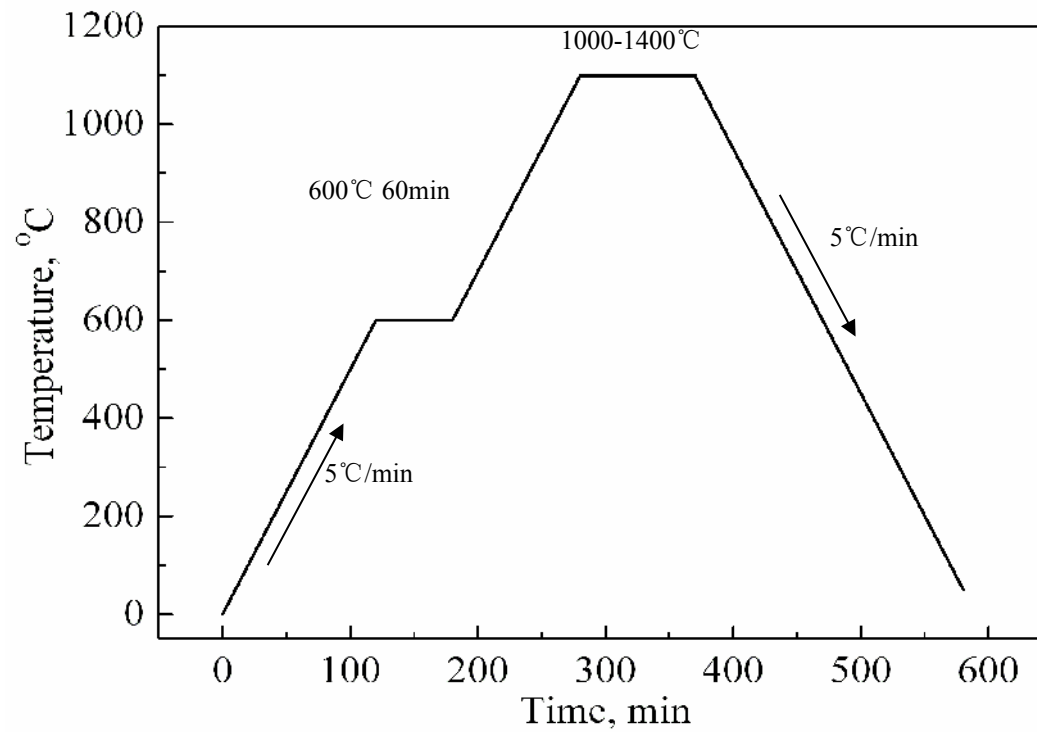


Fig. 3-1 High-temperature heat treatment procedure of the composites

3.1 Effects of heat treatment on the properties of C_f/KGP

- ◆ **Carbon fiber:** Unidirectional
- ◆ **Matrix:** $K_2O \cdot Al_2O_3 \cdot (3 \sim 5)SiO_2 \cdot 11H_2O$
- ◆ **Composites without and after heat treatment were denoted as:**
KC-W, KC-1000, KC-1100, KC-1200, KC-1300 and KC-1400, respectively
- ◆ **Fiber contents:** 20vol% for KC-W and 25vol% for others



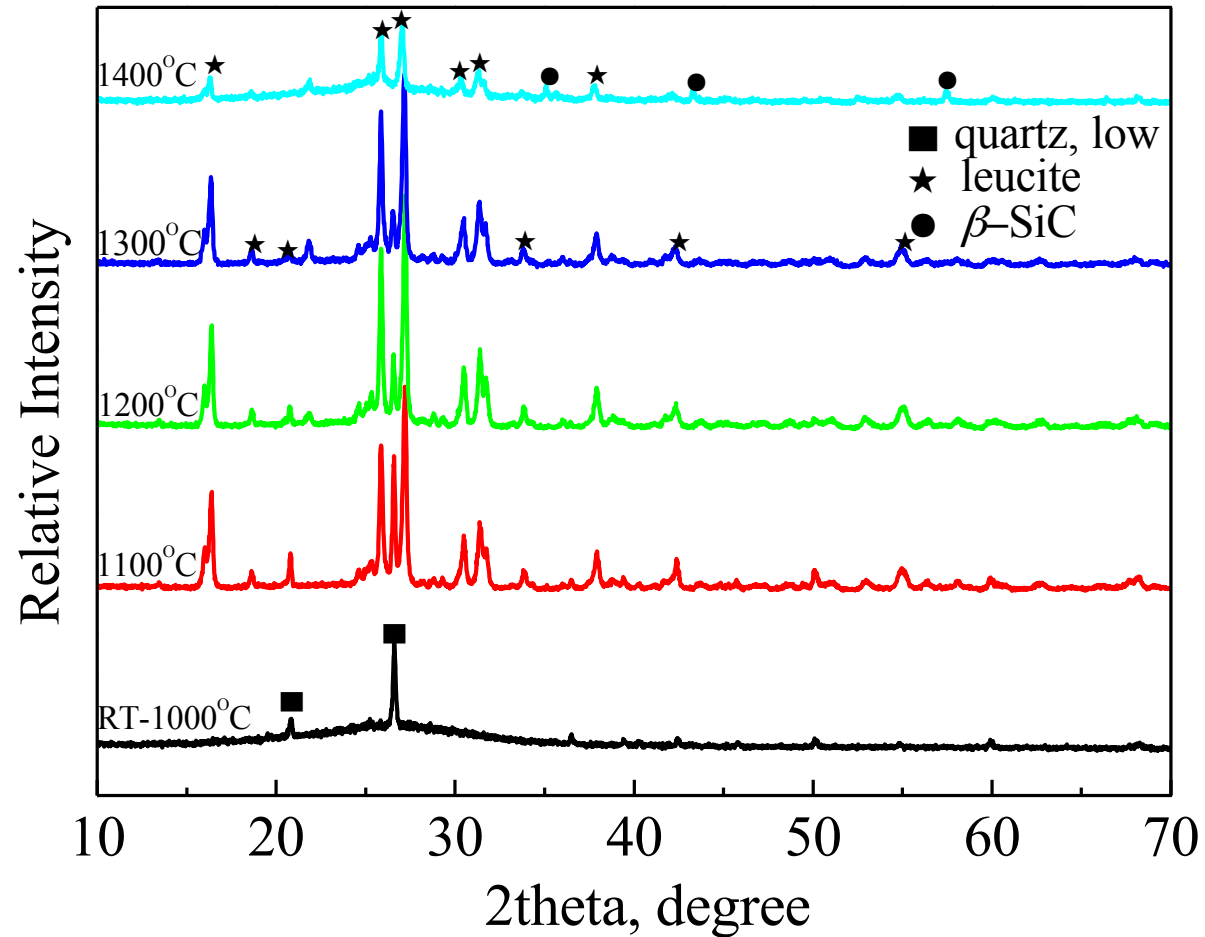


Fig. 3-2 XRD patterns of C_f/KGP heat treated at different temperature

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (4), 1447-1453.



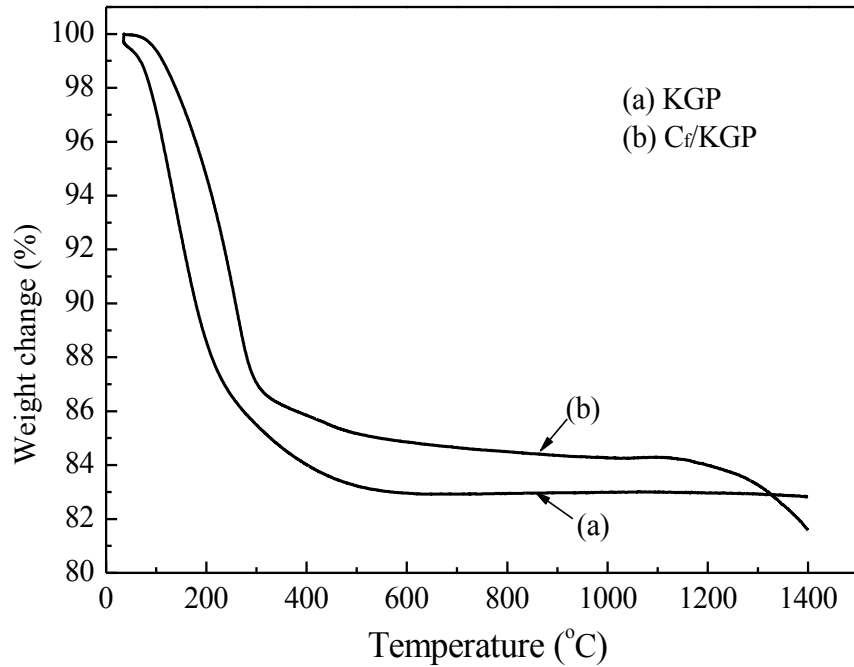


Fig. 3-3 Weight loss of KGP and C_f /KGP

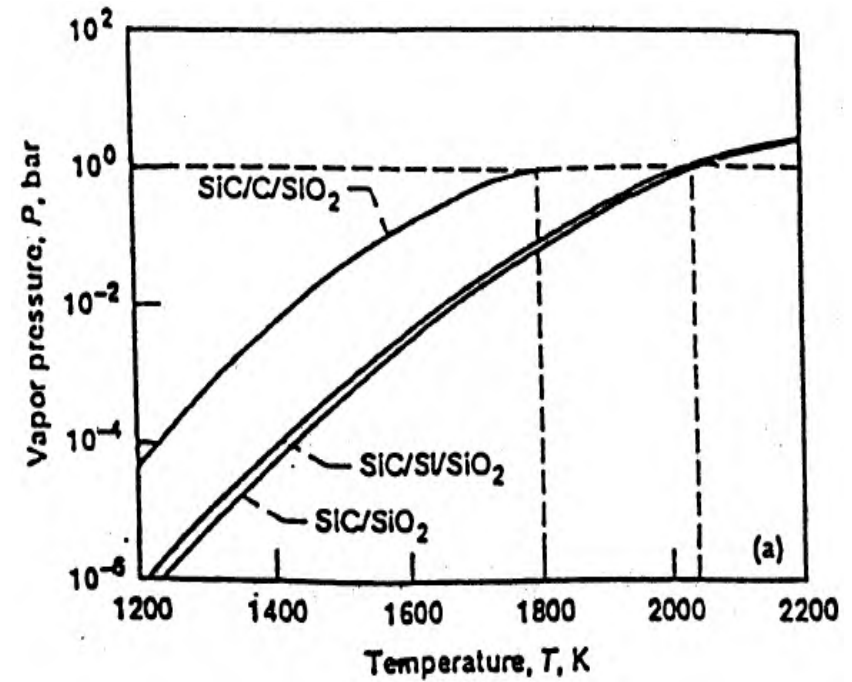
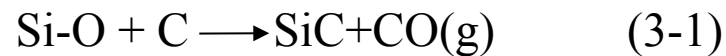


Fig. 3-4 Total vapor pressure of interfacial reaction of Si-C-O system as a function of temperature [32]



---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (4), 1447-1453.



Table 3-1 The reactions and standard Gibbs free energy

| Reaction | Standard Gibbs free energy (J·mol ⁻¹) |
|--|---|
| (1) $C_{(s)} + 0.5O_{2(g)} = CO_{(g)}$ | $-114400 - 85.77T$ |
| (2) $Si_{(s)} + C_{(s)} = SiC_{(s)}$ | $-73050 + 7.66T$ |
| (3) $Si_{(s)} + O_{2(g)} = SiO_{2(s)}$ | $-907100 + 175.73T$ |

Gibbs free energy for (3-1): $\Delta G^\circ = 605250 - 339.61T$

The theoretical reaction temperature is 1782°C



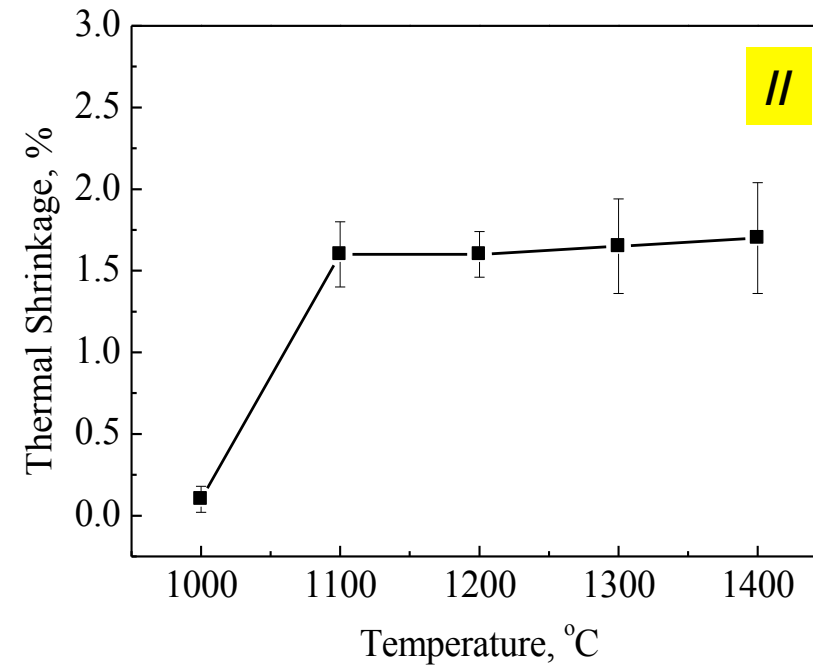
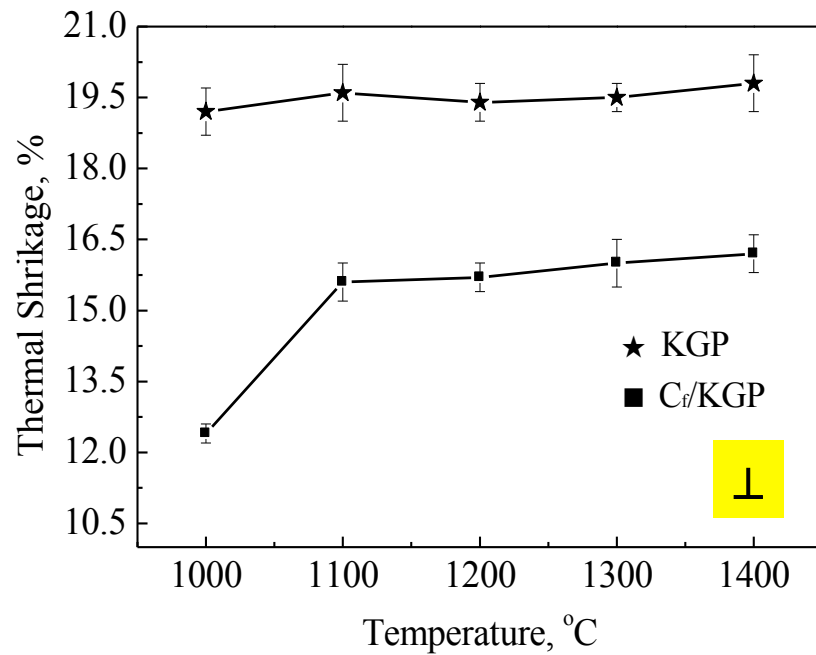


Fig. 3-5 Thermal shrinkages of the composite heat treated in different directions: (a) perpendicular and (b) parallel to the fiber direction

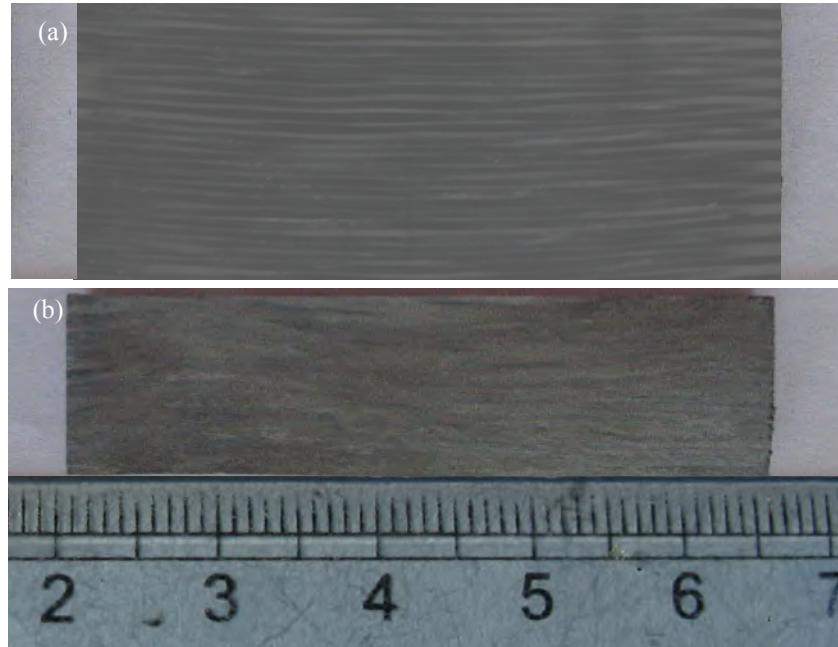


Fig. 3-6 Morphologies of C_f /KGP composite
(a) before and (b) after heat treatment at
1100°C

