

Introducing the huge potential of slate powder as a geopolymer raw material.

GeopolymerCamp 2025

TUESDAY, July 8

– Richard Wyn Huws, Penygroes, Gwynedd, UK

Llechi Slate Evolution Cyf



Slate Powder: A Circular Economy Opportunity

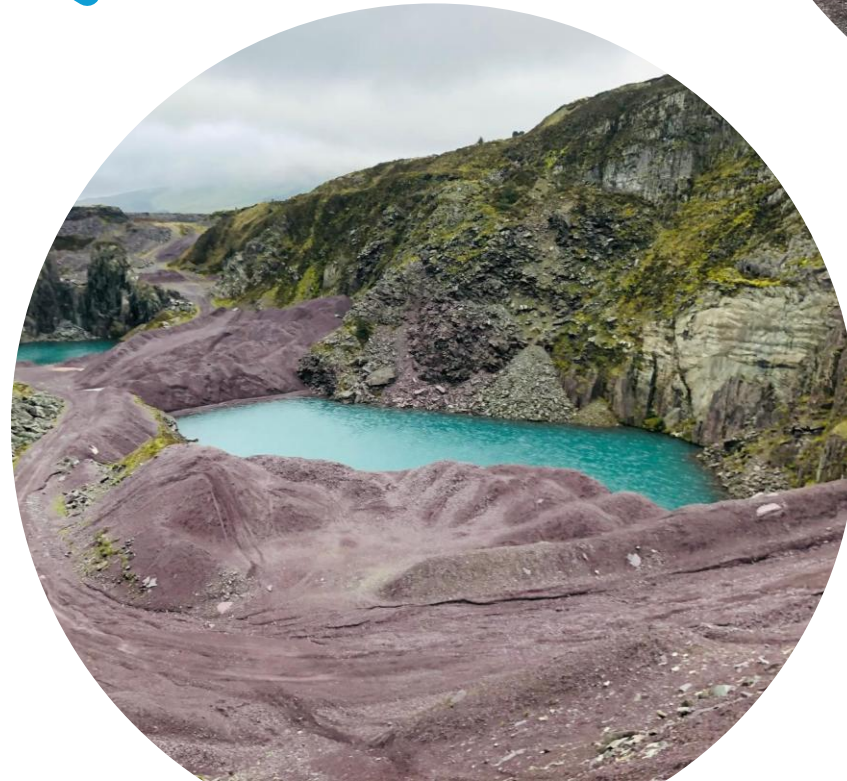
- Just to show the material in situ and the vast/millions of tonnes of it as waste and the circular economy.
- The powder is WASTE from WASTE — no tax to produce it!!!! It is available to be used now!!!
- No high heat required to produce it like Metakaolin.
- Consistent powder from the same quarries — unlike Fly Ash.
- To be able to use this slate powder as a Geopolymer material solves huge problems for the Environmental Agency in Wales.
- Will help Government all over the world by using Slate Waste safely and INNOVATIVELY.
- Very natural Earth material can withstand heat to a temp of 1450°C before it turns to GLASS!!!



