

EPSRC DIAMOND Consortium Nuclear Waste and Decommissioning

An assessment of drying shrinkage in metakaolin-based geopolymers

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Problem: Cracking of Geopolymers

Investigation why cracking occurs

Conclusions



Problem: Cracking of Geopolymers



Geopolymers cured over a long time at room temperature, without cover



Experiments

metakaolin, Na₂SiO₃ + NaOH solution was mixed

1 day cured in sealed bags at RT and removed

further curing in sealed bags for 56 days

molar ratio was altered

- Al:Si:Na:H₂O 1:2:1:x x= 7.5 to 10.5
- AI:Si:Na:H₂O 1:x:1:8 x= 1.6 to 2.4
- AI:Si:Na:H₂O 1:2:x:8 x= 0.75 to 1.3
- AI:Si:x:H₂O 1:2:1:8 x= Na/K



Influence of H₂O on cracking

Dilatometer results





Influence of H₂O on cracking

Shrinkage measured using an extensometer





Influence of H₂O on cracking

Flexural strength: 3 point bending test





Influence of H₂O on cracking

Shrinkage of Geopolymers





Influence of Si on cracking

Onset point of shrinkage measured using extensometer





Influence of Si on cracking

Shrinkage during heating determined by dilatometry





Influence of Si on cracking



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Influence of Na on cracking

Onset point of shrinkage measured using extensometer





Influence of Na/K and Si ratio on cracking

Onset point of shrinkage measured using extensometer





Influence of Na/K and Si ratio on cracking

Shrinkage during heating determined by dilatometry



Influence of Na/K and Si ratio on cracking

Comparing Na and K

	Na⁺	K+
Radius, Å	0.97	1.33
charge density [Z/r]	1	0.75
$\Delta H^{\circ}_{hydn}/kJmol^{-1}$	-406	-322

Na has high charge density, means it remains hydrated during geopolymerisation

K has smaller hydration sphere compared to Na and water bond weaker



Conclusion





Results to date

Influence of water saturation after cation has lost hydration sphere





Coming soon:





Thank you for your attention

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