Elaboration and study of the functional properties of geopolymer foams

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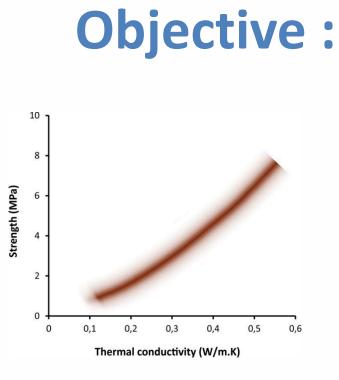
Geopolymer foams :

Foam concrete is a type of a lightweight concrete that consists of a cementitious binder with a high degree of void space.

Air incorporation enables to lighten the matter and to confer thermal and acoustic insulation properties.

Geopolymer foams can be used as acoustic and thermal insulators, filters or light weight fire proof insulators.





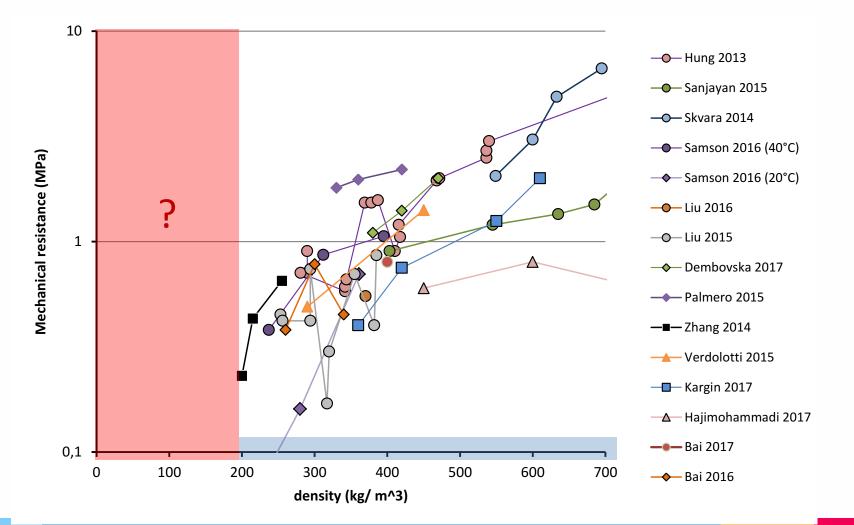
Example of an aerated Portland cement mortar Conciliate mechanical resistance of geopolymer foams and their acoustic and thermal insulation properties

→ Morphology optimization

Controlling the **nature**, **size** and **distribution** of voids is the most critical step in the production of these foams since the voids **determine the density and strength** of the foam concrete.

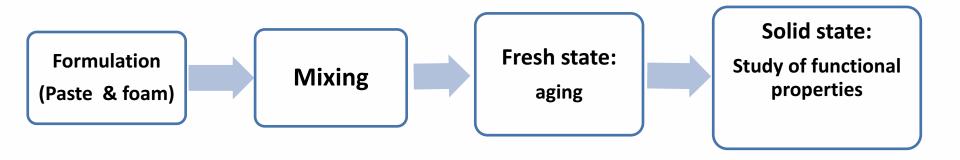
In the literature :