Some of the top countries for slate quarrying include

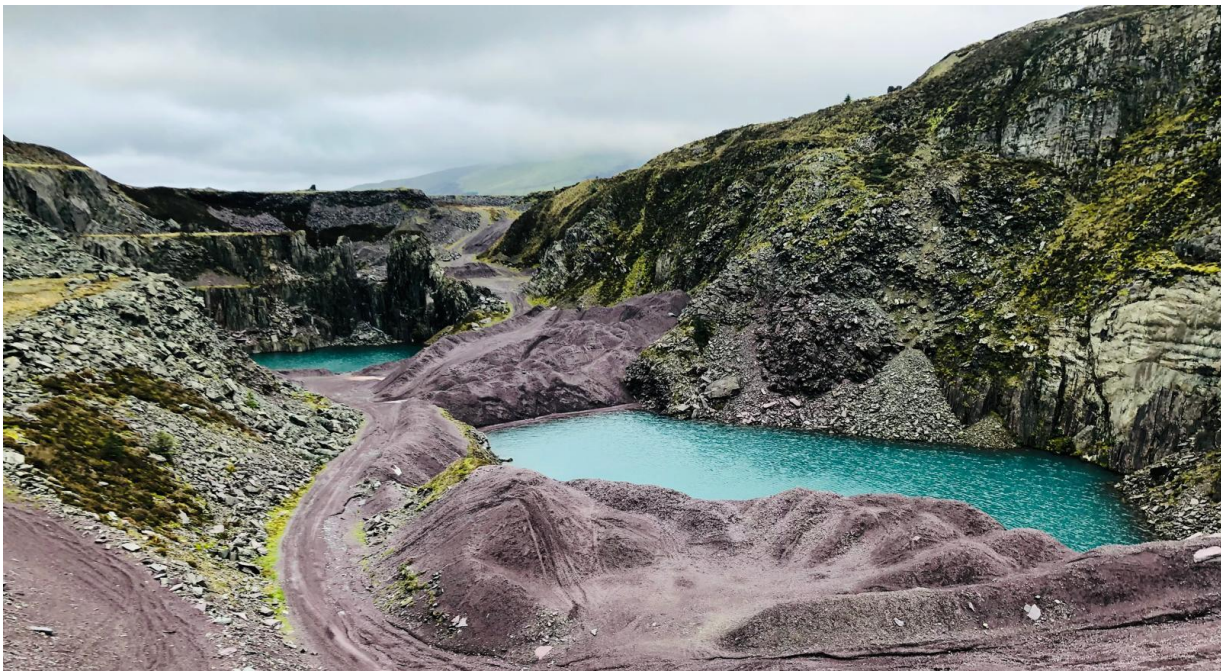
1. China
2. United States (e.g., Pennsylvania, Vermont)
3. **United Kingdom (e.g., Wales, Cornwall)**
4. Spain
5. Brazil
6. India

These countries have significant slate deposits and active quarries, supplying material for various uses like roofing, flooring, and construction



The Slate Landscape of
NorthWest Wales: Part of
the World Heritage Site





Slate can be a great material for making geopolymers

1. **Rich in aluminosilicates:** Slate contains aluminosilicate minerals, which are ideal for geopolymerisation reactions.
2. **High reactivity:** The fine-grained structure and mineral composition of slate can make it highly reactive, facilitating the geopolymerisation process.
3. **Abundant and accessible:** Slate is a relatively common rock, making it a potentially cost-effective source material.
4. **Durable properties:** Geopolymers made from slate can inherit its durability and resistance to weathering, making them suitable for various applications.

By utilising slate in geopolymer production, you can create sustainable, durable materials with potential uses in construction, infrastructure, and more.





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(71) Applicant(s):
Llechi Slate Evolution Cyf
10 Water Street, Penygroes, Caernarfon, Wales,
LL54 6LR, United Kingdom

(72) Inventor(s):
Richard Wyn Huws

(74) Agent and/or Address for Service:
Culverstons
20 Dawlish Road, Wirral, Merseyside, CH61 2XP,
United Kingdom

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(58) Field of Search:
INT CL B09B, C03C, C04B, G21F
Other: SEARCH-PATENT, SEARCH-NPL

(54) Title of the Invention: **Processes, compositions and uses of slate waste**
Abstract Title: **A glass composition formed from slate or slate waste**

(57) A glass formed from slate or slate waste is disclosed which comprises, in percent by weight:
45 to 65 percent SiO₂;
15 to 25 percent Al₂O₃;
8 to 15 percent Fe₂O₃; and
7 to 12 percent CaO.
The composition may be formed by adding limestone, magnesium carbonate and soda ash to slate or slate waste.
The composition may be formed into fibres. Also disclosed are a method of waste encapsulation using vitrified
slate/slate waste glass; a geopolymer composition characterised by slate/slate waste powder and alkaline activator;
and a geopolymer composition characterised by slate/slate waste powder and acidic activator.