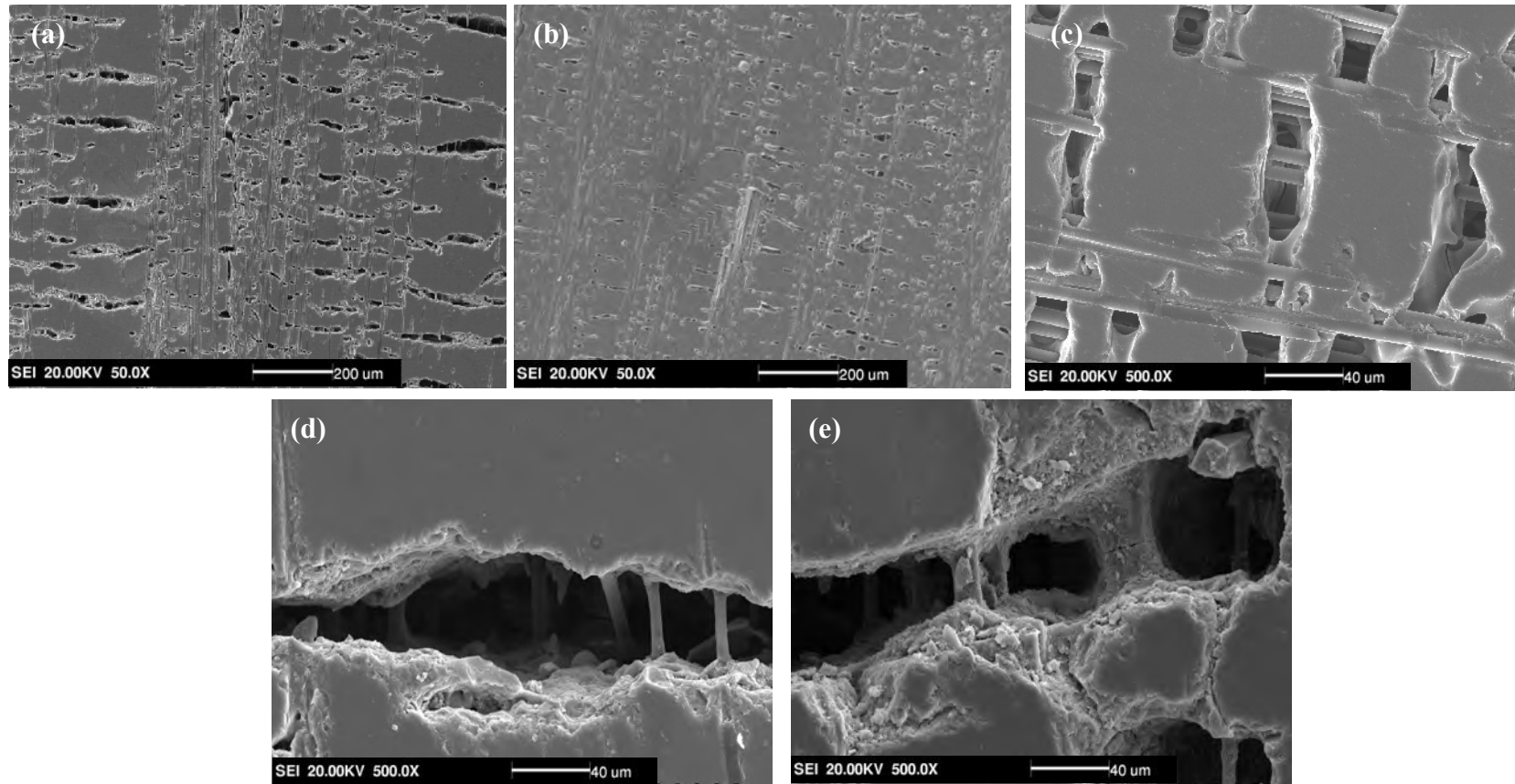


Fig. 3-7 Microstructure of polished cross section of:
(a) KC-1000, (b) KC-1100 (low magnification), (c) KC-1100 (high magnification), (d) KC-1300, (e) KC-1400

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (4), 1447-1453.



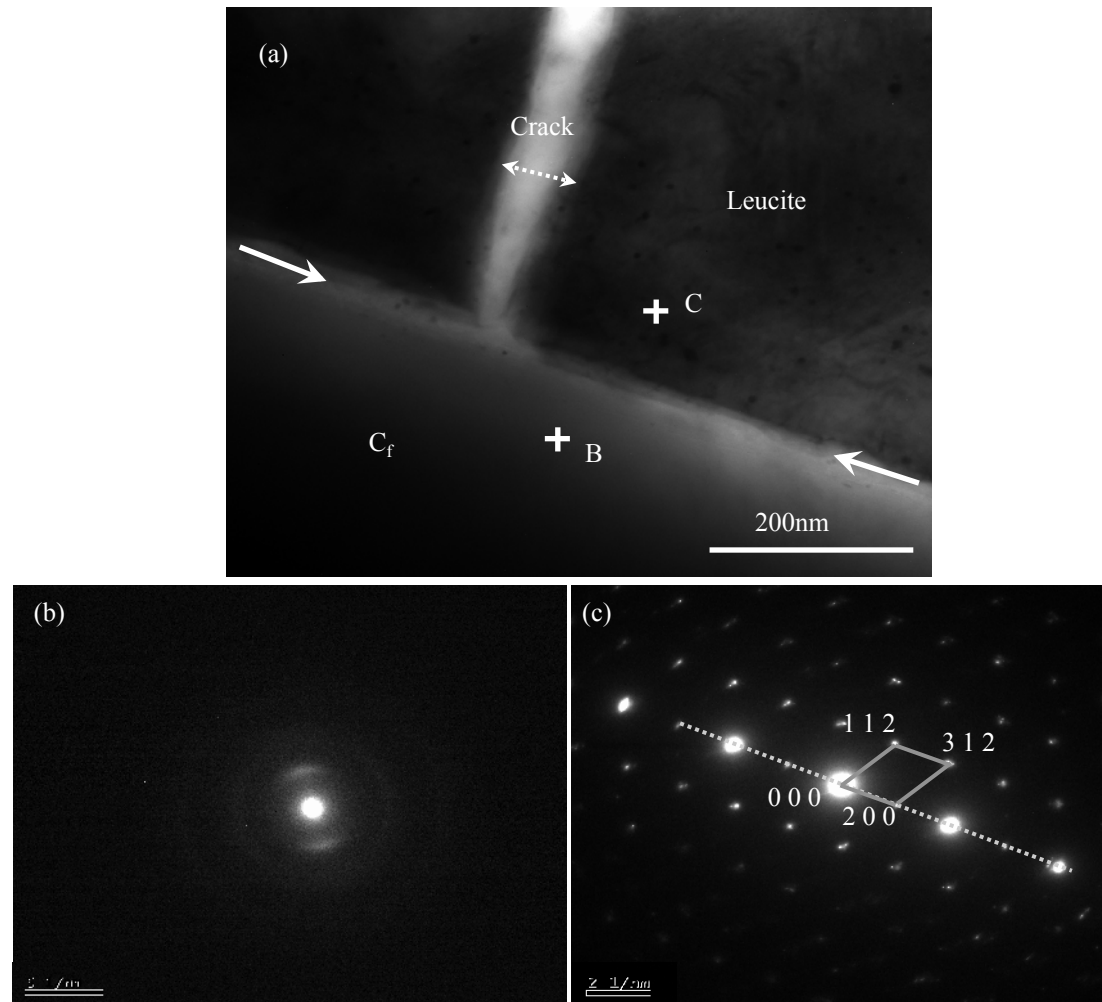


Fig. 3-8 Interface morphology of KC-1100:
 (a) TEM image, (b) and (c) SAD patterns of carbon fiber and matrix in area of B and C

$$\text{Residual stress: } \sigma = \frac{(\alpha_m - \alpha_f) \times \Delta T E_f V_f}{V_f \times \left(\frac{E_f}{E_m} - 1\right) + 1} \quad (3-2)$$

Table 3-1 Thermal residual stress in the composites during cooling process

| ΔT (°C) | 1100 | 1200 | 1300 | 1400 |
|------------------|-------|-------|-------|-------|
| σ_a (MPa) | 557.5 | 608.2 | 658.9 | 709.6 |
| σ_r (MPa) | 195.5 | 213.3 | 231.0 | 248.8 |



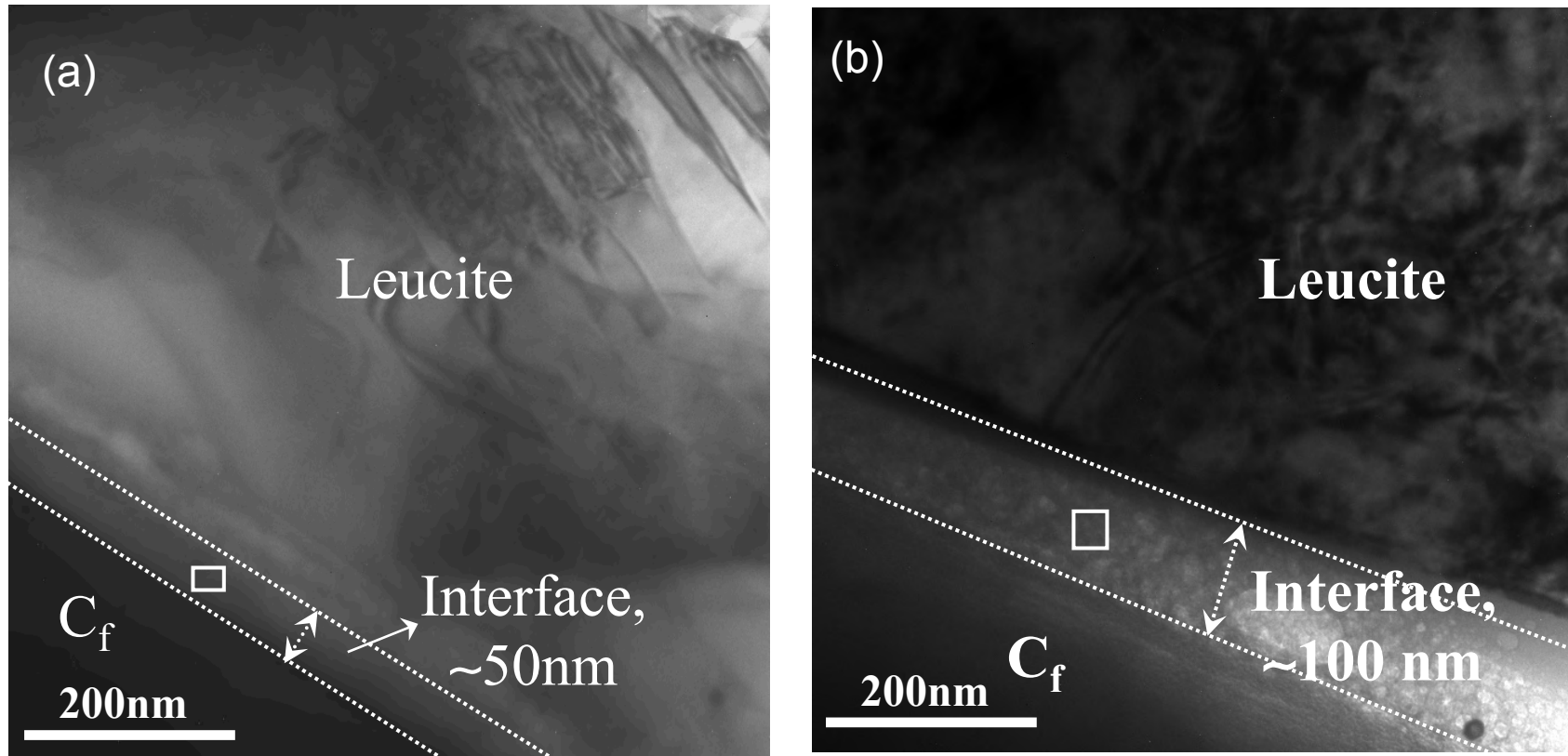


Fig. 3-9 Interface morphology of KC-1200 (a) and KC-1300°C(b)

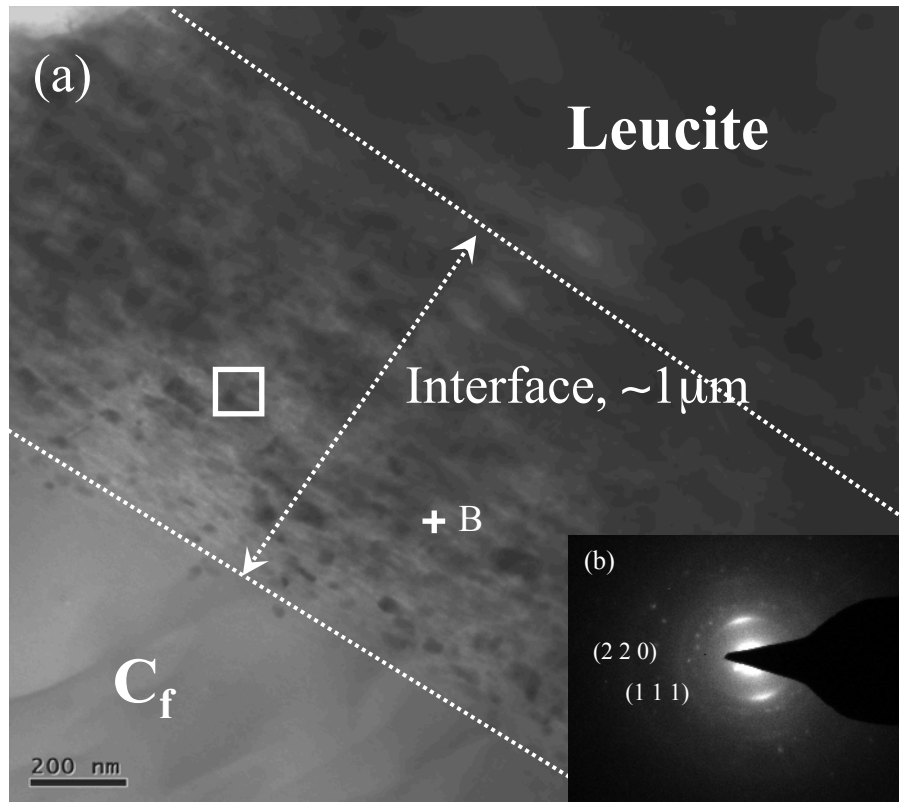


Fig. 3-10 Interface morphology of KC-1400

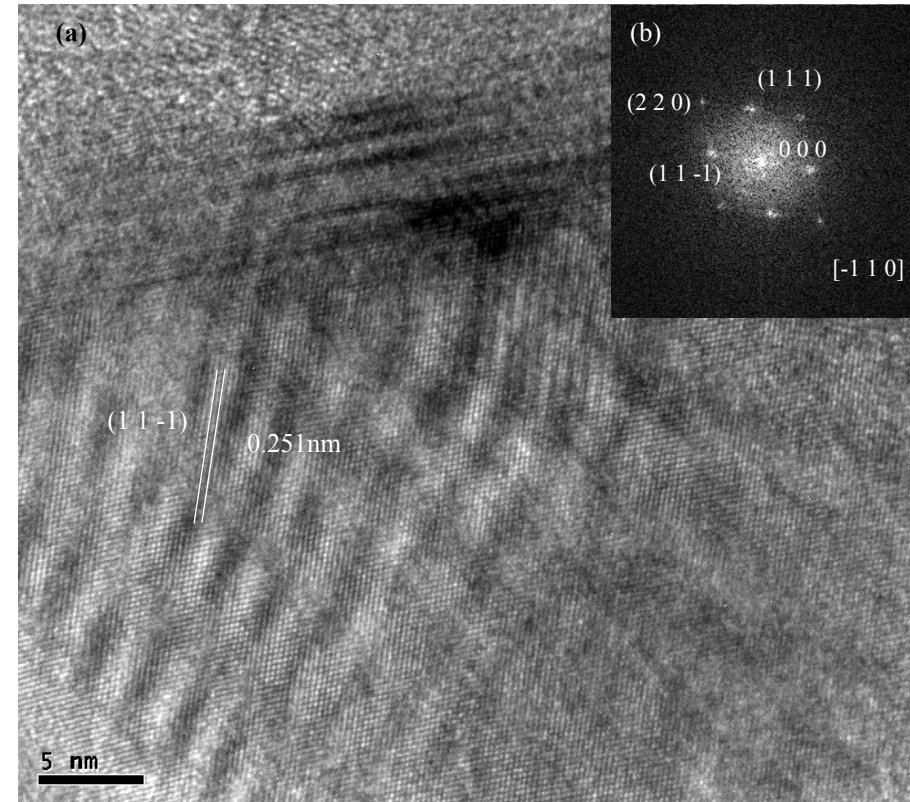


Fig. 3-11 HRTEM micrographs of area B in Fig.3-10: (a) HRTEM image and (b) FFT transformed image

