Granulation and coating with Geopolymer Binders

Henk Nugteren

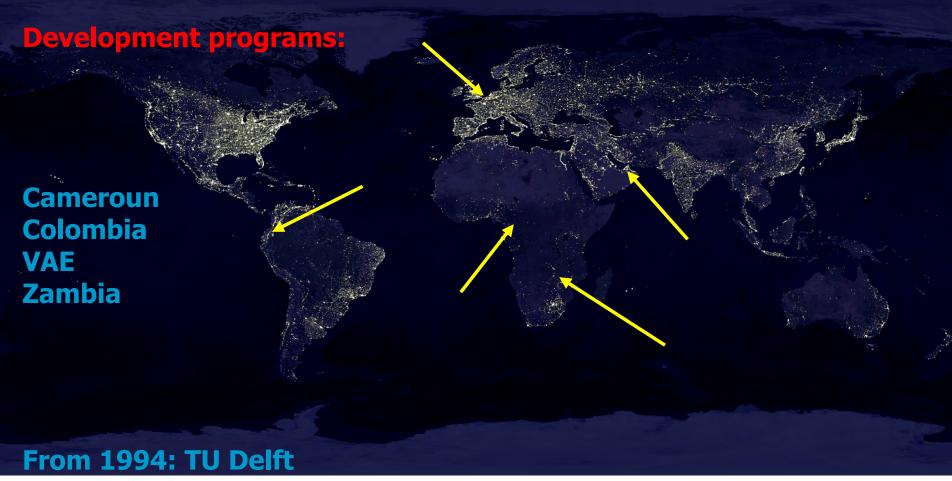
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Henk Nugteren

Study Geology, Mineralogy, Ore Deposits at VU Amsterdam 1978



Research: New Products from waste materials (PhD in 2010)



GRANULATION

GOAL

















Or is it an art?

Liquid content is crucial

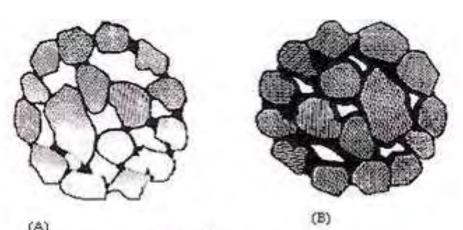




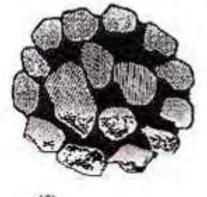
STAGES OF WETTING

A) Pendular state

B) Funicular state



C) Capillary state



Growth regimes:

- Steady Growth
- Induction Growth



High shear granulator

(Eirich R-02)







Raw Materials and Activator Liquids

SOLIDS:

- Type C Pulverised Fuel Ash
- Peat and Wood Ash (Netherlands and Finland)
- Granulated Blast Furnace Slag
- Polluted Sand
- Metakoalin

LIQUIDS:

- Water
- Potassium Silicate Solution
- Sodium Aluminate Solution (Waste from Aluminium Etching)



THE LIQUID CHALLENGE

$$[4SiO_2 \cdot Al_2O_3] + 10 OH^- + 3 H_2O \longrightarrow 2 [Al(OH)_4]^- + 4[SiO_2(OH)_2]^{2-}$$

$$[Al(OH)_{4}]^{-} + [SiO_{2}(OH)_{2}]^{2-} \xrightarrow{-H_{2}O} \begin{bmatrix} O^{-} & HO \\ | & | \\ HO - Si & -O - Al^{-} - OH \\ | & | \\ O^{-} & HO \end{bmatrix} \xrightarrow{}$$

$$\xrightarrow{polycondensation} \begin{bmatrix} \begin{vmatrix} & & & & & & \\ -Si-O & -Al^{-}-O & -Si-O & - \\ & & & & & \\ 0 & O & O & \\ & & & & & \end{bmatrix}_{n}$$