Properties of Slate as a Geopolymer Material

- Chemical element: SLATE
- SLATE is a fine-grained, foliated metamorphic rock.
- It is derived from shale, mudstone, or siltstone, which are sedimentary rocks.
- Non-toxic
- Already processed as a material — no extra cost to mine and prepare it as a Geopolymer Material
- NO TAX
- The size of particles of the powder will make a difference to cure time and viscosity.
- Environmental issues
- Slate does not react to or with chemicals or acid, therefore it can be mixed with many different materials.
- It is a non-conductive material, so it could be an insulator = Geopolymer Insulator!!!

Chemical Composition of Slate Regions

NANTLLE VALLEY (SiO_2 62.18 wt.%)

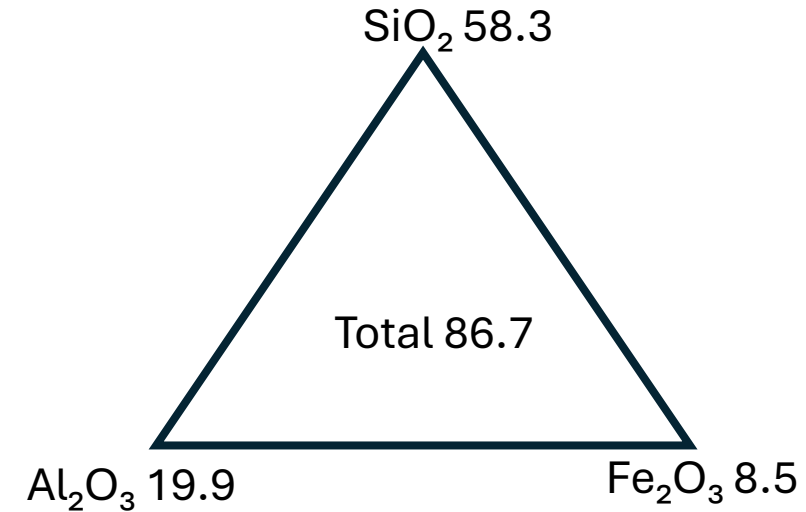
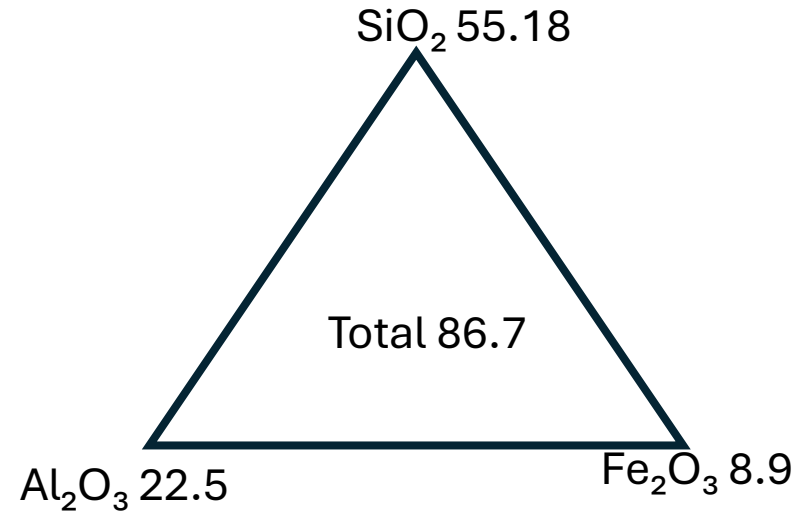
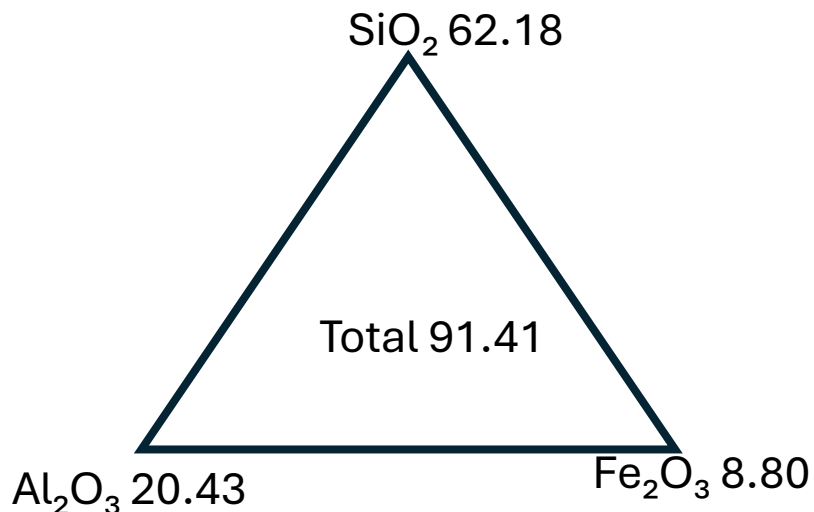
- Al_2O_3 20.43
- Fe_2O_3 8.80
- TiO_2 = 0.88
- CaO = 0.73
- Na_2O = 1.58
- MgO = 2.15
- K_2O = 3.24