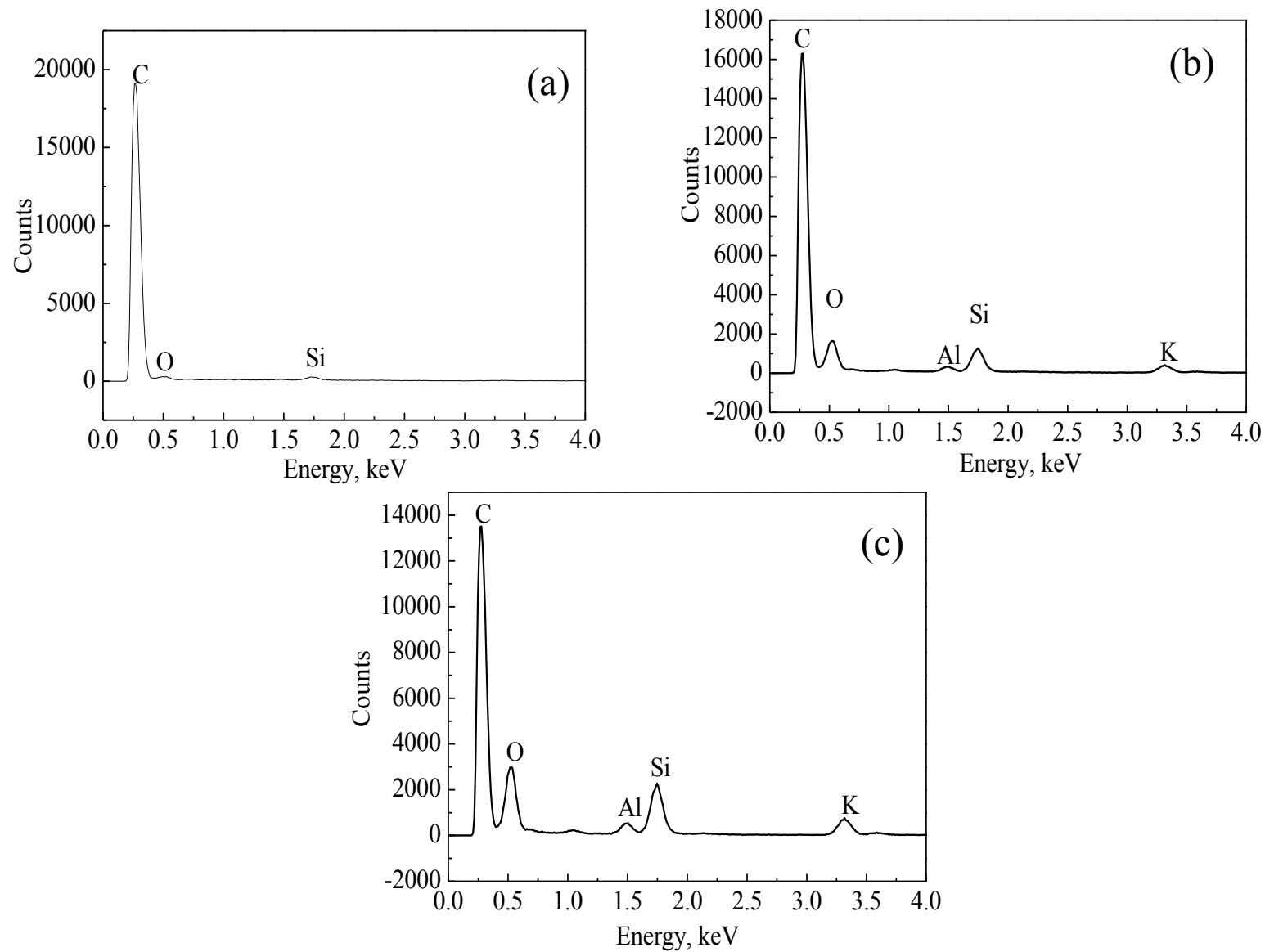


Fig. 3-12 Element distributions in the interface region:
 (a) KC-1200, (b) KC-1300, (c) KC-1400

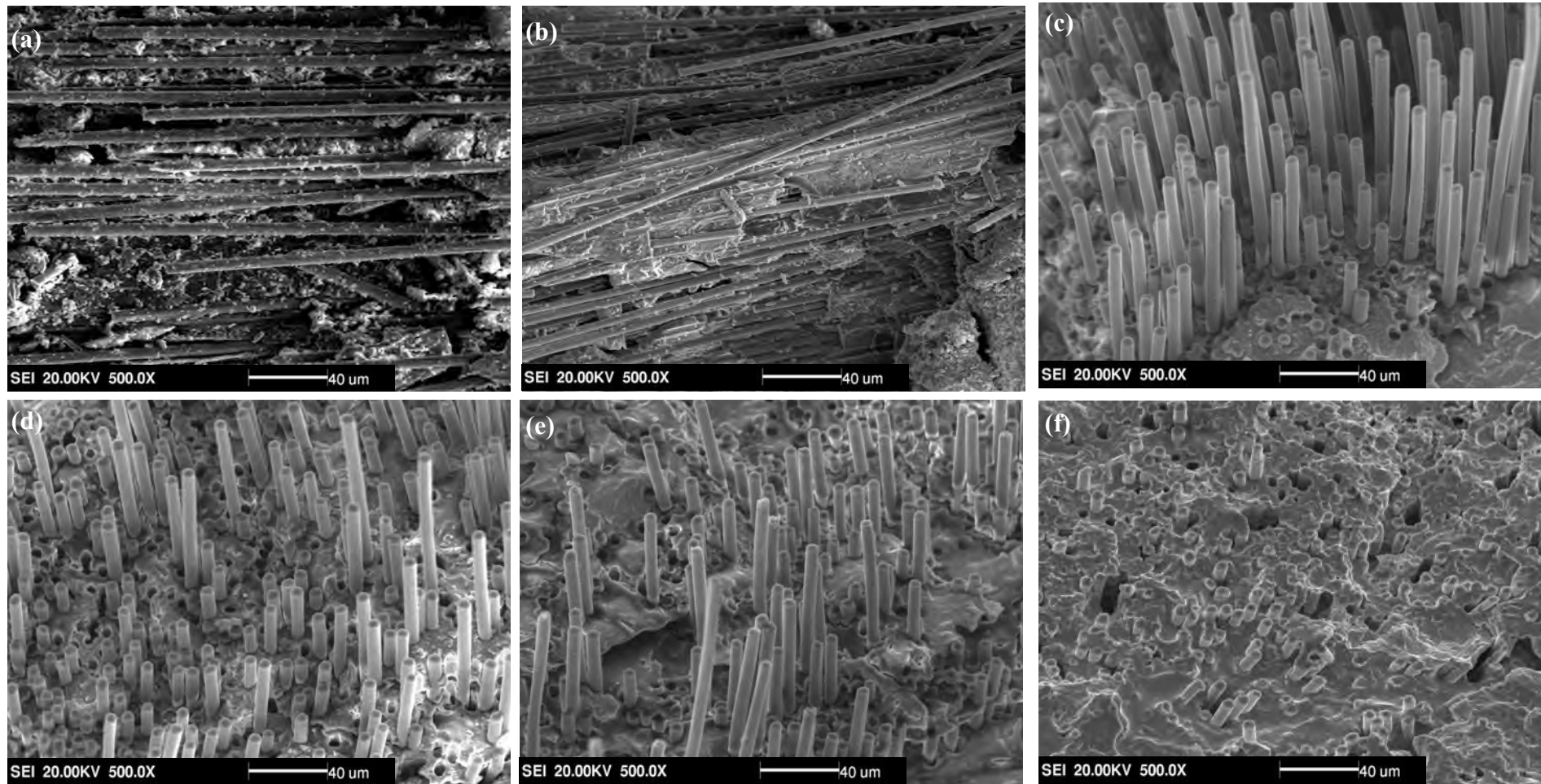


Fig. 3-13 Fractographs of C_f/KGP composites heat treated at different temperature: (a) KC-B, (b) KC-1000, (c) KC-1100, (d) KC-1200, (e) KC-1300, (f) KC-1400

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (4), 1447-1453.



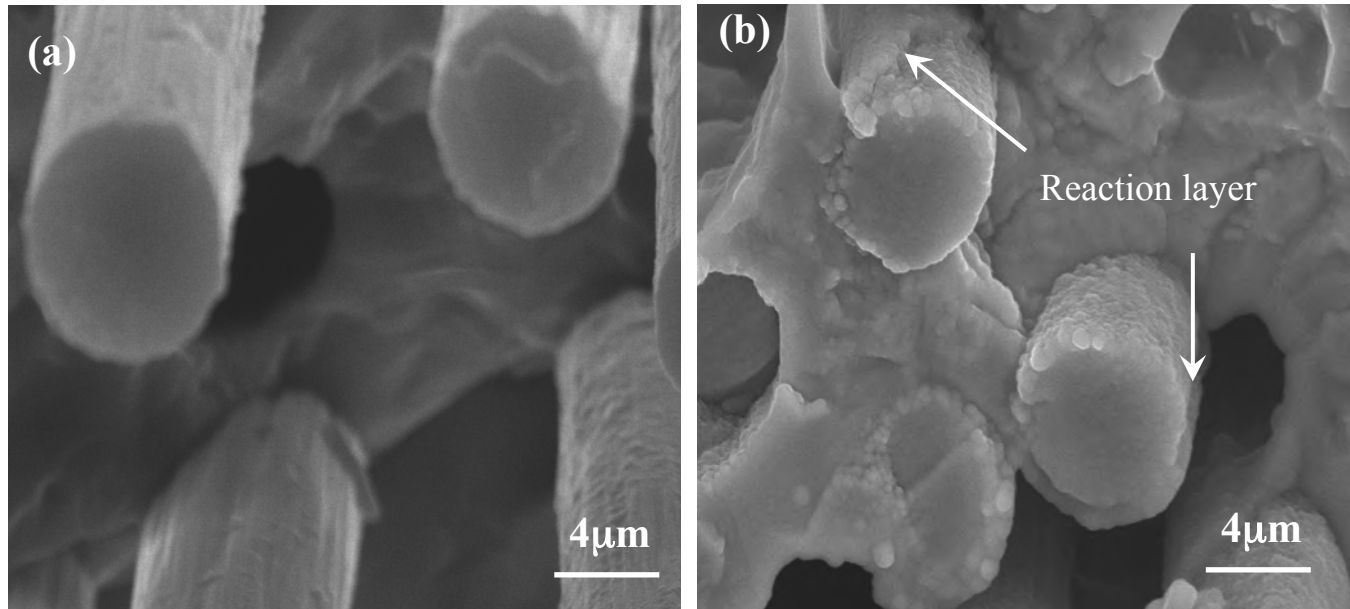


Fig. 3-14 Magnified fractograph of (a) KC-1100, (b) KC-1400

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (4), 1447-1453.



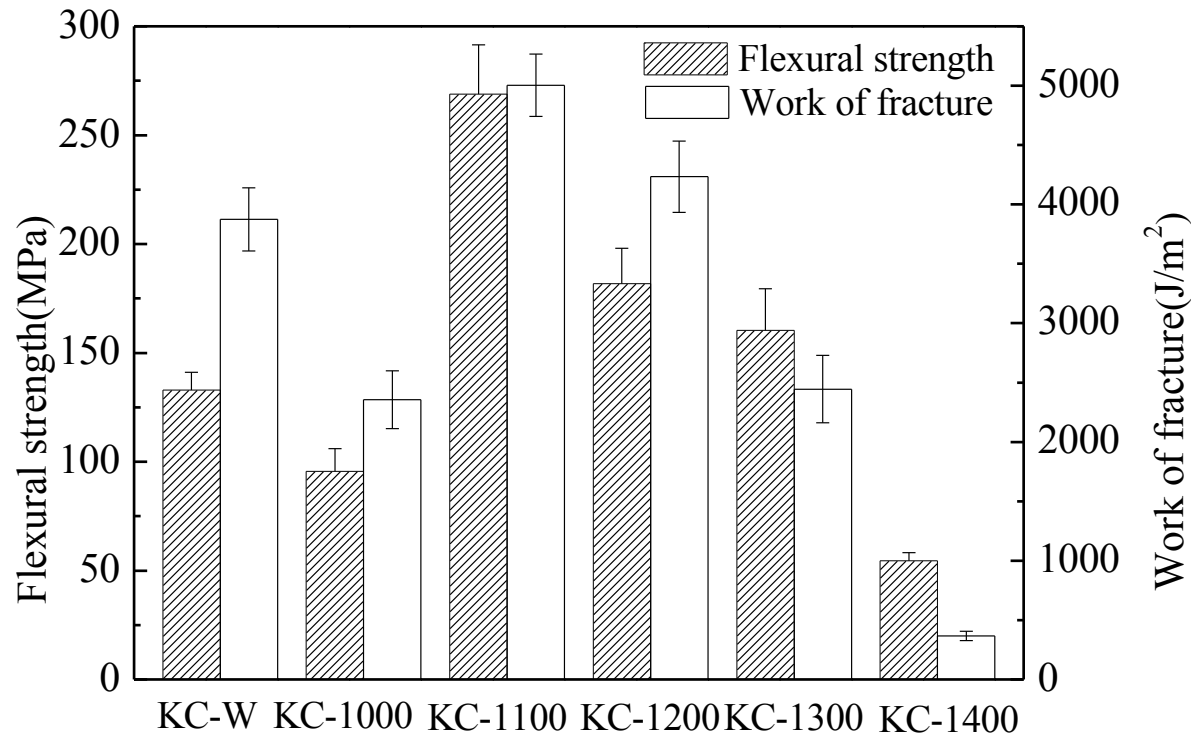


Fig. 3-15 Variations of flexural strength and work of fracture of C_f /KGP composites heat treated at different temperatures

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (4), 1447-1453.



Table 3-2 Mechanical properties of geopolymer and C_f/KGP composites after heat treatment

| Samples | Flexural strength (MPa) | Work of fracture (J·m ⁻²) | Fracture toughness (MPa·m ^{1/2}) | Elastic modulus (GPa) |
|-----------------|-------------------------|---------------------------------------|--|-----------------------|
| Geopolymer | 12.3±1.2 | — | 0.2±0.04 | 10.3±1.2 |
| Leucite ceramic | 70.0±6.8 | — | 1.3±0.16 | 65.0±6.3 |
| KC-W | 132.9±8.2 | 3874.5±266.8 | — | 36.5±3.4 |
| KC-1000 | 95.6±10.5 | 2354.8±243.6 | — | 30.4±4.1 |
| KC-1100 | 268.9 ±14.6 | 5003.4±254.4 | 7.21±0.87 | 67.4±4.2 |
| KC-1200 | 181.7±16.3 | 4233.4±299.5 | 6.57±0.64 | 55.9±4.8 |
| KC-1300 | 160.3±19.2 | 2444.6±283.2 | 4.9±0.57 | 50.6±3.6 |
| KC-1400 | 54.6±3.7 | 366.6±37.5 | 1.05±0.31 | 41.5±4.2 |

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (4), 1447-1453.



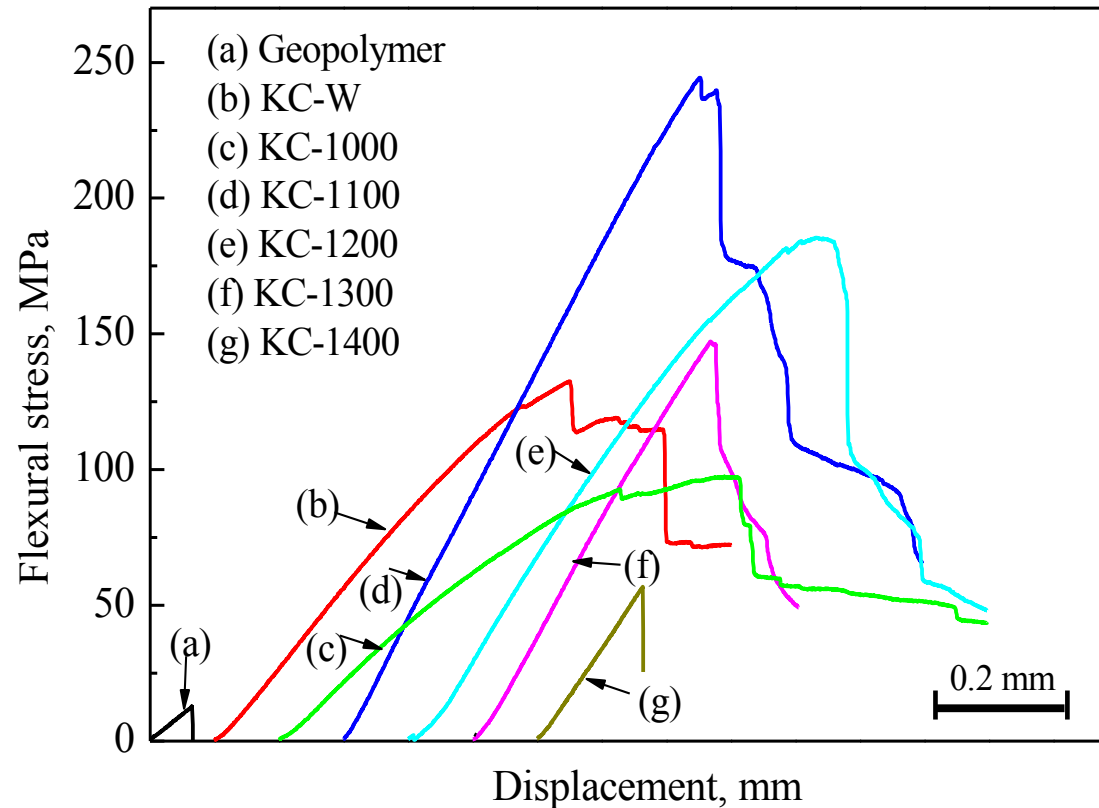


Fig. 3-16 Flexural stress-displacement curves of geopolymer and C_f /KGP composites heat treated at different temperatures

---- Peigang He, Dechang Jia. *Ceramics International*. 2010, 36 (4), 1447-1453.