> The densities of Geopoymer foams made till now are over 200 Kg/m³



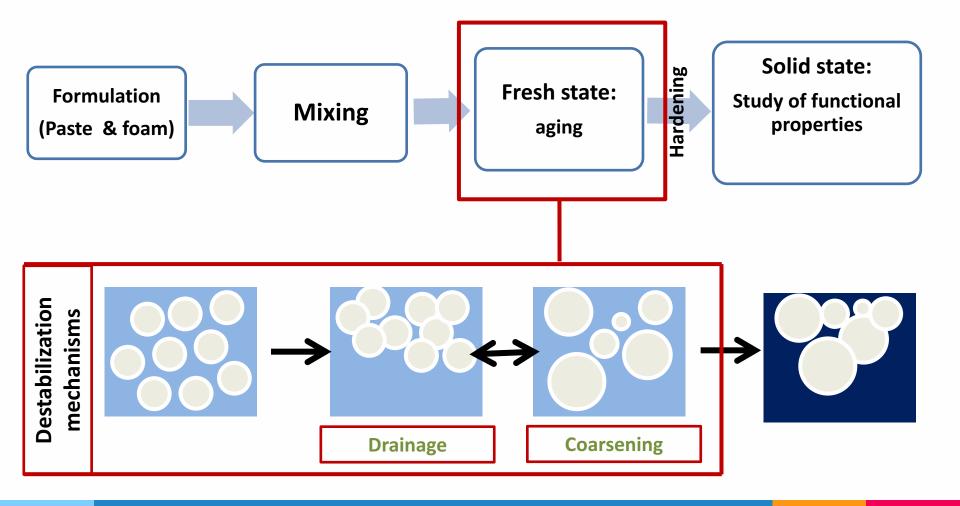


Major issue encountered during the optimization process: the morphology evolves strongly before setting



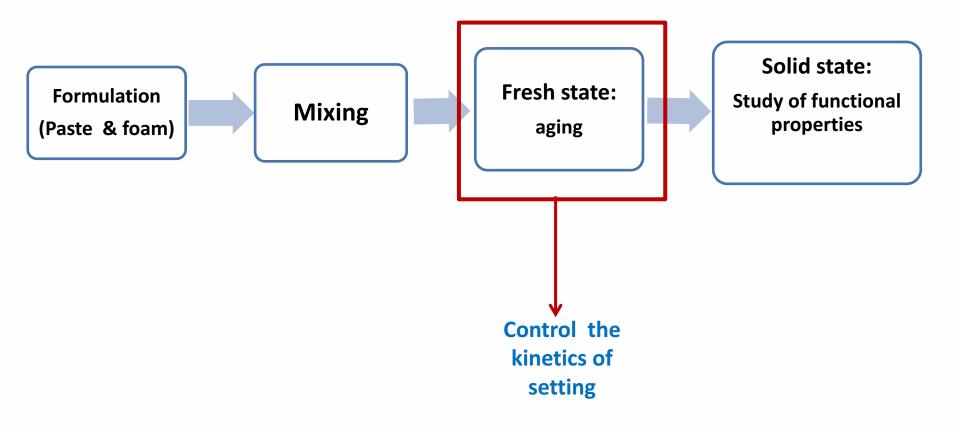
Issue:

Major issue encountered during the manufacture: the morphology evolves strongly before setting

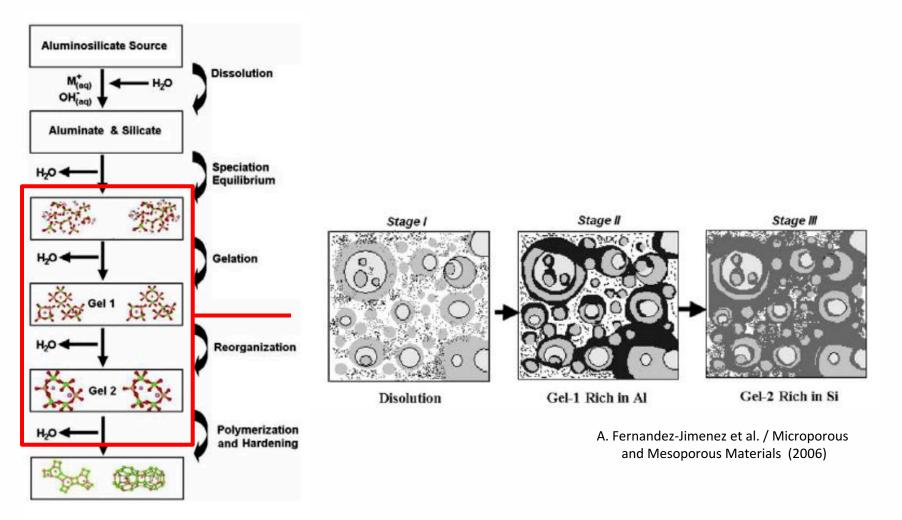


Issue:

Major problem incountred during the optimization process: the morphology evolves strongly before the setting



