

INSTITUT

**GÉOPOLYMÈRE**

***Prof. Dr. Joseph Davidovits***

[www.geopolymer.org](http://www.geopolymer.org)



# Geopolymer in Wikipedia



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The name Geopolymer was first applied to these materials by [Joseph Davidovits](#)<sup>[1]</sup> in the 1970s, although similar materials had been developed in the former [Soviet Union](#) since the 1950s, originally under the name "soil cements".<sup>[2][3]</sup> However, this name never found widespread usage in the English language, as it is more often [applied](#) to the description of soils which are consolidated with a small amount of [Portland cement](#) to enhance strength and stability.

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Geopolymers are an example of the broader class of [alkali-activated binders](#), which also includes alkali-activated [metallurgical slags](#) and other related materials.<sup>[4]</sup>



# Research



## Research

Much of the drive behind research carried out in academic institutions involves the development of geopolymers as a potential large-scale replacement for [concrete](#) produced from Portland cement. This is due to geopolymers' alleged lower [carbon dioxide](#) production emissions, greater chemical and thermal resistance and better mechanical properties at both ambient and extreme conditions.

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There is some debate<sup>[[by whom?](#)]</sup> as to whether geopolymer cement has lower CO<sub>2</sub> emissions compared to Portland cement. Calcination of limestone in production of Portland cement is responsible for CO<sub>2</sub> emissions (one ton of cement produced releases one ton of CO<sub>2</sub>), while some processes of formation of lime also release CO<sub>2</sub>. Mainly it is the ratio of CO<sub>2</sub> reduction that is under debate, and it is process-dependent.



# Production

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Geopolymers are generally formed by reaction of an aluminosilicate powder with an alkaline silicate solution at roughly ambient conditions.

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Industrial and high-tech applications rely on more expensive and sophisticated siliceous raw materials.



# Theory

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The majority of the Earth's crust is made up of Si-Al compounds. Davidovits proposed in 1978 that a single [aluminium](#) and [silicon](#)-containing compound, most likely geological in origin, could react in a [polymerization](#) process with an alkaline solution. The binders created were termed "geopolymers"

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Some have *alleged* that ancient "Roman cement" is a geopolymer, but in reality this material is chemically unlike alkali activated geopolymers because it is made using lime and forms calcium-silicate-hydrates, making it much closer to Portland cement from a chemical standpoint.





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The inorganic polymer network is in general a highly-coordinated 3-dimensional aluminosilicate [gel](#), with the negative charges on tetrahedral Al(III) sites charge-balanced by [alkali metal cations](#).



# History



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However, because Roman cement forms calcium-silicate-hydrates, and requires calcined limestone as a reactant/precursor, it is more similar to Portland cement than alkali-activated "geopolymer cements" such as Pyrament cement of LoneStar.[\[6\]](#)[\[7\]](#)

## References

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- <sup>2</sup>. <sup>^</sup> [Stabilization/solidification of hazardous and radioactive wastes with alkali-activated cements](#) Science Direct *Journal of Hazardous Materials* 2005-08-13
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- <sup>4</sup>. <sup>^</sup> Shi, Caijun; Krivenko, Pavel V.; Roy, Della M. (2006). *Alkali-Activated Cements and Concretes*. Abingdon, UK: Taylor & Francis.
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- <sup>6</sup>. <sup>^</sup> [Davidovits, Joseph](#); Morris, Margie (1988). *The pyramids: an enigma solved*. New York: Hippocrene Books. ISBN 0-87052-559-X.
- <sup>7</sup>. <sup>^</sup> [Davidovits, Joseph](#); Aliaga, Francisco (1981). "[Fabrication of stone objects, by geopolymeric synthesis, in the pre-Incan Huanka civilization \(Peru\)](#)". *Making Cements with Plant Extracts*. Geopolymer Institute. Retrieved 2008-01-09.



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MK-750 alumoxyl group:  $\text{Si-O-Al=O}$

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alkaline medium (Na, K, Ca): poly(silicates), poly(silico-aluminates)

acidic medium (Phosphoric acid): poly(phospho-siloxo), poly(alumino-phospho)



MK-750 alumoxyl group:  $\text{Si-O-Al=O}$

reacts in both systems.