3.2 Effects of heat treatment on the properties of C_f/CsKGP

- ◆ **Carbon fiber:** Unidirectional
- ◆ **Matrix:** $0.8\text{K}_2\text{O} \cdot 0.2\text{Cs}_2\text{O} \cdot \text{Al}_2\text{O}_3 \cdot 5\text{SiO}_2 \cdot 11\text{H}_2\text{O}$
- ◆ **Composites without and after heat treatment were denoted as:**
CsKC-W, CsKC-1000, CsKC-1100, CsKC-1150, CsKC-1200
and CsKC-1300, respectively
- ◆ **Fiber contents:** 20vol% for CsKC-W and $\sim 25\text{vol}\%$ for others



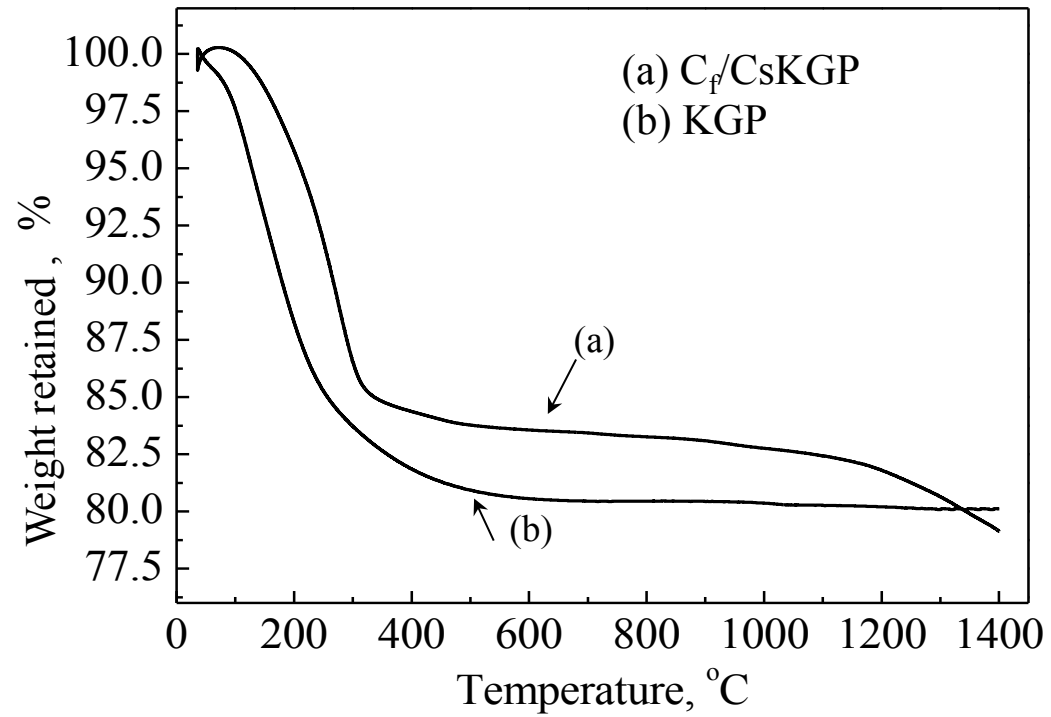


Fig. 3-17 Weight retained of KGP and C_f/CsKGP composites

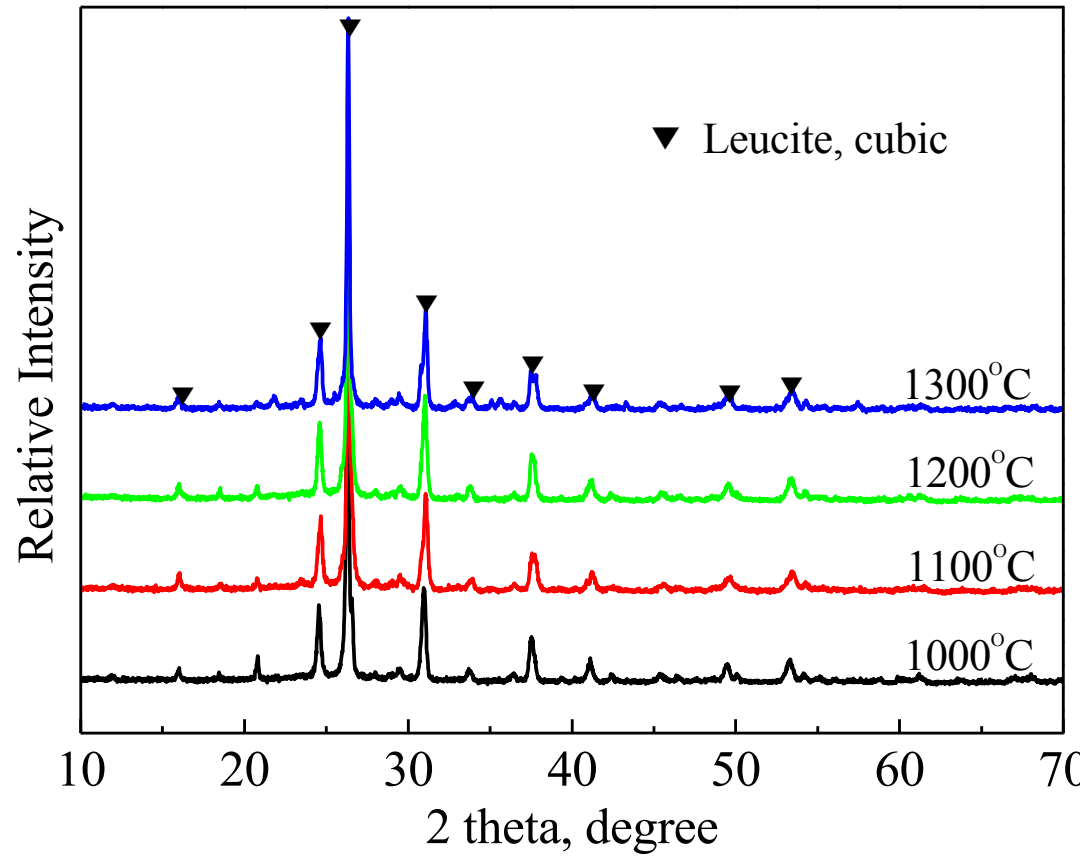


Fig. 3-18 XRD patterns of $C_p/CsKGP$ heat treated at different temperatures

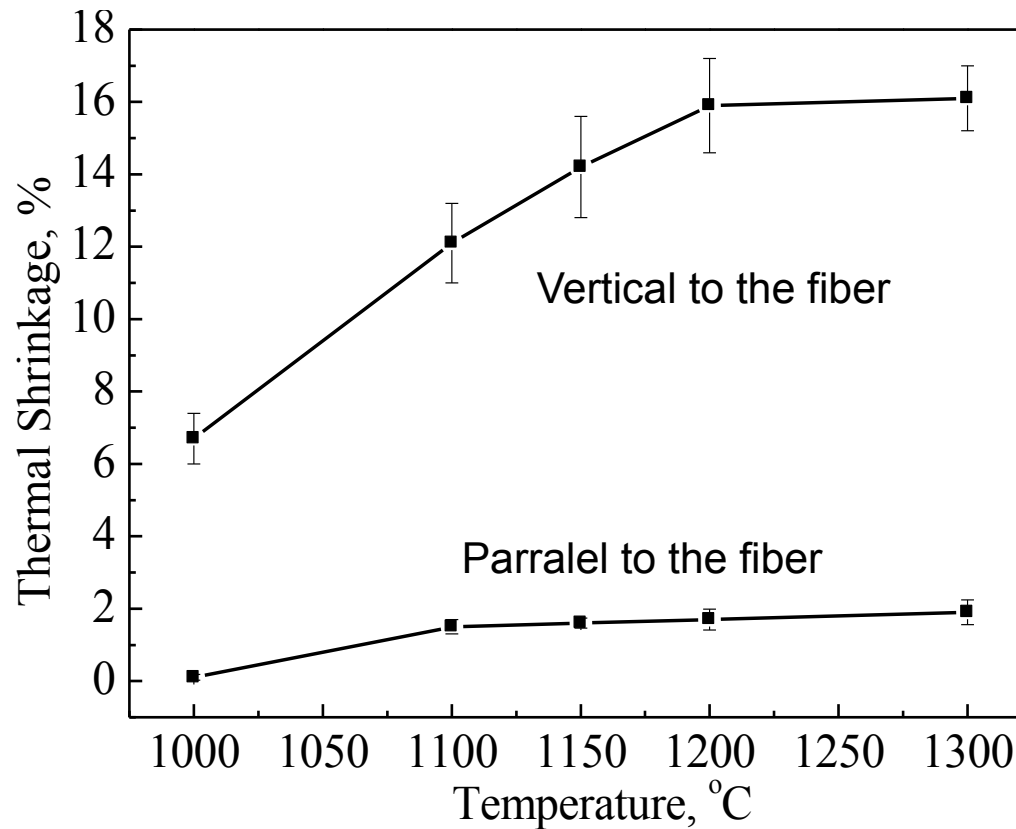


Fig. 3-19 Thermal shrinkages of $C_f/CsKGP$ heat treated at different directions

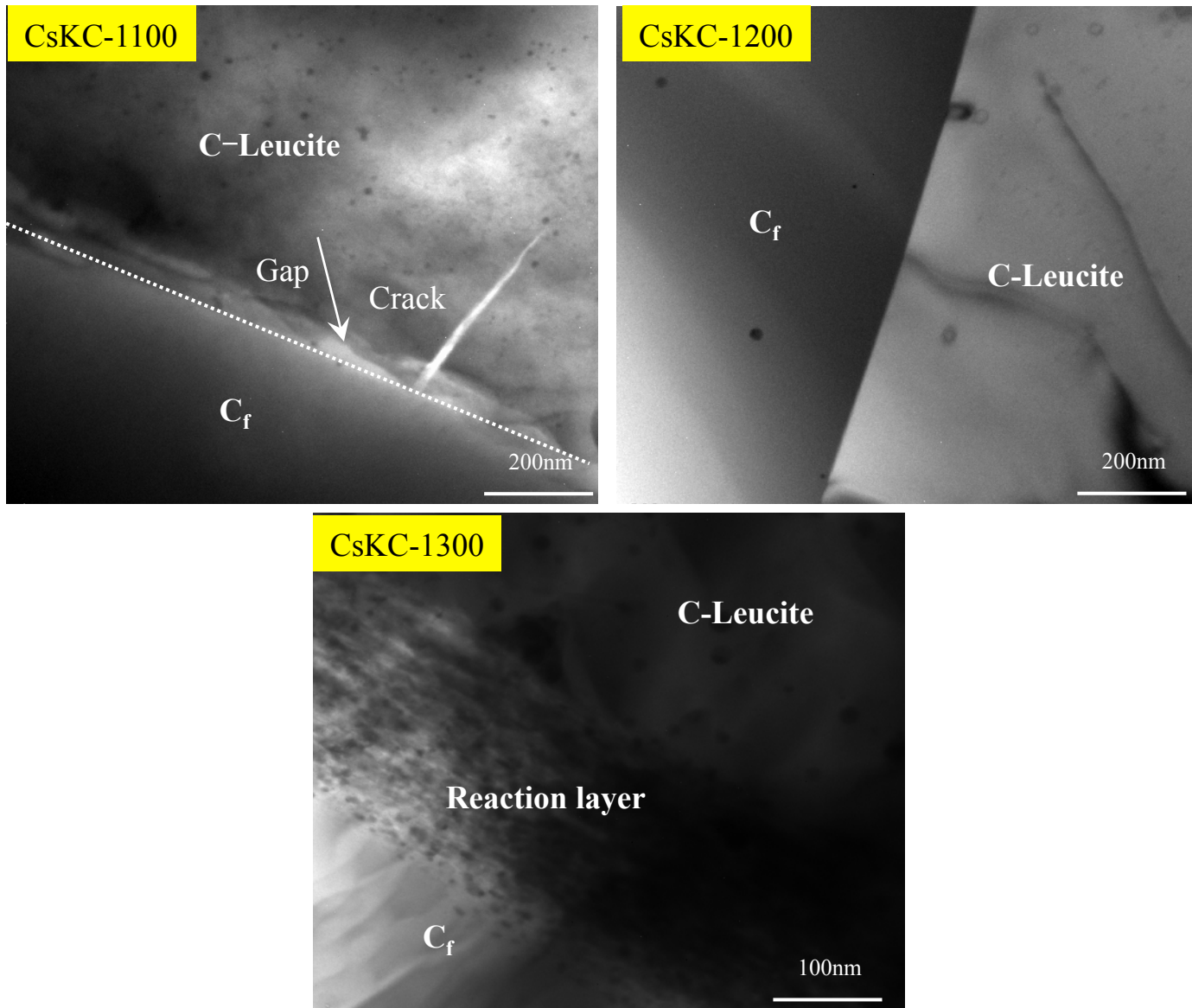


Fig. 3-20 Interface morphology of C_f /CsKGP heat treated at different temperatures :
(a) CsKC-1100, (b) CsKC-1200, (c) CsKC-1300

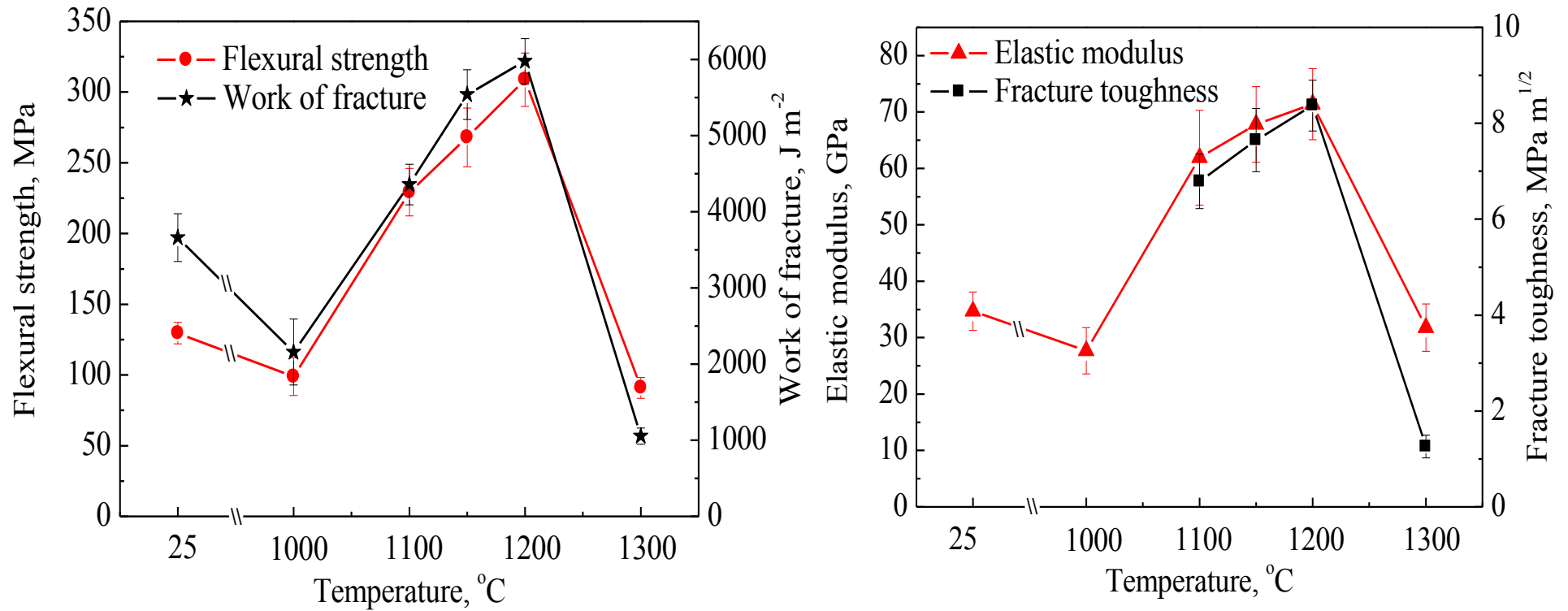


Fig. 3-21 Variations of the mechanical properties of $C_f/CsKGP$ heat treated at different temperatures: (a) Flexural strength and work of fracture, (b) Young's modulus and fracture toughness

Table 3-3 Mechanical properties of C_f/CsKGP composites heat treated at different temperature

| Samples | Flexural strength (MPa) | Work of fracture (J·m ⁻²) | Fracture toughness (MPa m ^{1/2}) | Elastic modulus (GPa) |
|-----------|----------------------------|--|---|--------------------------|
| CsKC-W | 129.6±7.5 | 3659.9±314.7 | — | 34.7±3.4 |
| CsKC-1000 | 98.8±13.4 | 2158.7±431.5 | — | 27.7±4.1 |
| CsKC-1100 | 229.2±16.8 | 4357.9±267.7 | 6.79±0.57 | 61.9±8.4 |
| CsKC-1150 | 268.0±20.7 | 5538.4±327.5 | 7.65±0.66 | 67.8±6.7 |
| CsKC-1200 | 278.7±18.9 | 5977.4±297.1 | 8.37±0.53 | 71.4±6.3 |
| CsKC-1300 | 90.9±7.4 | 1054.8±106.7 | 1.26±0.24 | 31.8±4.2 |