Kinetics of setting and rheology:



Duxson et al. (2006)

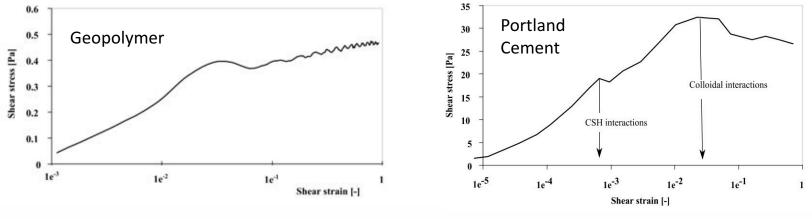
Kinetics of setting and rheology:

Rheological behavior of fresh geopolymer pastes :

• Geopolymer pastes can be assimilated to semi-dilute suspensions of non-colloidal particles in a Newtonian fluid of high viscosity.

(Roussel & al, 2014)

• Fresh geopolymer pastes have a very low yield stress.



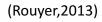
A.Favier et al. / Soft Matter (2013)

→ Understanding of the time evolution of rheology of geopolymer pastes is a major issue for the elaboration process

Materials & Methods:

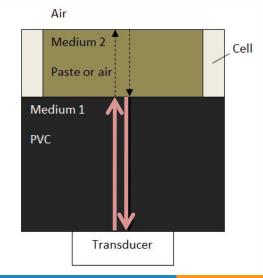
- Metakaolin (Argical M1200s) + Sodium Silicate solution
- - Ultrasonic rheology (f=100кнz)

$$G' = G = \frac{Z_1^2}{\rho} \frac{(1+R)^2}{[1-R]^2}$$



R: Reflection coefficient

1 Z₁: acoustic impedance of the medium 1



Methods:

• Proton NMR (H¹NMR) :

- Monitoring of relaxation times T_1 and T_2 which characterize respectively the times of return to the equilibrium of the longitudinal and transversal components of magnetic moments,

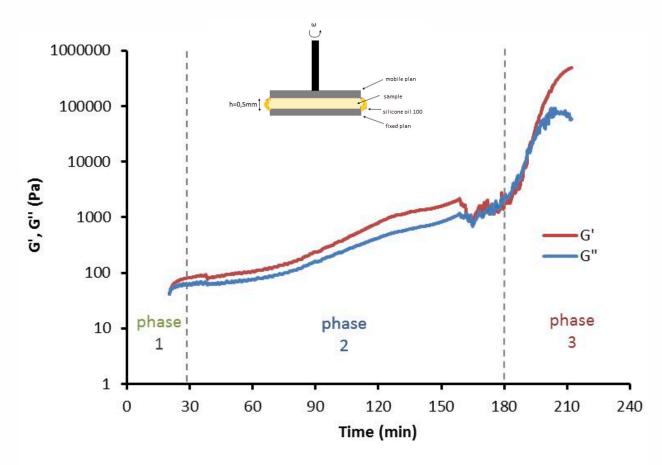
T_{1,2} ~ V/S, already used to characterize the setting of Portland cement.
(P. Faure et al. / Magnetic Resonance Imaging (2008))

 \rightarrow Application of Proton NMR to geopolymer pastes.



Bruker Minispec MQ20

Monitoring of the elastic modulus using standard rheology

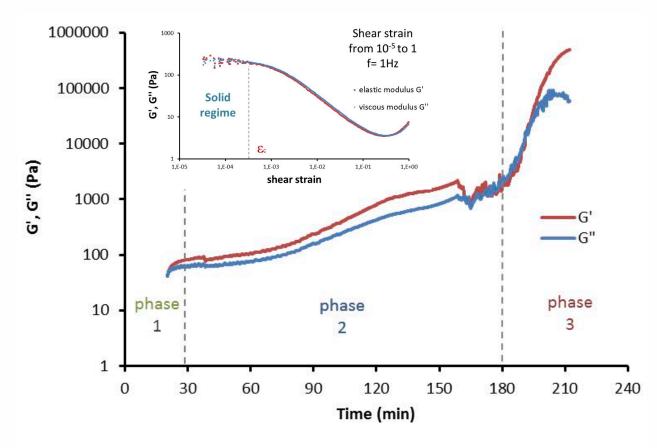


Phase 1 : formation of an Al-rich gel around Metakaolin grains (gel1)