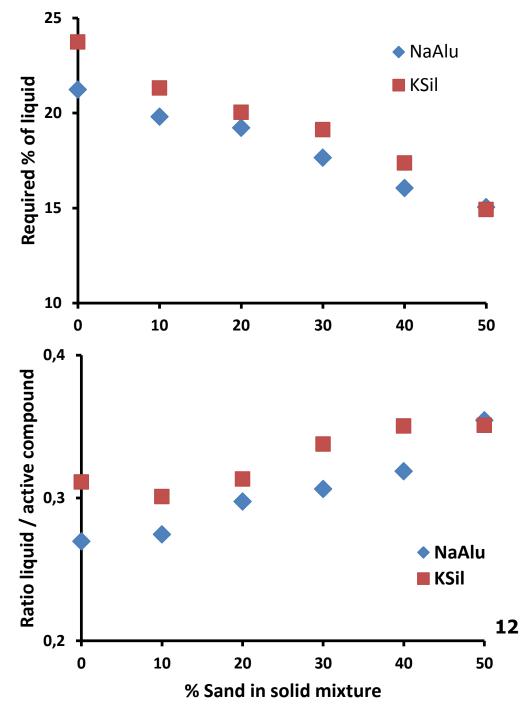
ゟ **TU**Delft

LIQUID REQUIRED

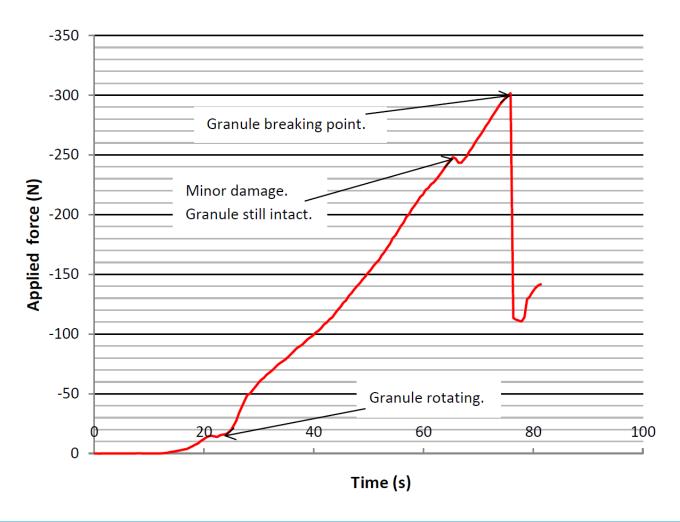
More sand:

- less liquid in mix
- higher liquid to precursor ratio

Geopolymer Camp, Sain



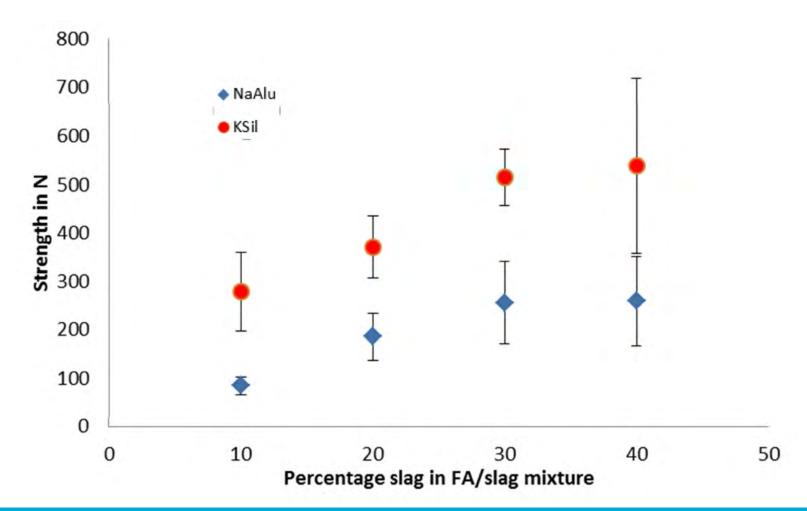
STRENGTH MEASUREMENTS





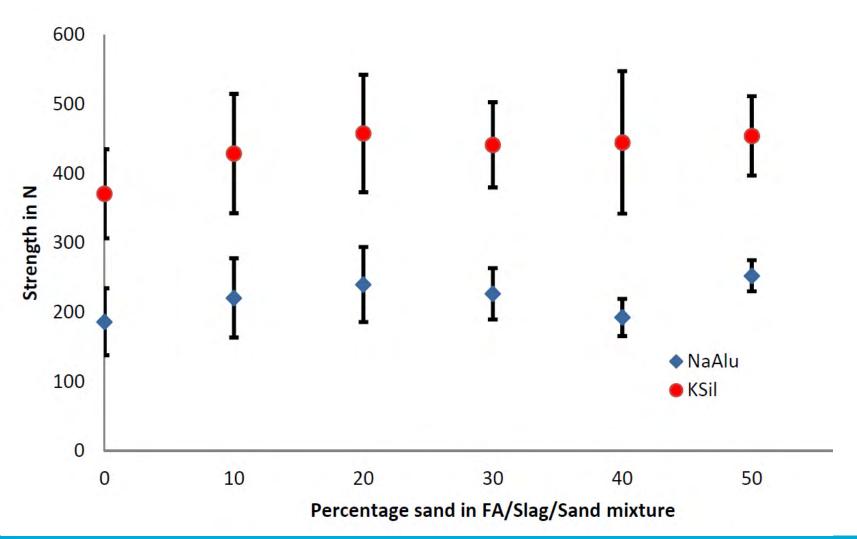


STRENGTH OF GRANULES



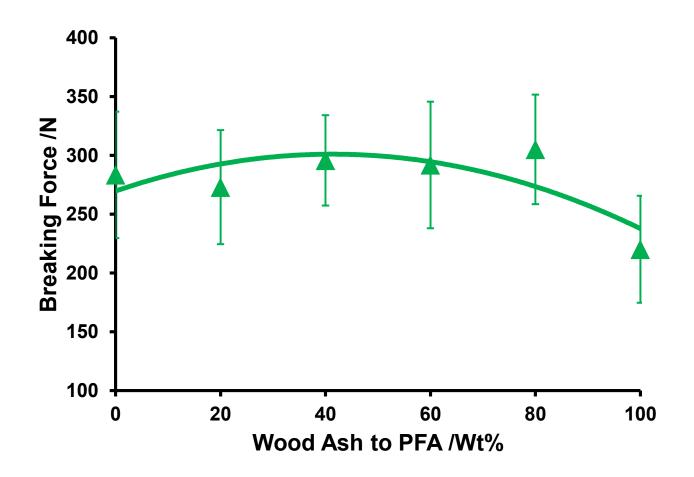


STRENGTH OF GRANULES





WOOD ASH PERFORMANCE

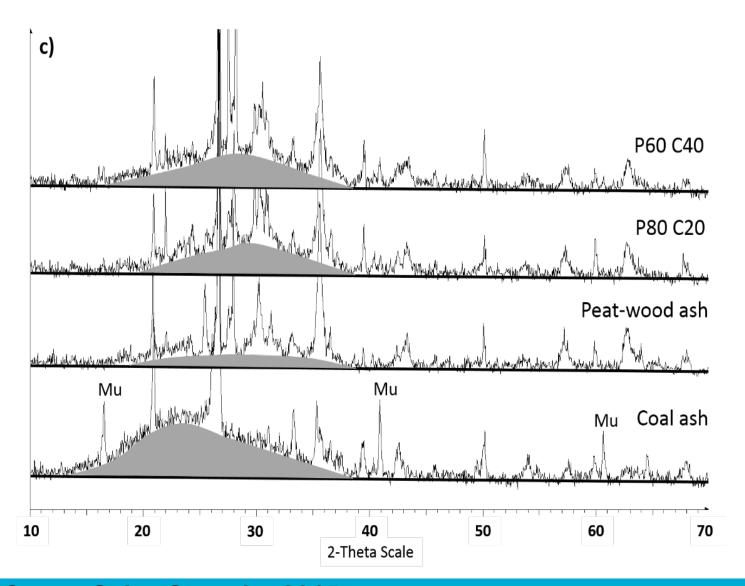




But:

Do we really have chemical reaction?