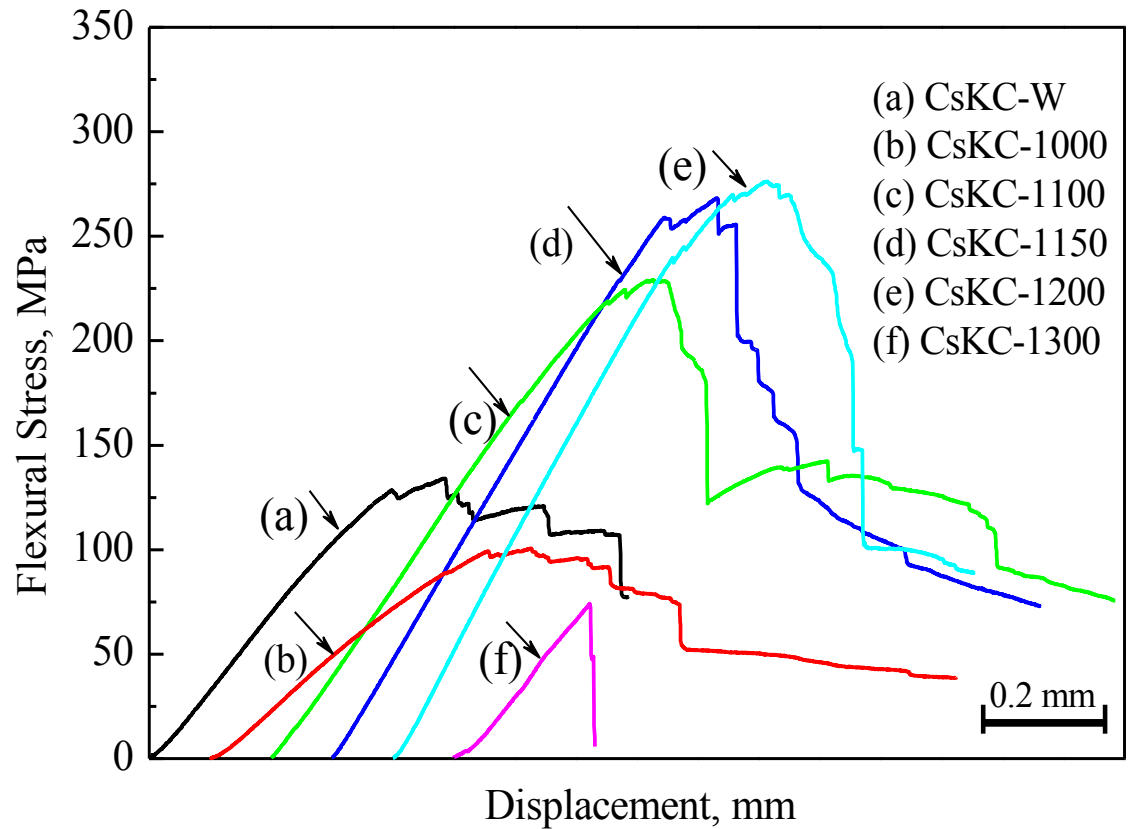


Fig. 3-22 Flexural stress–displacement curves of $C_f/CsKGP$ heat treated at different temperatures

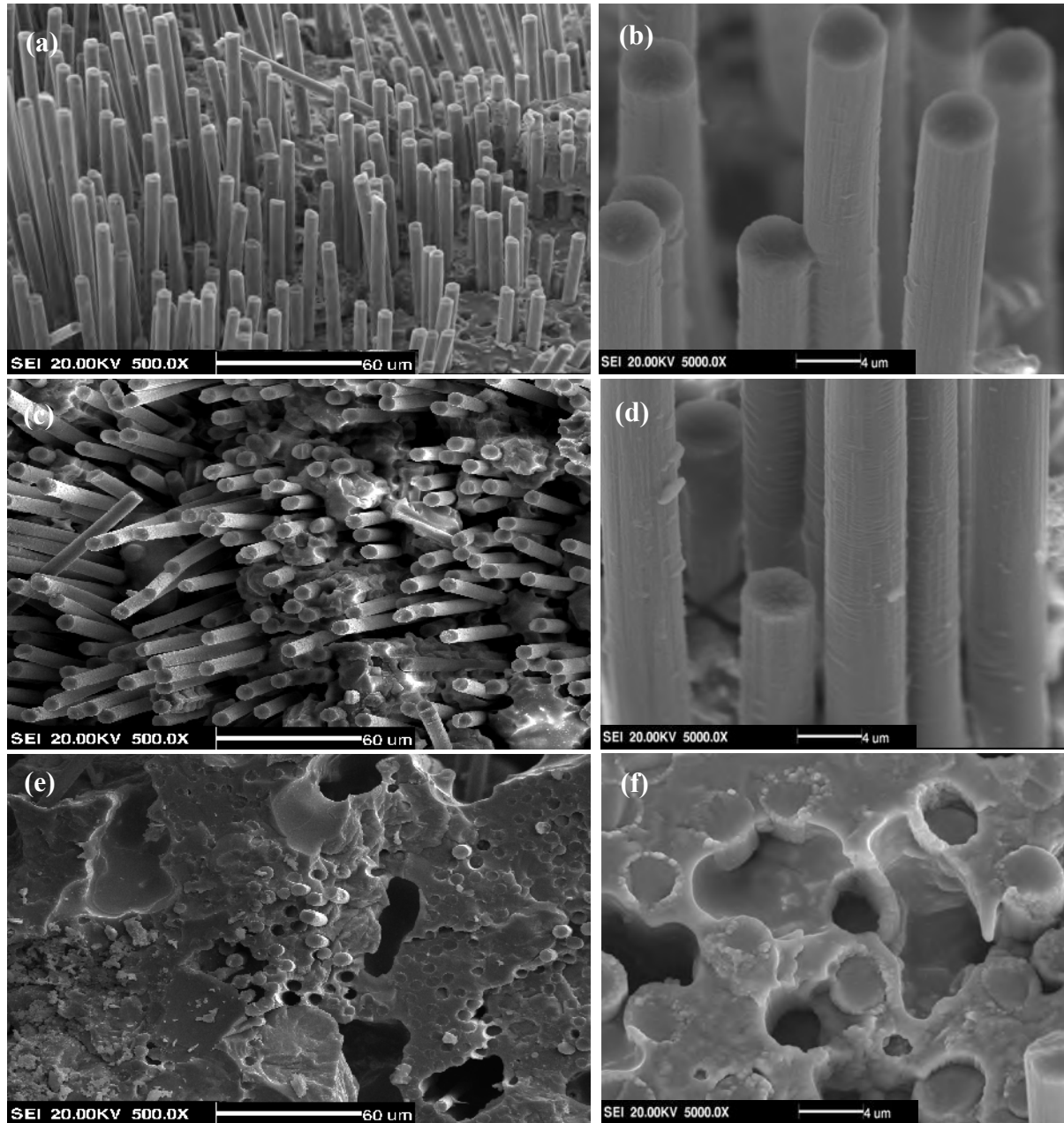


Fig. 3-23 Fractographs of $C_f/CsKGP$ composites heat treated at different temperature:
(a), (b) 1100°C, (c), (d) 1200°C and (e), (f) 1300°C

3.3 Summary

- ◆ After proper heat treatment, the C_f /geopolymer composite can be directly converted into the carbon fiber reinforced leucite ceramic matrix composites, and mechanical properties could be greatly enhanced.
- ◆ The property improvement can be attributed to the densified and crystallized matrix, and the enhanced fiber/matrix interface bonding based on the fine-integrity of carbon fibers.
- ◆ For the composite heat treated at higher temperature ($\geq 1300^\circ\text{C}$), the mechanical properties lowered substantially and it tended to fracture in a very brittle manner owing to the seriously degraded carbon fibers.



4. Sol-SiO₂ impregnation on the ambient and high temperature mechanical properties of C_f/leucite composites

4.1 Microstructure of the composites after repeated impregnation

4.2 Room temperature mechanical properties

4.3 Non-isothermal oxidation of the composites

4.4 High-temperature mechanical properties and fracture mechanism

4.5 Summary



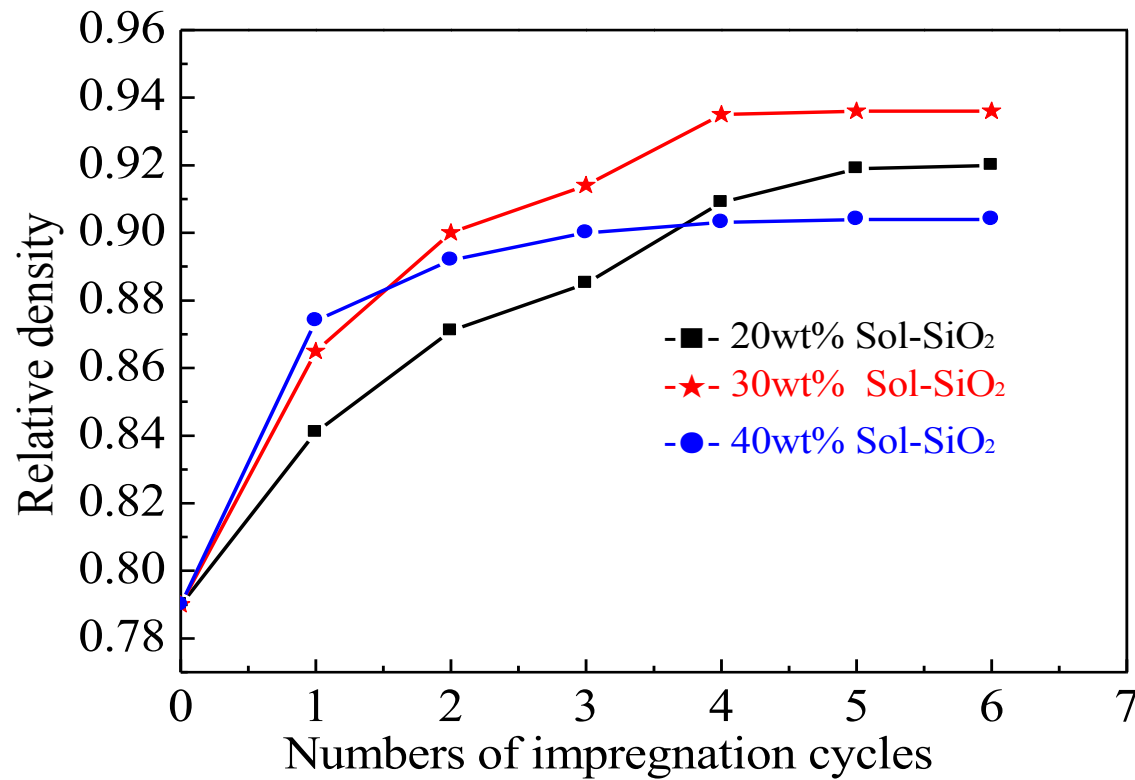


Fig. 4-1 Relative density vs. number of impregnation/drying cycles with different sol-SiO₂ concentration

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.



4.1 Microstructure of the composites after repeated impregnation

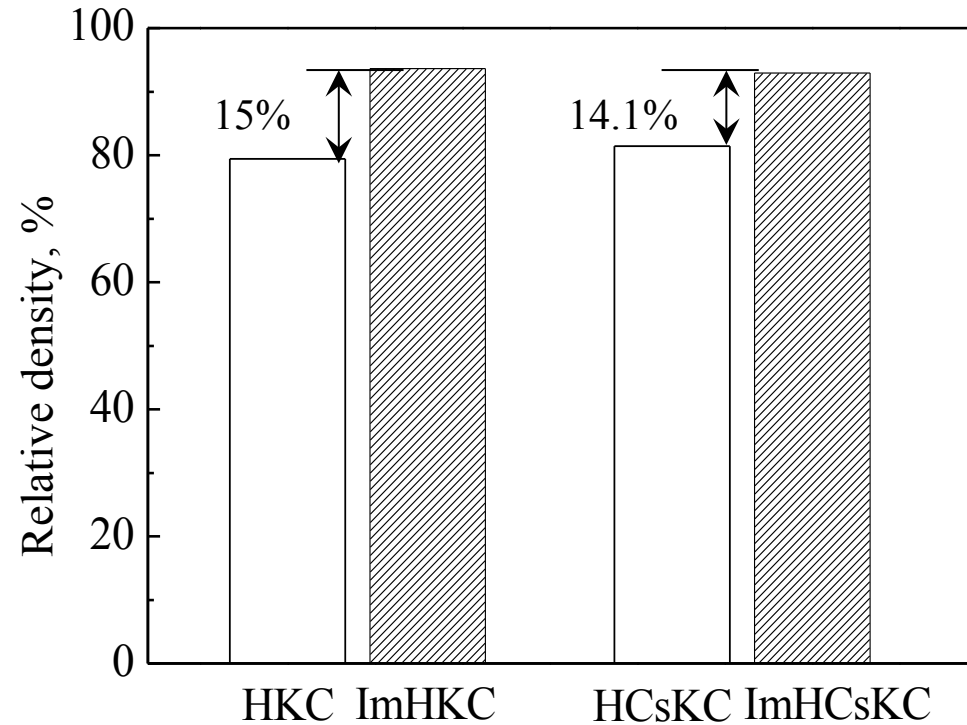


Fig. 4-2 Relative density of the composites before and after sol-SiO₂ impregnation

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.



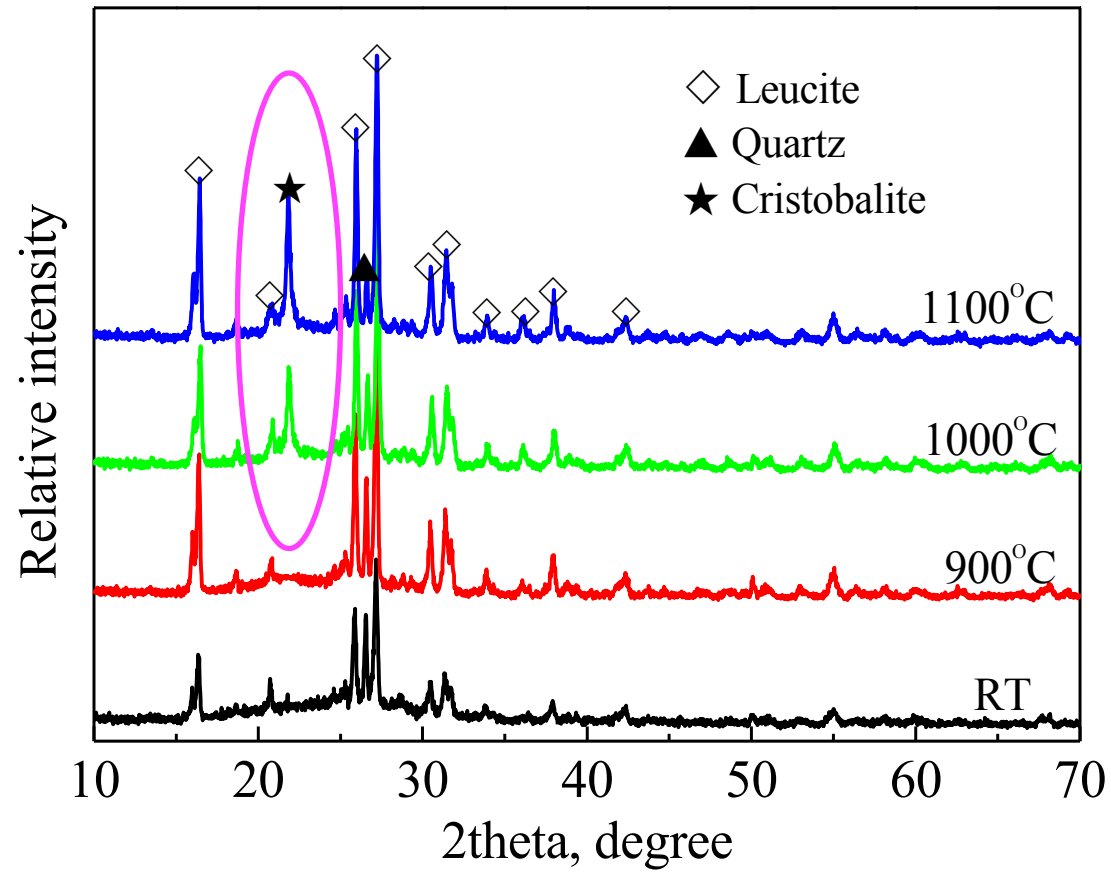


Fig. 4-3 XRD patterns of the impregnated composite treated at different temperatures

Before sol-SiO₂ impregnation

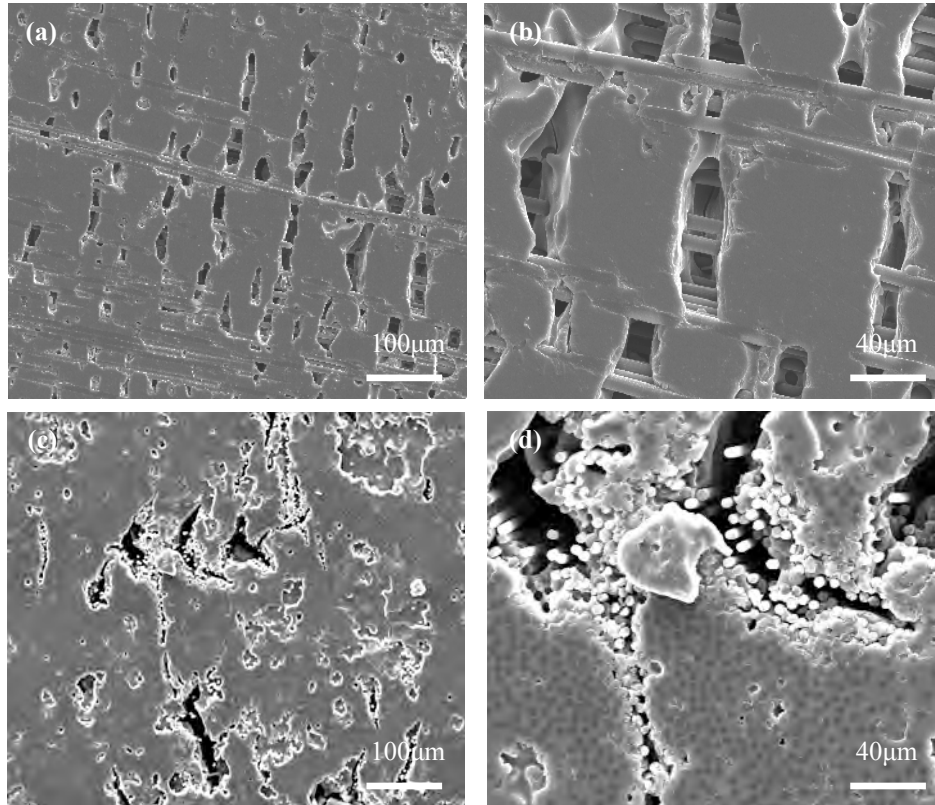


Fig. 4-4 Microstructures of the polished cross section of HKC composite:(a), (b) in the axial direction of fibers; (c), (d) in the radial direction