Phase 2 : induction phase

Phase 3 : formation of a gel of a higher connectivity (gel2) → setting

Monitoring of the elastic modulus using standard rheology

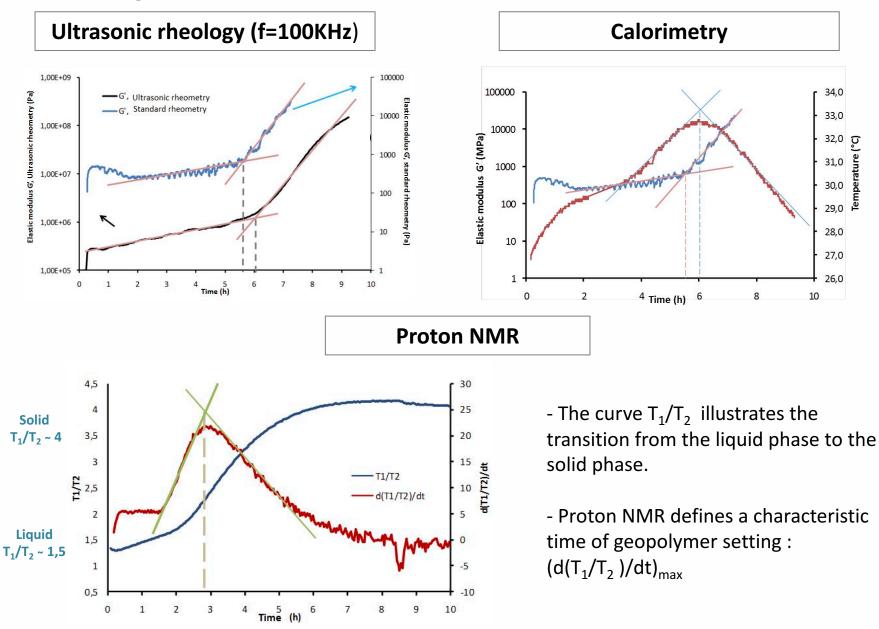


Phase 1 : formation of an Al-rich gel around Metakaolin grains (gel1)

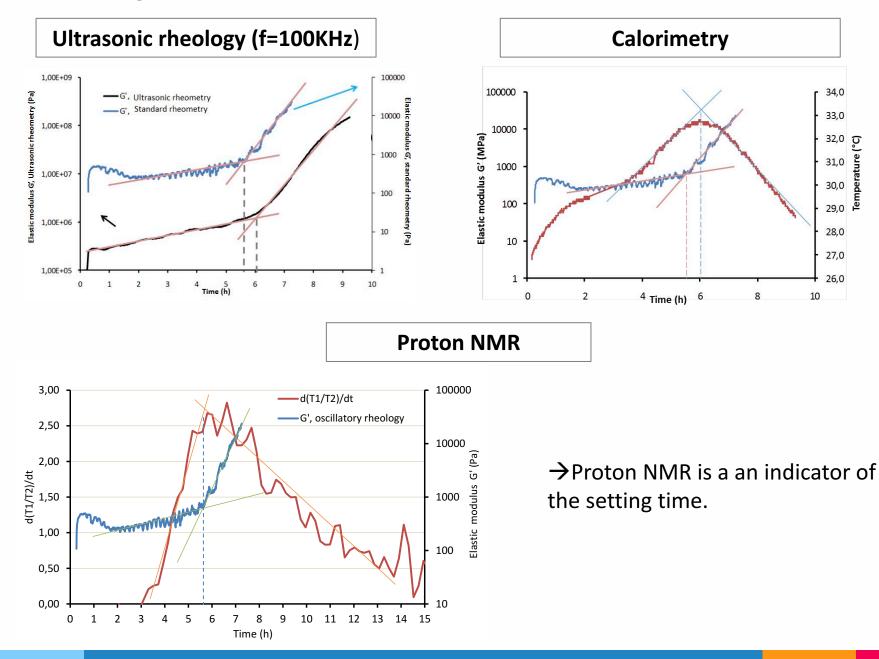
Phase 2 : induction phase

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Monitoring of the elastic modulus with:

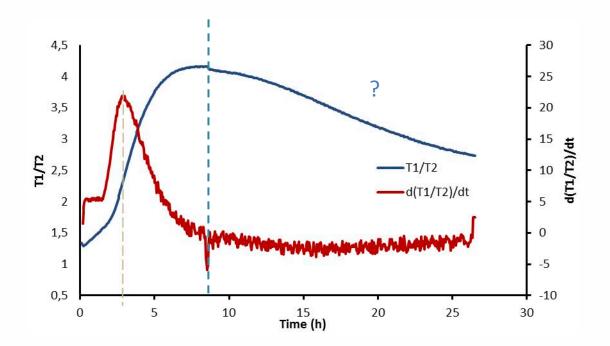


Monitoring of the elastic modulus with:



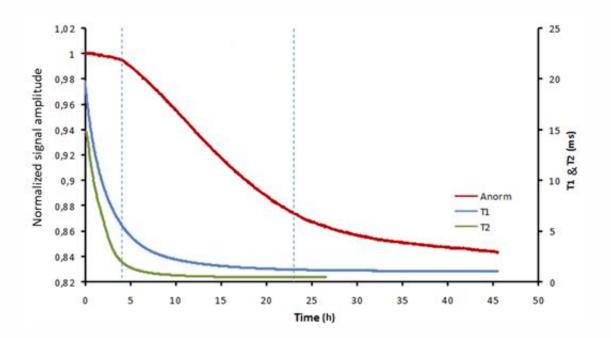
Proton NMR

• Allows to monitor the setting for long periods of time

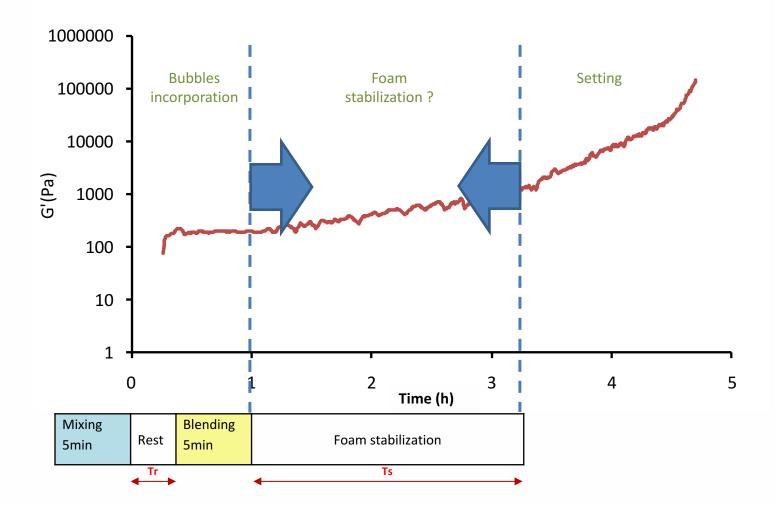


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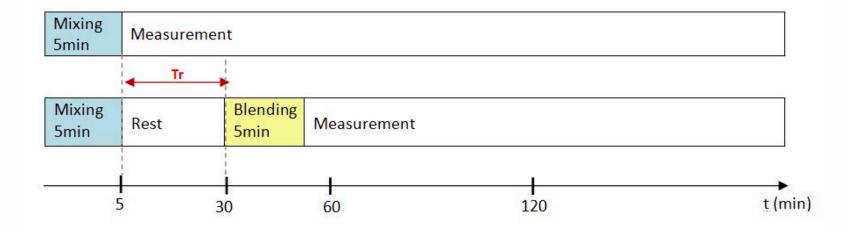