XRD



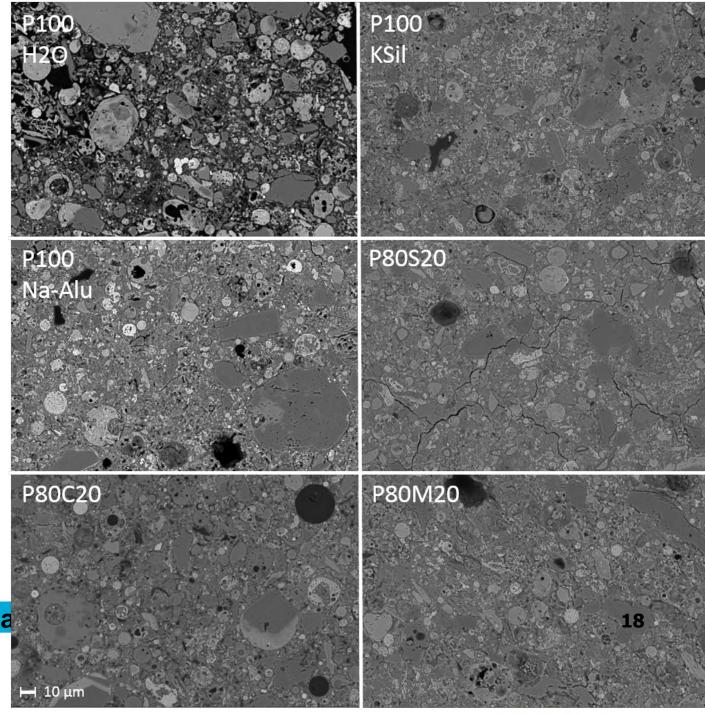


But:

Do we really have chemical reaction?

FESEM BSE

Geopolymer Ca



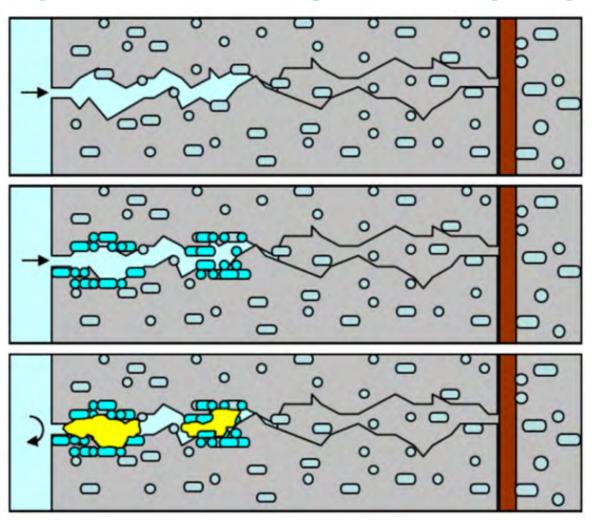
CONCLUSIONS

- Granulation of geopolymer precursors with spraying liquid activators is feasible
- Peat / wood ash with relatively small amounts of silica and alumina can still produce strong granules
- Operating windows (liquid / solid ratio) are narrow but can be maintained even when the geopolymeric reaction takes place simultaneously
- High amounts of polluted inerts can be added without losing strength
- XRD and FESEM show that geopolymeric binders have formed