After sol-SiO₂ impregnation

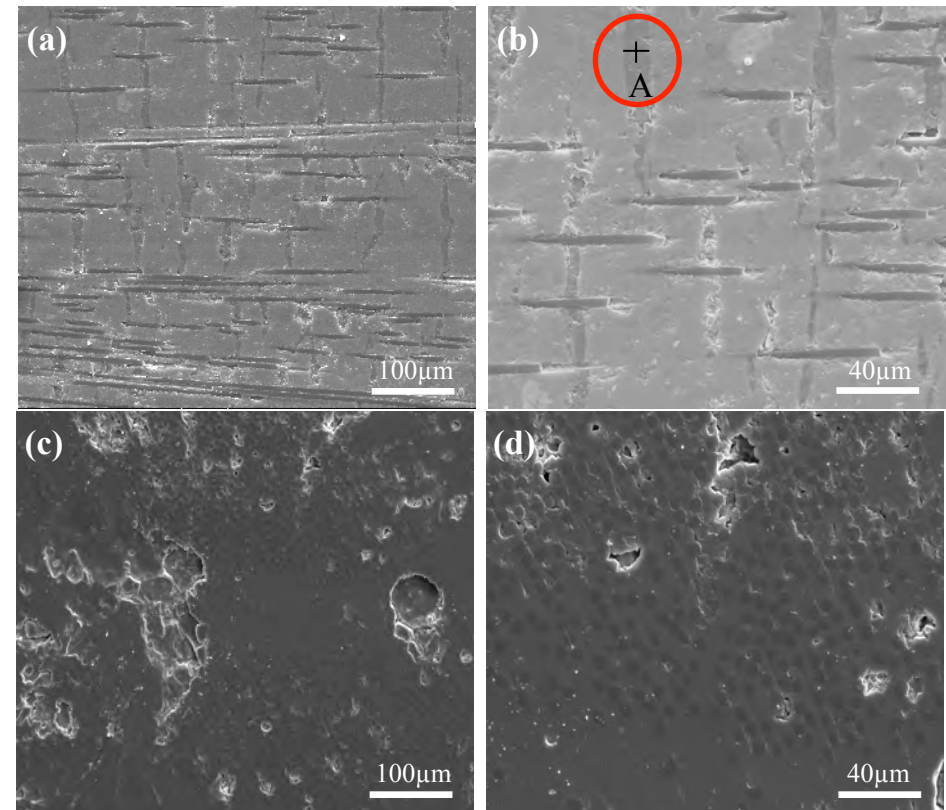


Fig. 4-5 Microstructure of the polished cross section of ImHKC composite: (a), (b) in the axial direction of fibers; (c),(d) in the radial direction

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.



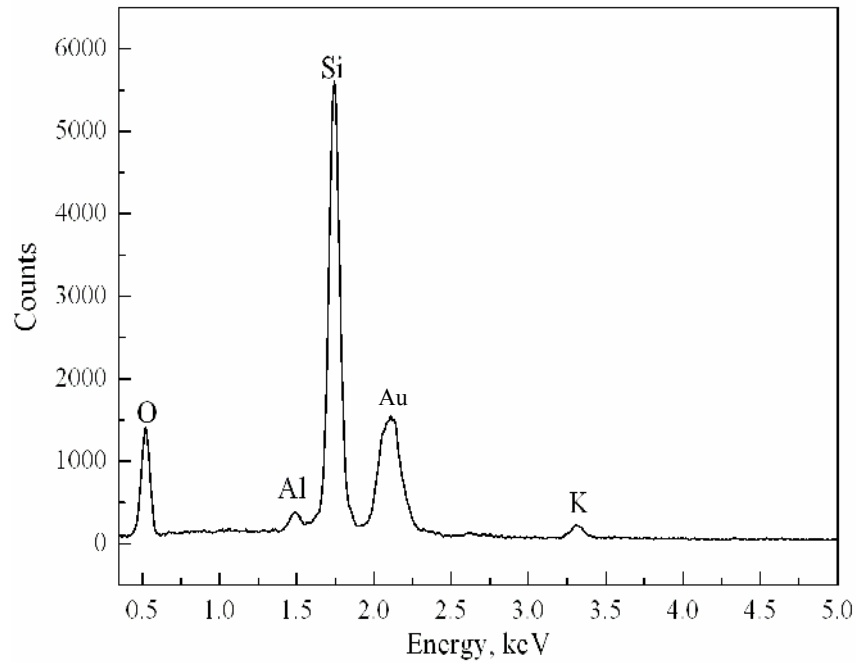


Fig. 4-6 Energy dispersive spectrum of point A in the Fig. 4-5

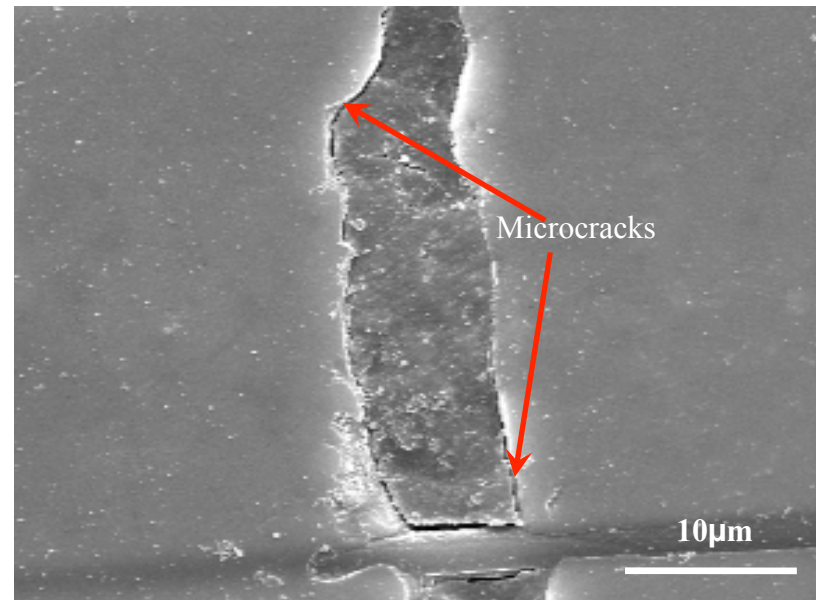


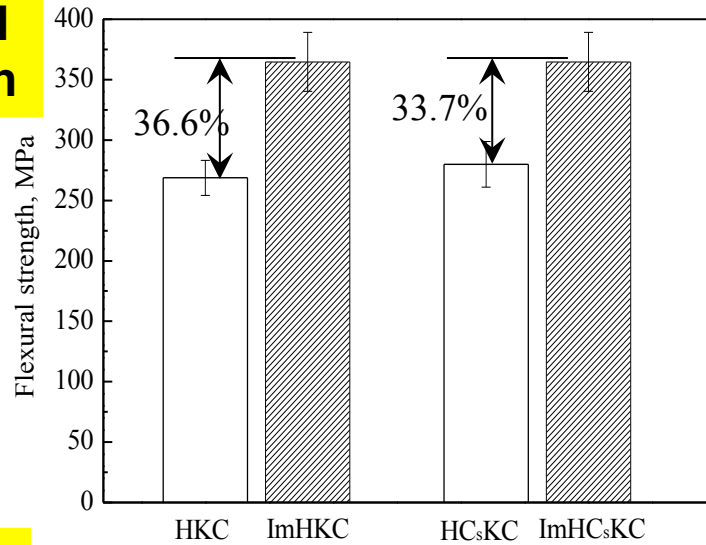
Fig. 4-7 Microstructure at a higher magnification of a sol-SiO₂ sealed crack

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.

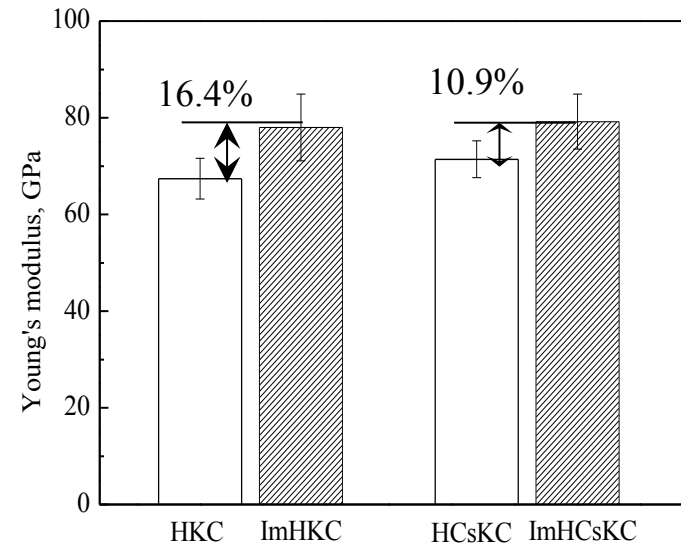


4.2 Room temperature mechanical properties

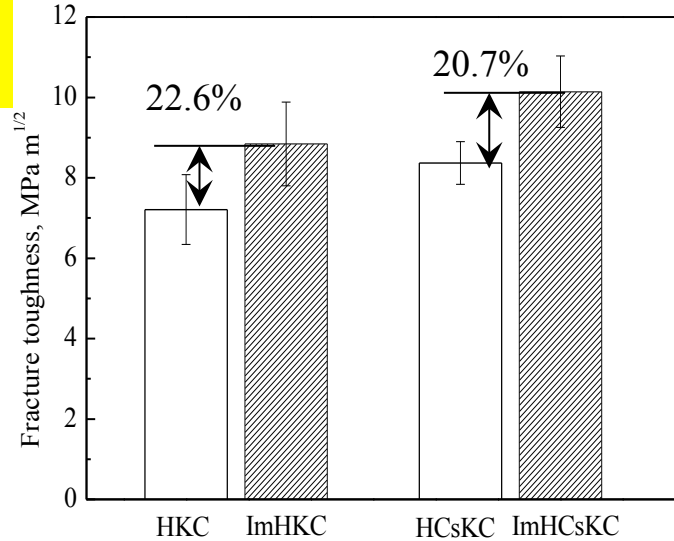
Flexural strength



Young's Modulus



Fracture toughness



Work of fracture

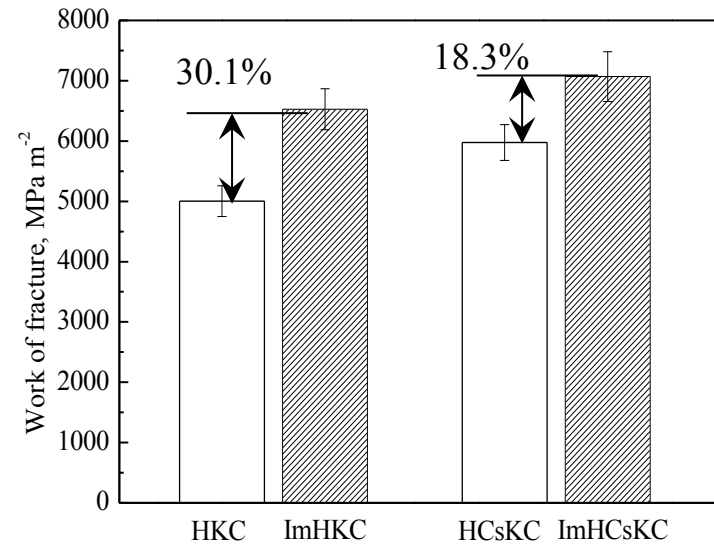


Fig.4-8 Mechanical properties of (Im)HKC and (Im)CsHKC

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.

Table 4-1 Room temperature properties of HKC and ImHKC

| Specimen | HKC | ImHKC |
|--|--------------|--------------|
| Density (g/cm ³) | 1.81 | 2.16 |
| Relative Density (%) | 79.0 | 93.4 |
| Flexural strength (MPa) | 268.9 ±14.6 | 364.7±24.4 |
| Young's modulus (GPa) | 67.4±4.2 | 78.0±6.9 |
| Fracture toughness (MPa·m ^{1/2}) | 7.21±0.87 | 8.84±1.04 |
| Work of fracture (J·m ⁻²) | 5003.4±254.4 | 6527.4±342.1 |

Table 4-2 Room temperature properties of HCsKC and ImHCsKC

| Specimen | HCsKC | ImHCsKC |
|--|--------------|--------------|
| Density (g/cm ³) | 1.85 | 2.28 |
| Relative Density (%) | 81.4 | 92.9 |
| Flexural strength (MPa) | 278.7±18.9 | 376.4±21.3 |
| Young's modulus (GPa) | 71.4±3.8 | 79.2±5.7 |
| Fracture toughness (MPa·m ^{1/2}) | 8.37±0.53 | 10.14±0.89 |
| Work of fracture (J·m ⁻²) | 5977.4±297.4 | 7068.6±413.8 |

4.3 Non-isothermal oxidation behavior

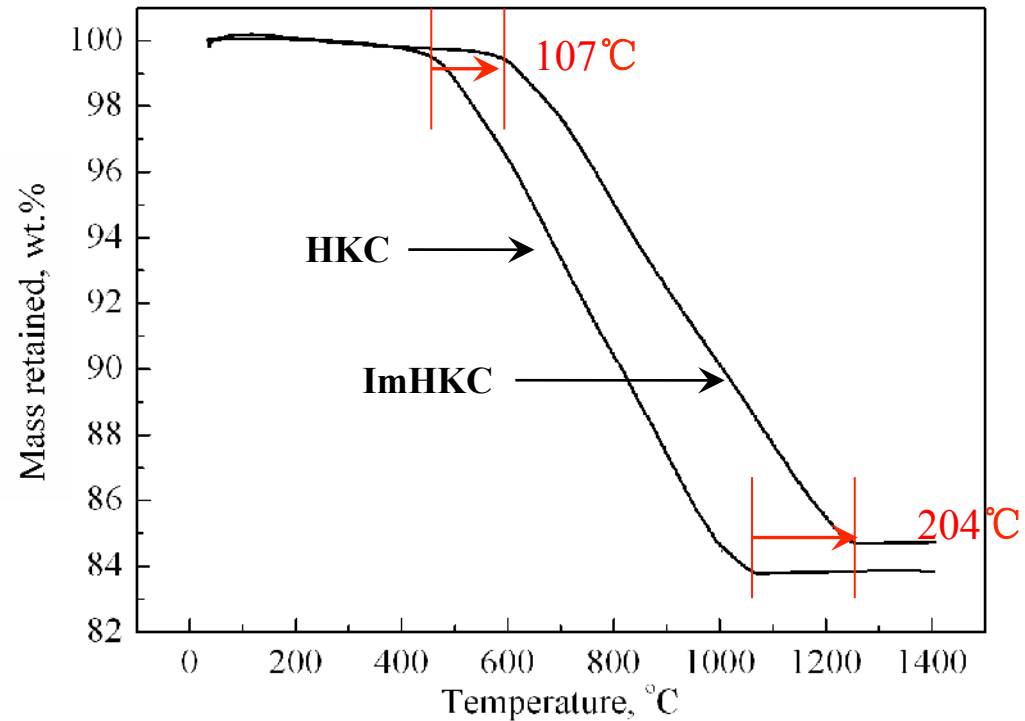


Fig. 4-9 Non-isothermal oxidation weight loss of the HKC and ImHKC composites

4.4 High-temperature mechanical properties

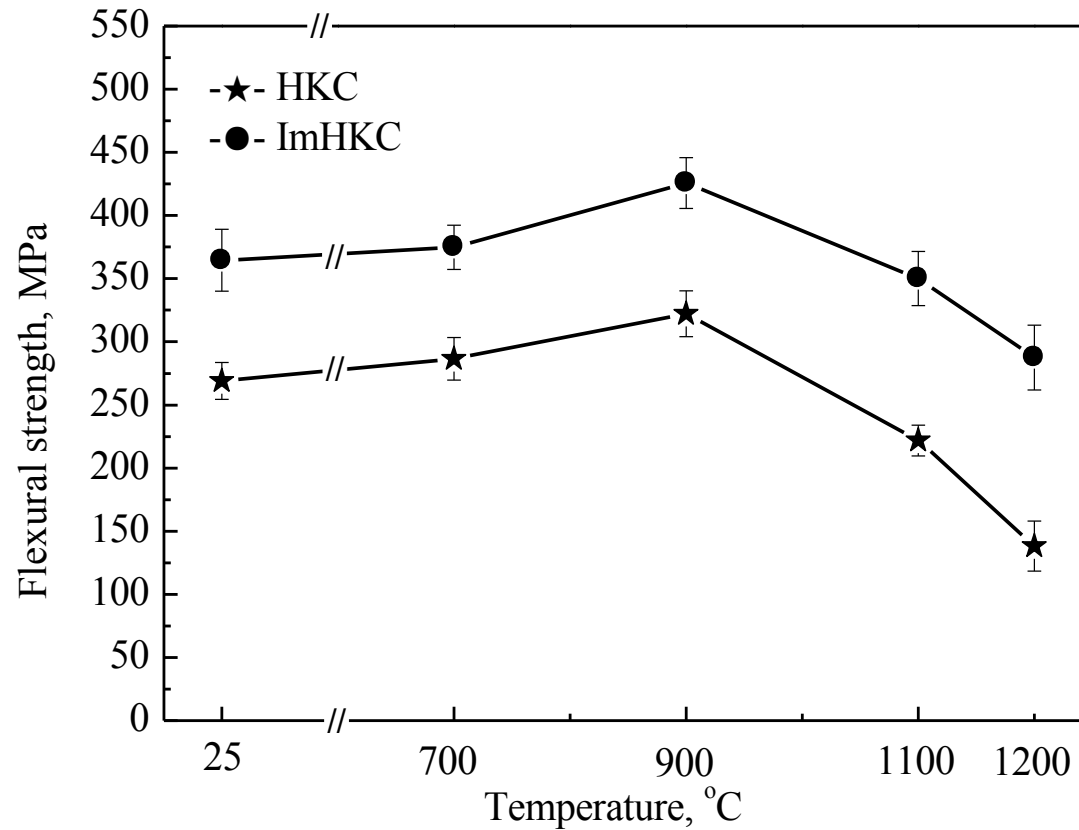


Fig. 4-10 Elevated temperature flexural strength of HKC and ImHKC

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.