Strategy for bubbles stabilization

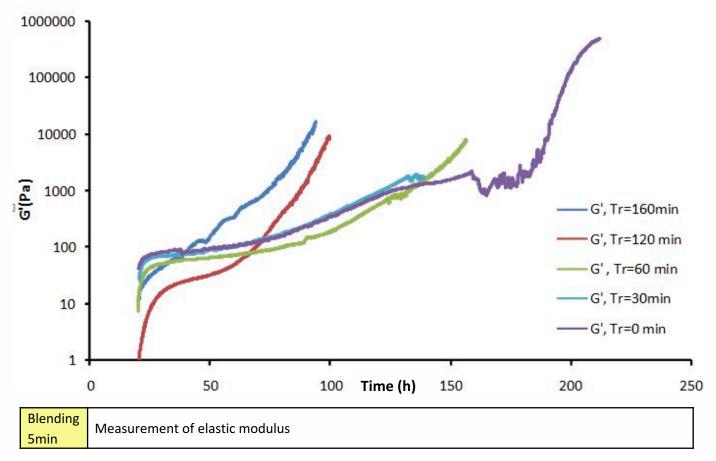


Influence of air incorporation on setting time

Sequence for the measurement of the effect of mixing on the time of setting Second mixing simulates foam incorporation

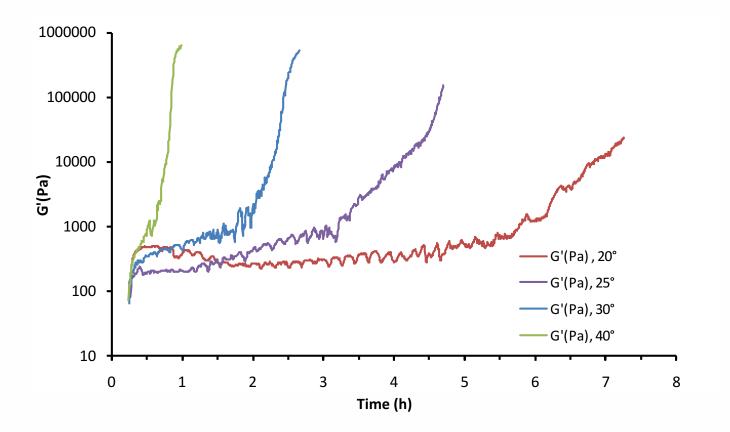


Influence of air incorporation on setting time



- Gel 1 reforms systematically after each blending in the same way.
- Time of induction is as much reduced as the time of rest before mixing is important

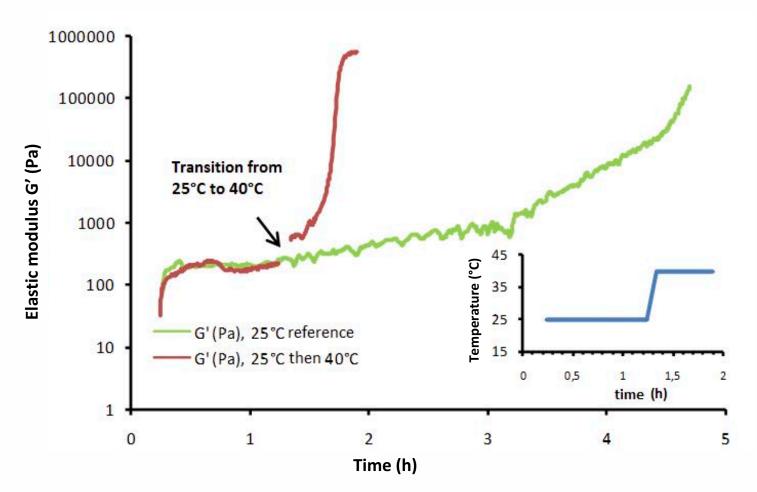
Influence of temperature on the time of setting



The difference between the times t (10000Pa) at 20°C and 25°C is 2h43min

Effect of a temperature step on the time of setting

Unleash the setting with a temperature step from 25°C to 40°C \rightarrow Gain in time of foam stabilization



Preliminary work for geopolymer foaming:

Tween 80 (Sigma-Aldrich) (M. Strozi Cilla et al. / Ceramics International 40 (2014))





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Thank you !