COATING OF SELFHEALING PARTICLES

PRINCIPLE
OF
SELFHEALING

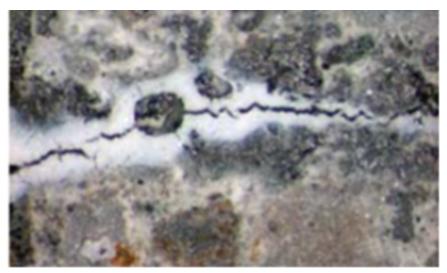




Concrete without healing agent



Just cracked



After healing period

Concrete with healing agent



Just cracked



After healing period

Low Shear Pan Granulator



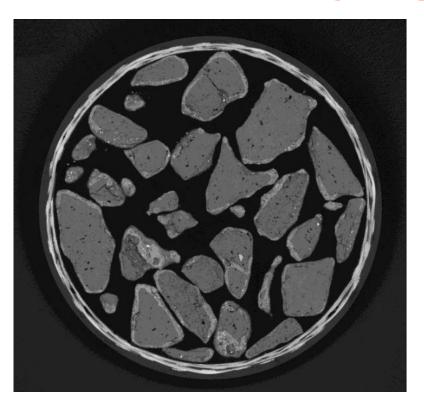


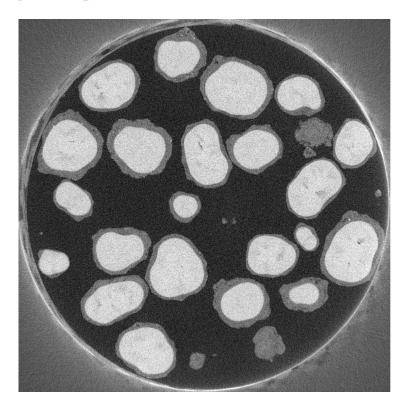


Geopolymer Camp, Saint-Quentin, 2015



Shape of particles to be coated CT Scans

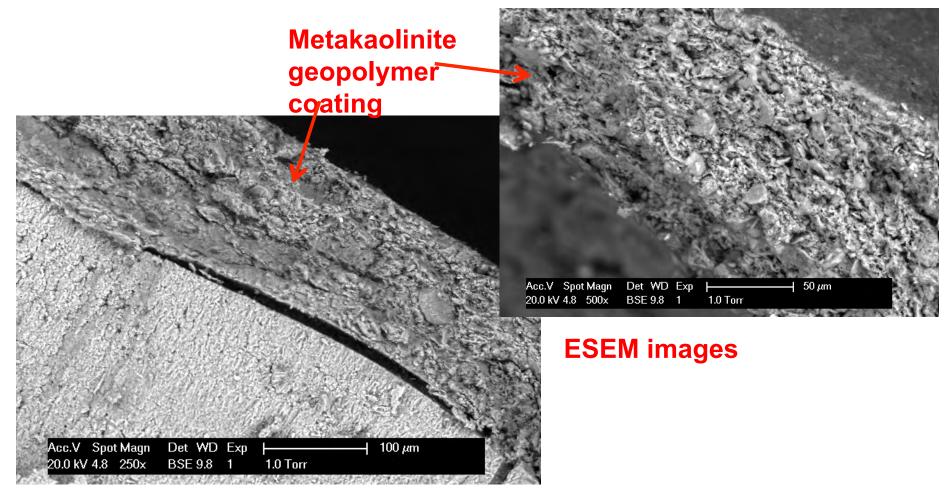








Coated particles

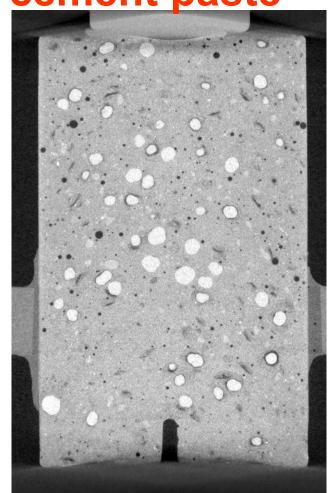




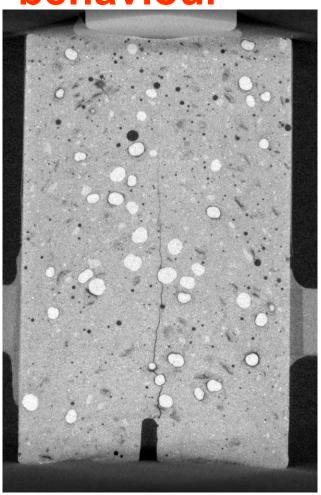
Water tightness

Looching colution	Comple	Ca	Al	Si	
Leaching solution	Sample	(mg/l)	(mg/l)	(mg/l)	
6 M ЦС	Coated core	28.2	15.7	2.4	
6 M HCl	Coating only	3.4	31.3	2	
1 M N-011	Coated core	1.6	85.5	629	
1 M NaOH	Coating only	1.9	248	1123	