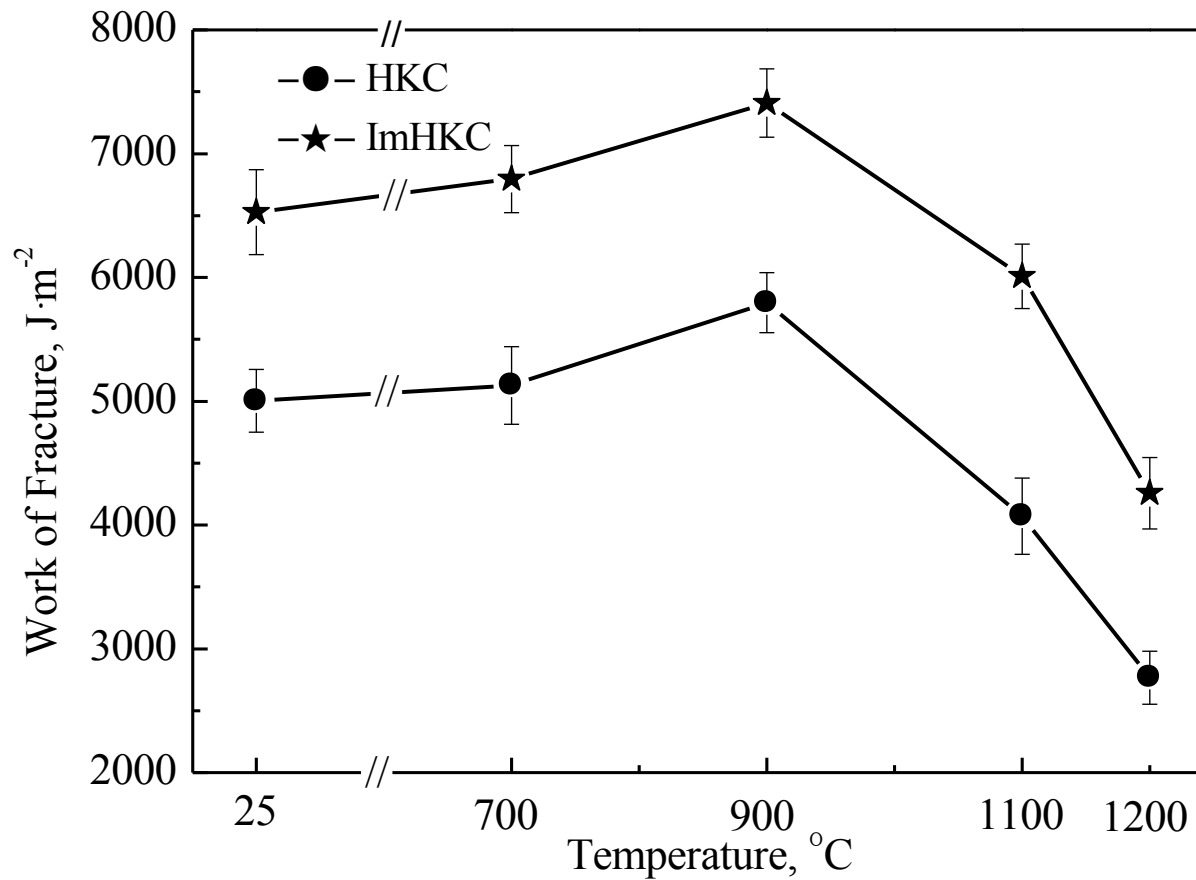


Fig. 4-11 Elevated temperature work of fracture of HKC and ImHKC

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.



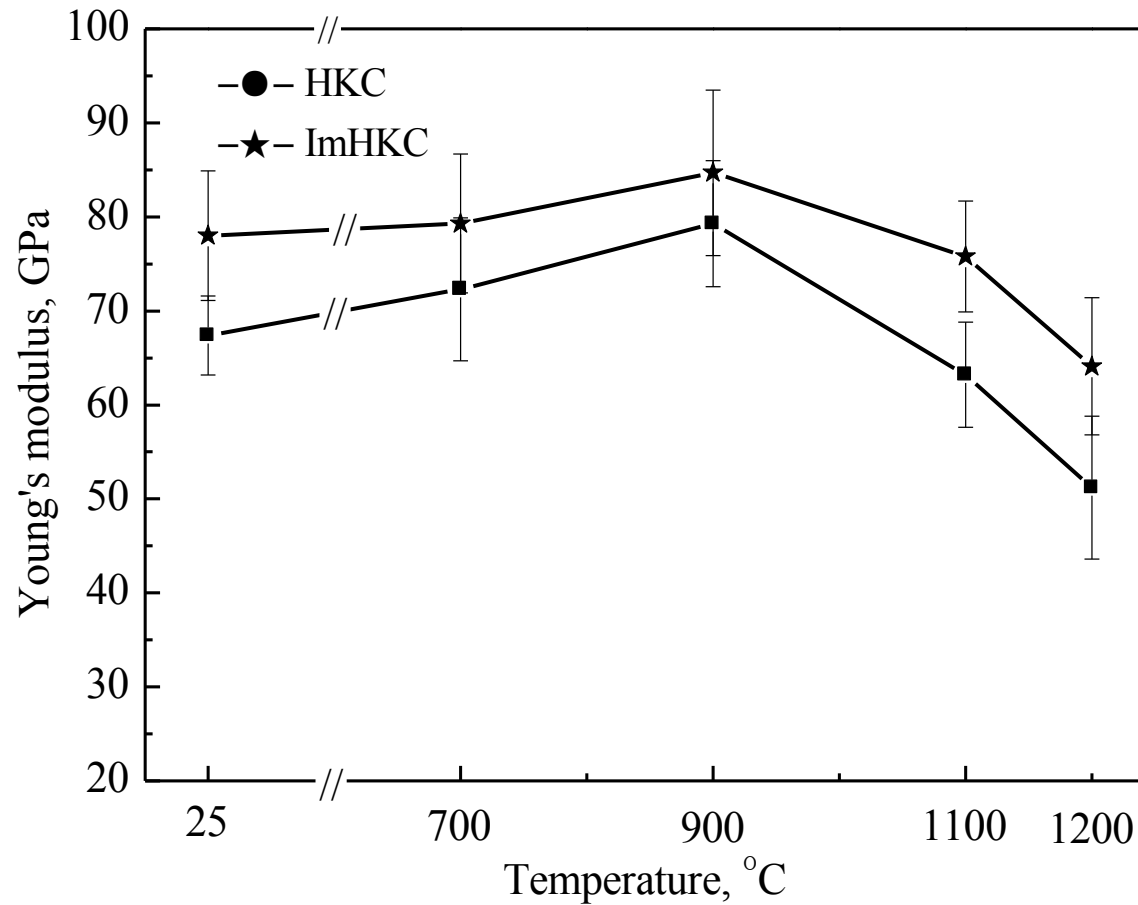


Fig. 4-12 Elevated temperature Young's modulus of HKC and ImHKC

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.



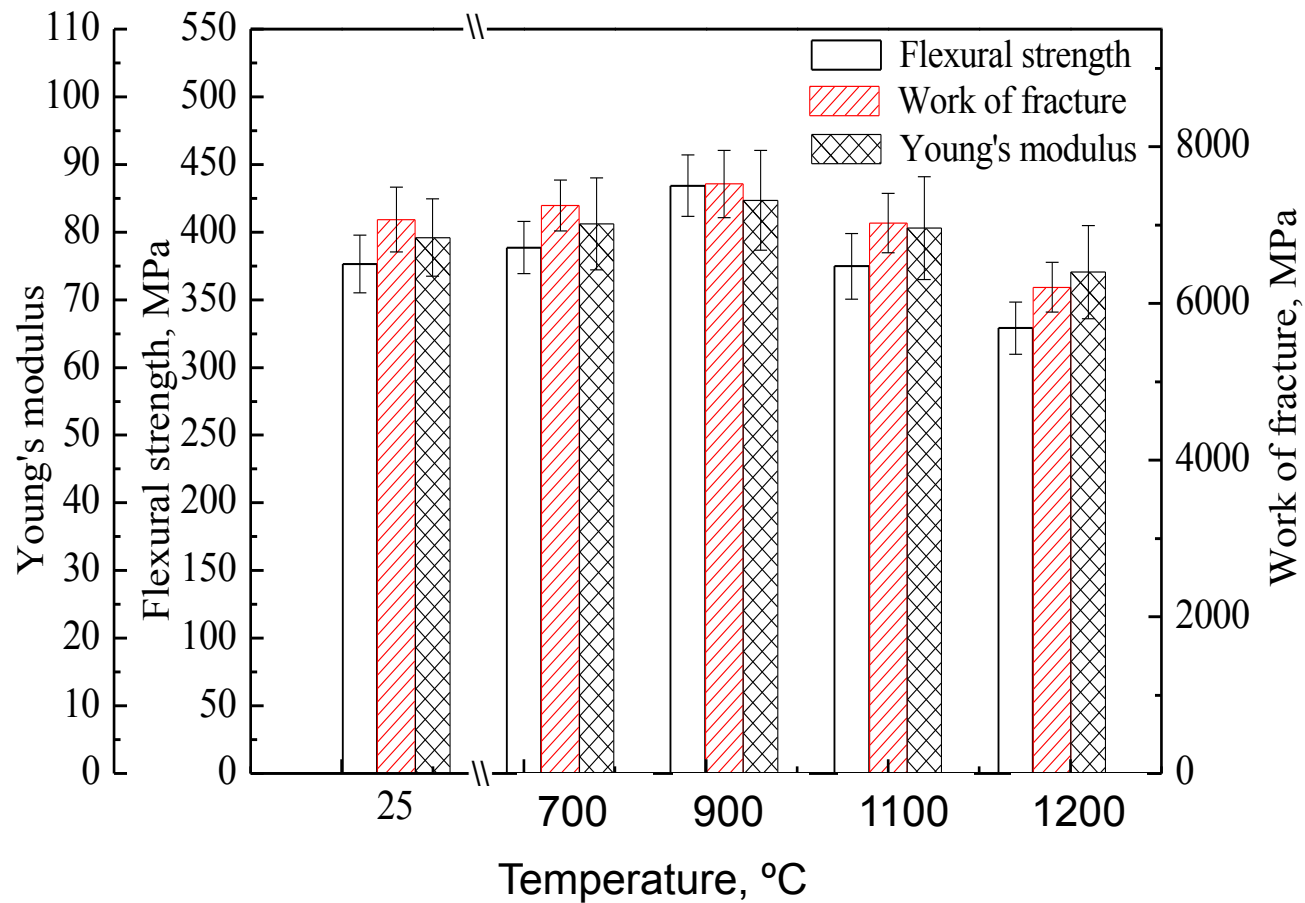


Fig. 4-13 Variation of the elevated temperature mechanical properties of ImHCsKC

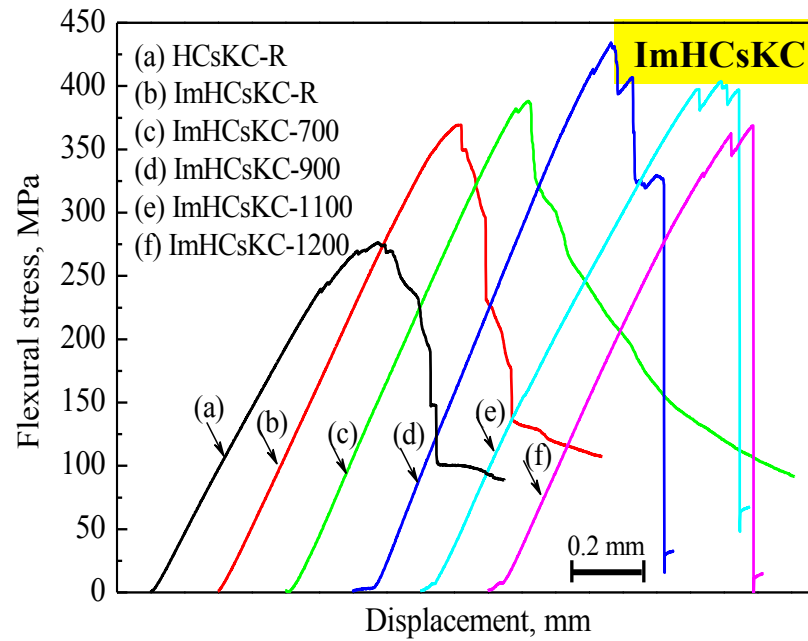
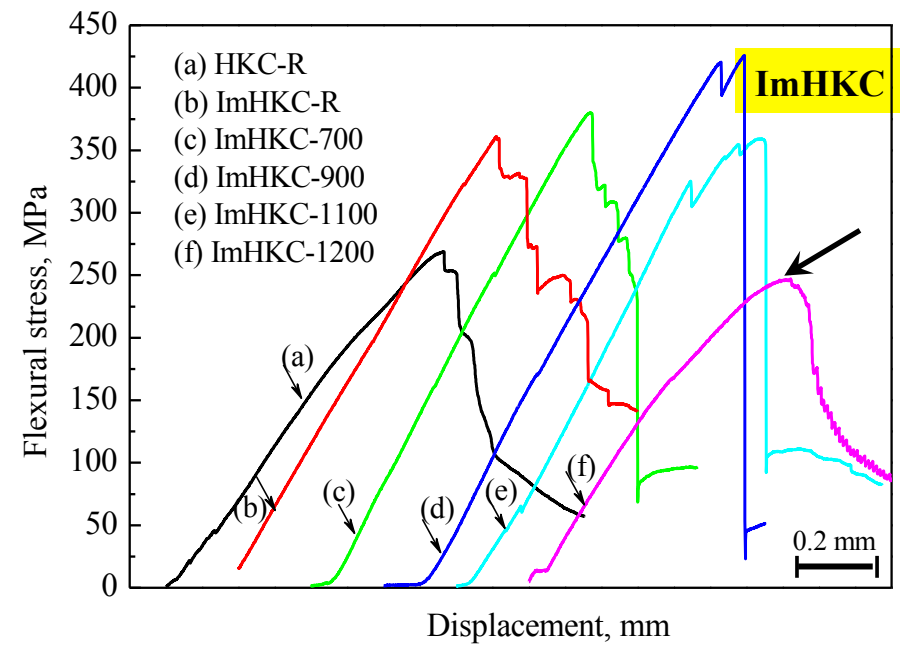
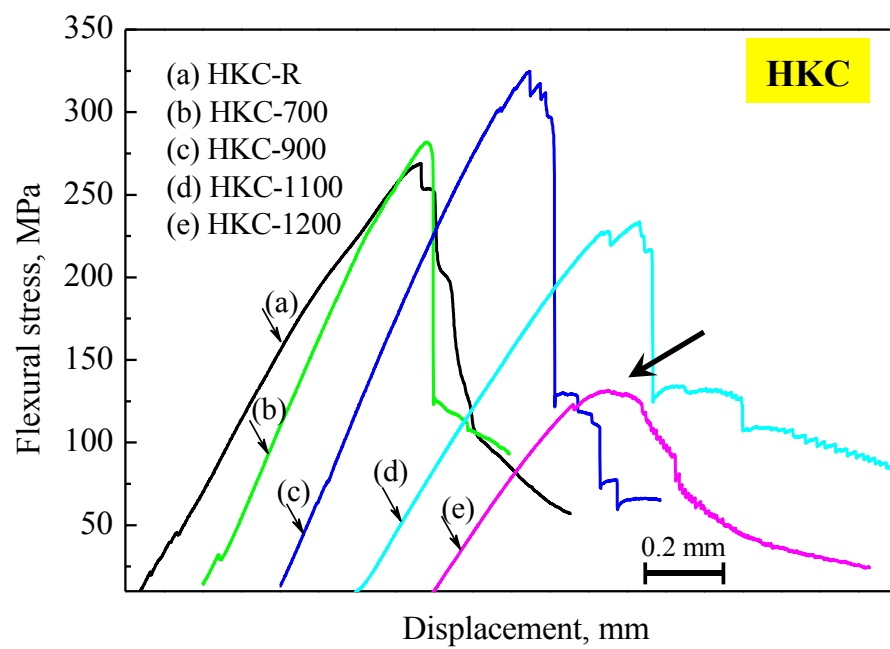


Fig.4-14 Flexural stress-displacement curves at different temperature

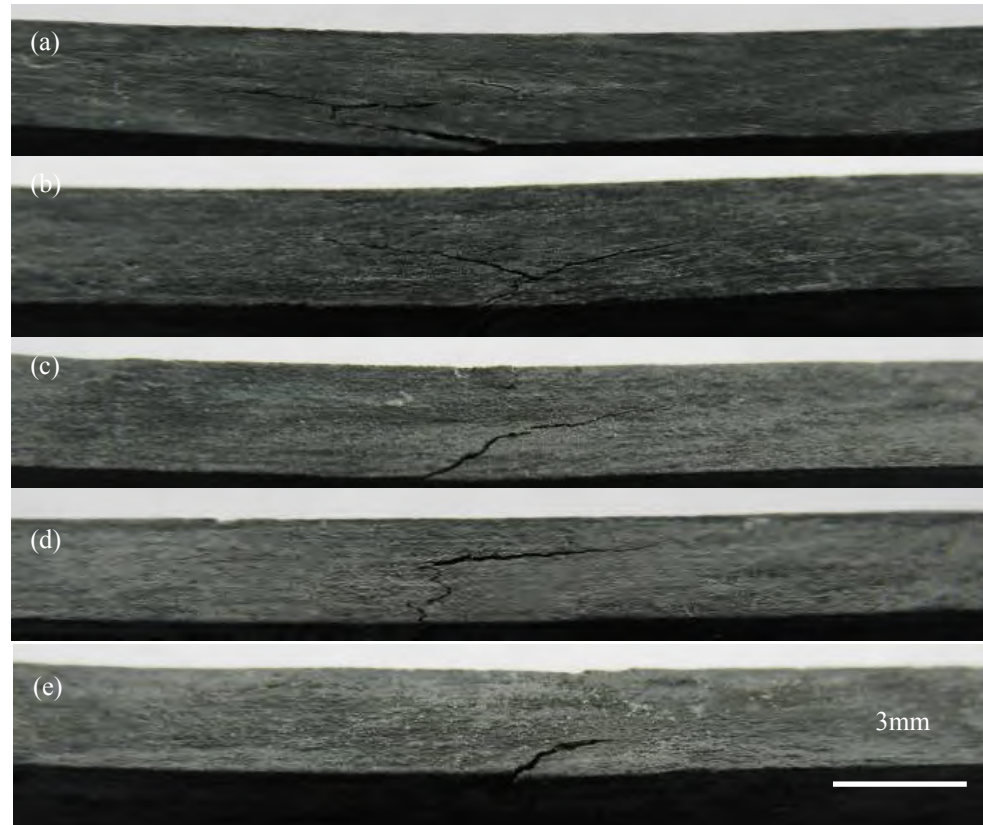


Fig. 4-15 Typical morphologies of ImHC specimens after bending tests at (a) RT, (b) 700°C, (c) 900°C, (d) 1100°C and (e) 1200°C

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.



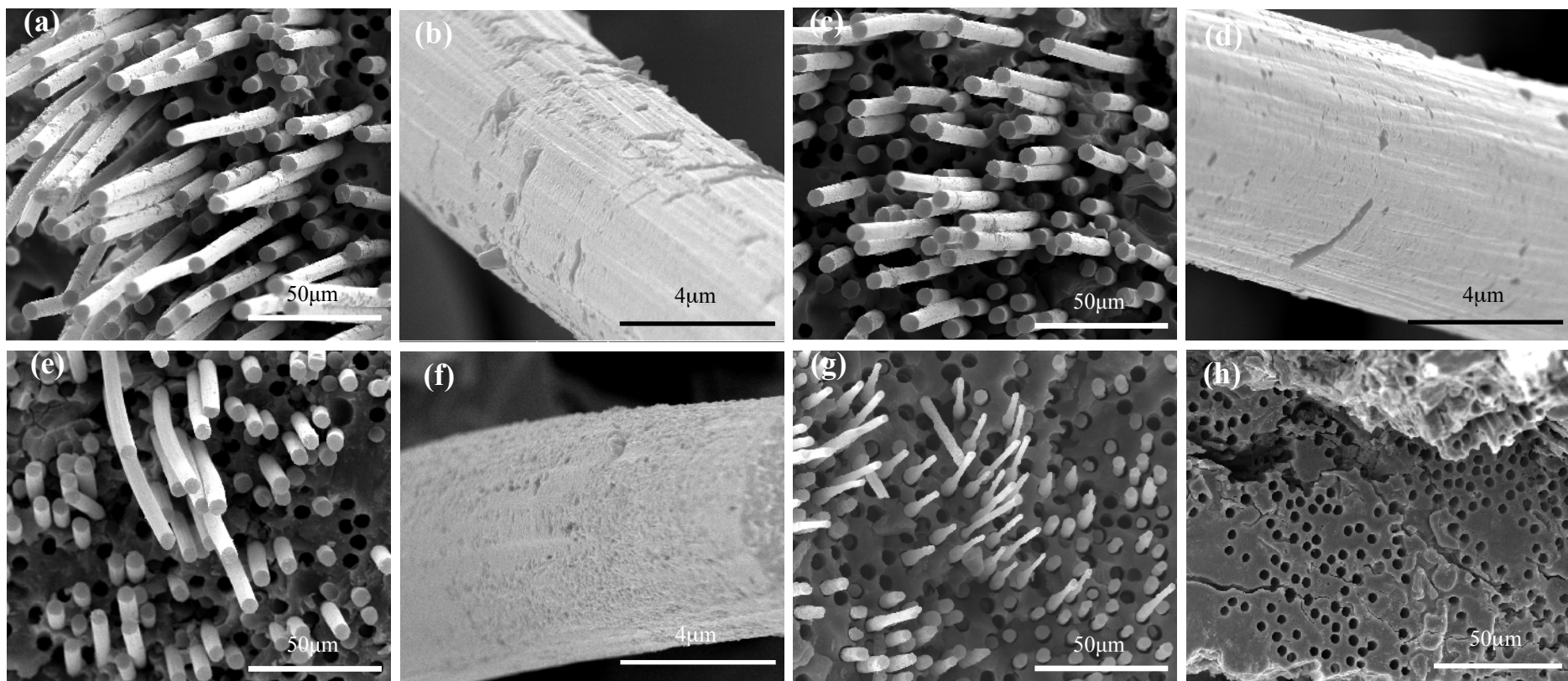


Fig. 4-16 Fractographs near the tensile sides of:
(a), (b) ImHKC-R; (c), (d) ImHKC-700; (e), (f) ImHKC-900; (g) ImHKC-1100; (h) ImHKC-1200

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.



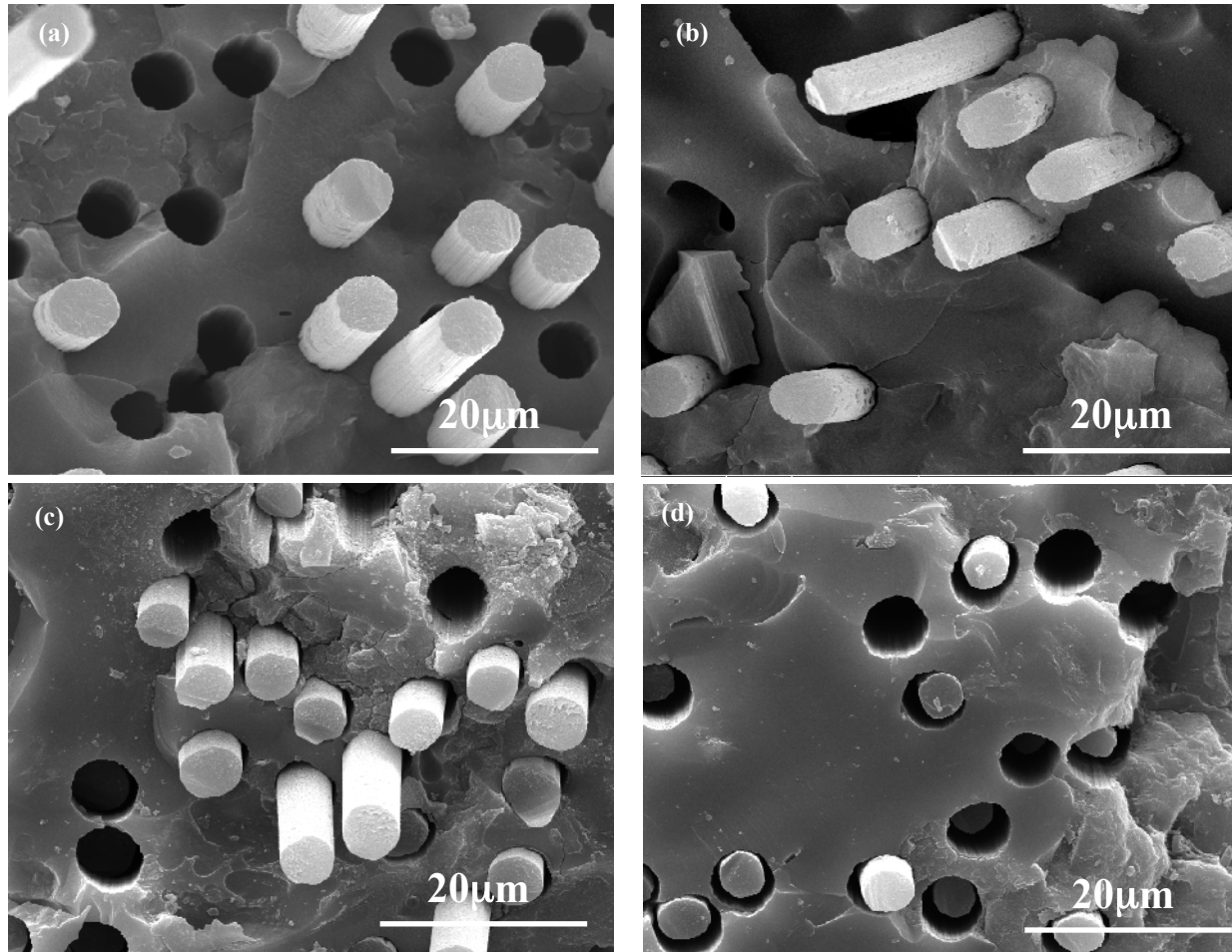


Fig. 4-17 Fractographs obtained at room temperature of ImHC after flexural strength test at: (a) RT, (b) 700°C, (c) 900°C, (d) 1100°C. Samples were prepared by broking-off the ImHC specimens in cryogenic liquid nitrogen and similar positions were select for observation in each sample

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.

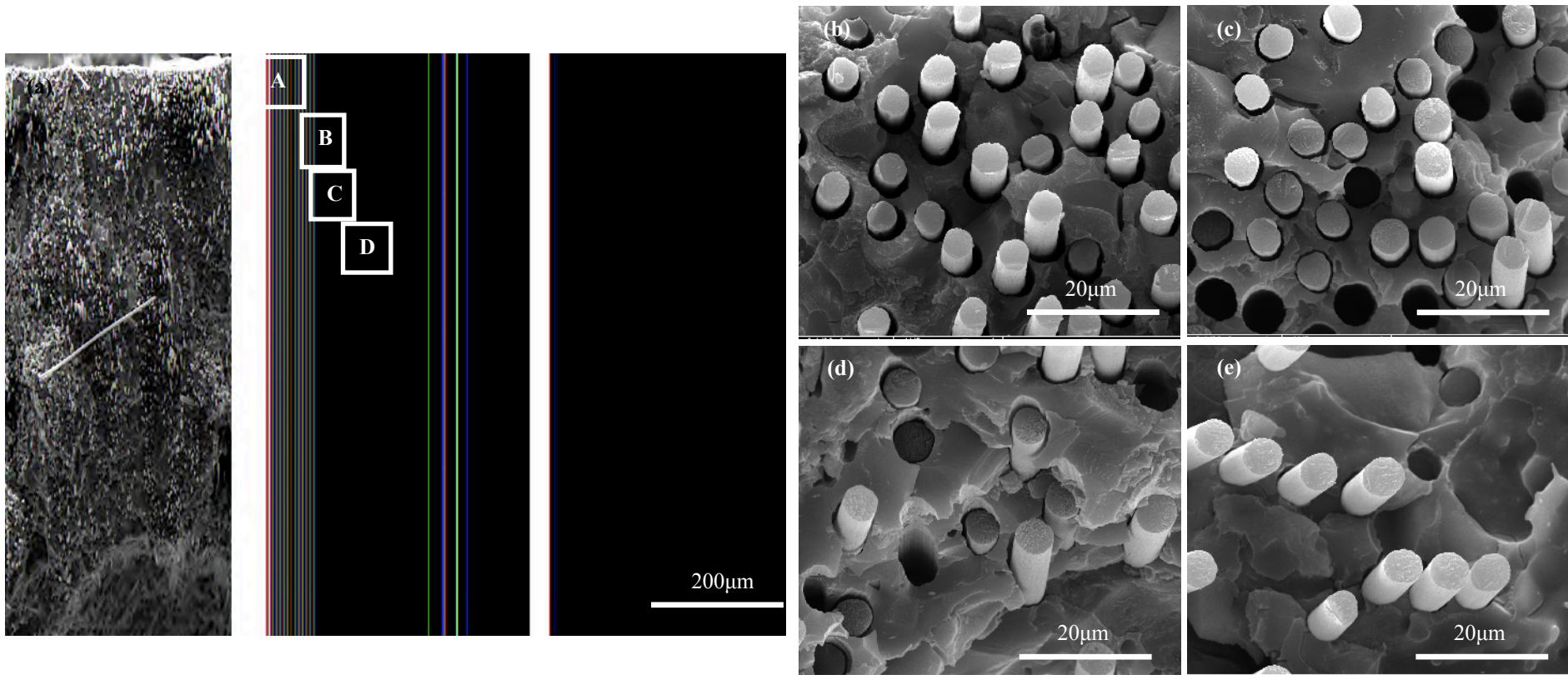


Fig. 4-18 Fractographs obtained at room temperature of ImHC after flexural strength test at 1100: (a) low magnification fractograph, (b)-(e) corresponding to the area of A-D in (a)

$$\text{Degree of Oxidation} = \left(1 - \frac{R_d}{R_0}\right) \times 100\%$$

R_d : Average radius of carbon fiber after test,
 R_0 : Initial radius of carbon fiber.

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.

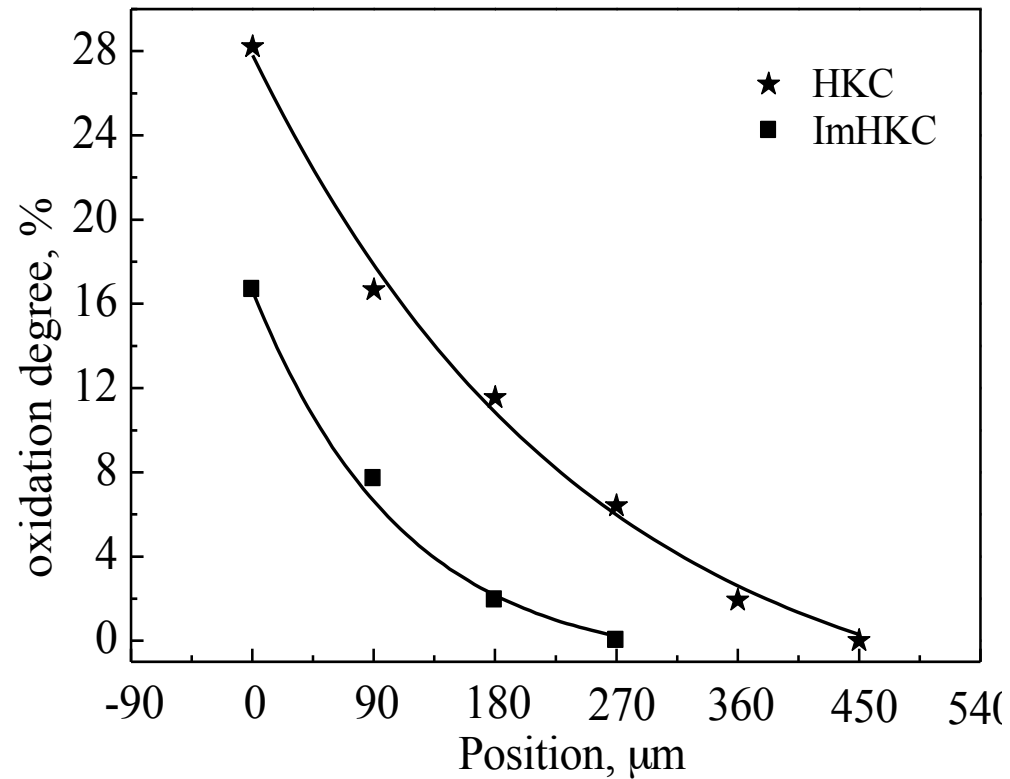


Fig. 4-19 The degree of fiber oxidation vs. distance from the surface of the two composites

---- Peigang He, Dechang Jia. *Journal of the European Ceramic Society*. 2010, 30 (15), 3053–3061.



4.4 Summary

- ◆ Repeated sol-SiO₂ impregnation was an efficient method to seal the cracks and pores in the heat-treated composites and thus greatly enhance their relative densities and ambient and high temperature mechanical properties.
- ◆ The impregnated composites possess the density of 2.15~2.2g·cm⁻³, and showed much high specific strength. Composites before & after impregnation fractured in a non-brittle manner at ambient & high temperatures.
- ◆ Over an elevated temperature range from 700°C to 900°C, the strength of the two composites showed anomalous gains and reached their maximum values at 900°C.
- ◆ The increased strength was attributed to the cracks blunting and release of residual strength based on the integrity of the carbon fiber.
- ◆ Compared with the composites without sol-SiO₂ impregnation, the impregnated one showed superior high temperature properties, which was attributed to the improved fiber integrity due to its much denser microstructure.



5. CONCLUSIONS

- ① Geopolymer can be converted into ceramic materials heat treated at proper high temperature range. The resulted ceramics possess adjustable coefficient of thermal expansion, which makes them excellent candidate matrix as structural materials.
- ② With proper heat treatment, carbon fiber reinforced geopolymer composites can be converted into C_f /Leucite composites, which showed excellent high temperature mechanical properties up to 1200°C.
- ③ Geopolymer technology together with cerium incorporation, high-temperature heat treatment and sol-SiO₂ impregnation etc provides a new method to fabricate a new composites of low cost and high specific strength, which is attractive for elevated temperature applications.



Thanks for your attention!