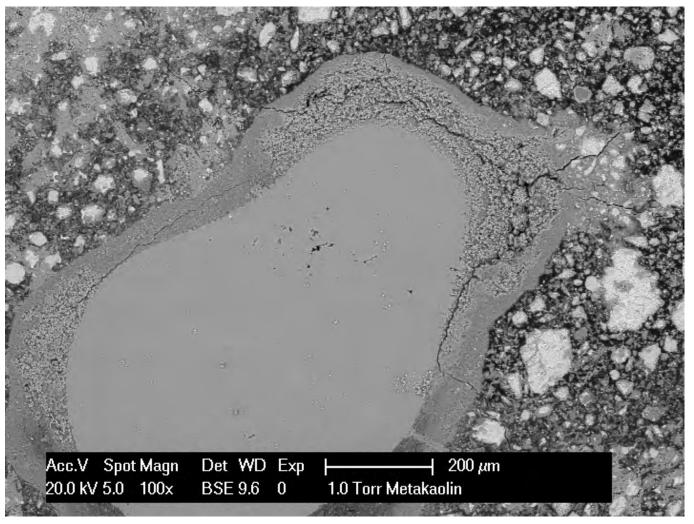
7 TUDelft Adhesion to cement paste



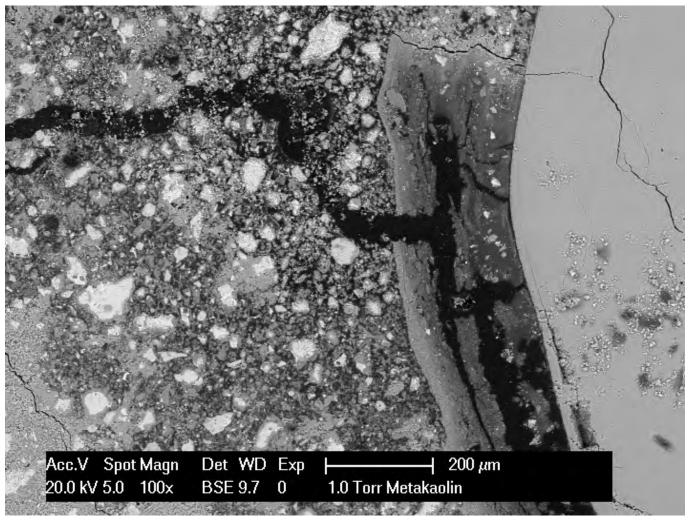
Cracking behaviour



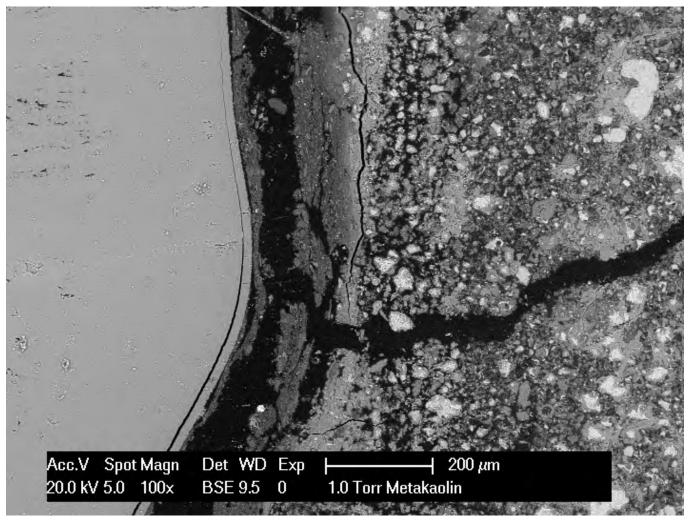




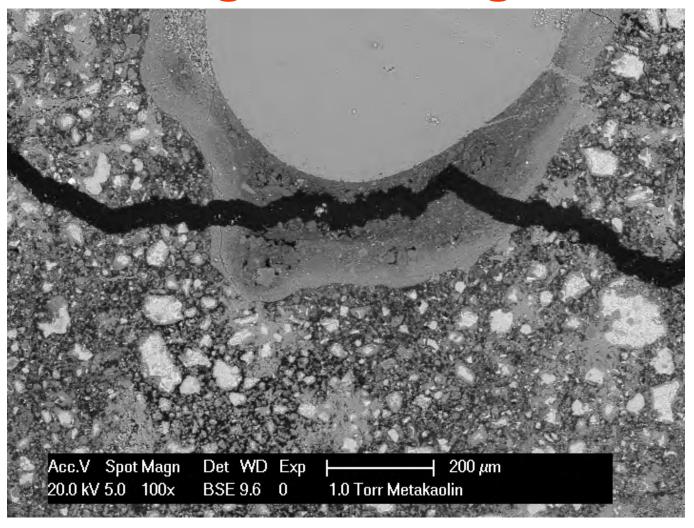












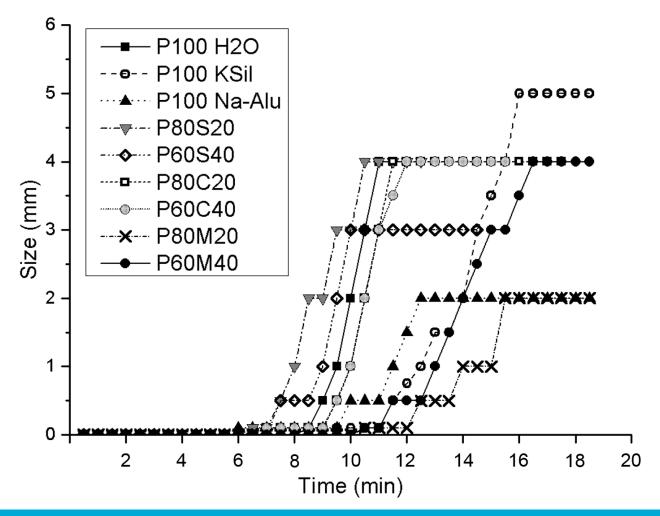


CONCLUSIONS

- Geopolymer coatings were obtained by carefully dosing metakaolinite powder and spraying liquid activators
- Coatings are uniform and water tight
- Adhesion to cement paste is excellent
- Cracks developed in the cement paste follow paths connecting the embedded particles
- Coatings will also crack and make selfhealing agents available when required



Induction Growth







Composition of solid precursors

	Chemical composition, %								Particle size				
Sample	CaO	SiO ₂	Al_2O_3	Fe ₂ O ₃	Na ₂ O	K ₂ O	MgO	P ₂ O ₅	TiO ₂	SO ₃	<10% [µm]	<50% [µm]	<90% [µm]
Р	11.6	44.9	10.7	20.1	1.5	2.1	3.0	2.6	0.4	2.0	2.4	17.4	135.6
S	39.5	34.5	9.9	0.5	0.4	0.3	8.1	0.0	1.1	3.4	1.0	10.3	31.9
С	4.9	54.3	22.9	8.0	1.1	1.7	1.8	0.7	1.2	8.0	3.0	30.2	133.7
M	0.1	59.5	32.8	1.4	0.1	0.6	0.1	0.0	1.9	0.0	1.0	8.0	53.1

∦ TUDelft

SODIUM ALUMINATE ACTIVATOR

Waste rinsing bath from aluminium etching:

Al 69-85 g.kg⁻¹
NaOH (free) 17-30 g.kg⁻¹
Si 0.3 g.kg⁻¹
S 2 g.kg⁻¹
Cr 4 mg.kg⁻¹
Others <1 mg.kg⁻¹

Black particles are NaAlO₂ with:

Minor quantities: Mg, Fe, Si, Ca and S (0.5-5%)

Mn, Zn and Cu (\pm 2000 ppm)

Traces: Cr, Pb, Ti, V and Ni (<350 ppm)



ACKNOWLEDGEMENTS

MSc and BSc students
Yvar de Groot
Jan-Jaap Hofman
Rick Weststrate
Juho Yliniemi
Stephan de Koster
Steven Lawant

Arno Keulen, Henk Jonkers, Renée Mors and Gabrie Meesters