BLAENAU (SiO_2 55.3 wt.%)

- Al_2O_3 22.5
- Fe_2O_3 8.9
- TiO_2 = 1.0
- CaO = 0.4
- Na_2O = 2
- MgO = 2.3
- K_2O = 3

BETHESDA (SiO_2 58.3 wt.%)

- Al_2O_3 19.9
- Fe_2O_3 8.5
- TiO_2 = 0.9
- Ca = 0.7
- Na = 0.9
- Mg = 1.9
- K_2O = 3

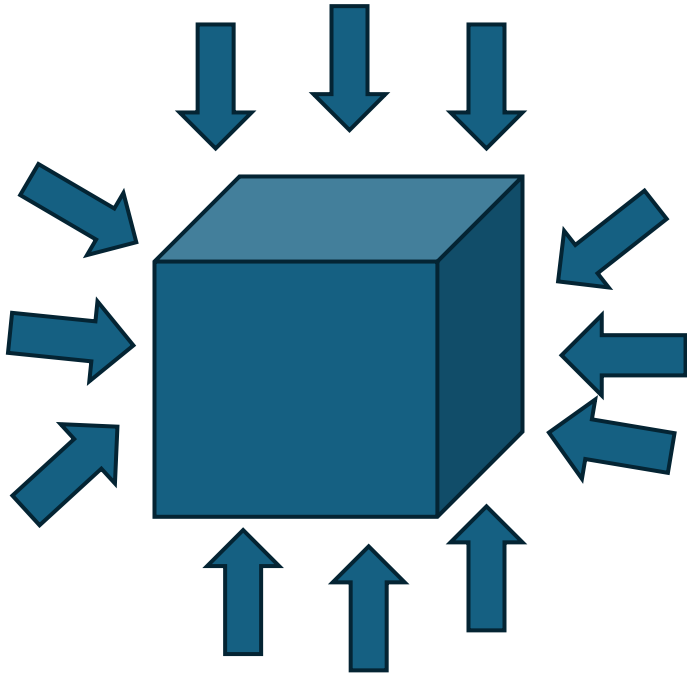


Thermal Properties and Microwave Curing

- It can withhold heat for a long time
- Great for insulation in the construction world or sector
- Density of Nantlle valley slate = $2.7 - 2.8 \text{ g/cm}^3$.
- The other secret of slate is that it can be cured via microwaves!!!
- Description of how this is possible
- BUT BE VERY VERY CAREFUL — small seconds at a time at less than $<300 \text{ watts}$ / 90 Watts good



Slate and Microwave Compatibility



- MICROWAVE is OK with igneous, sedimentary, and metamorphic rocks — and they can all react differently.
- Microwave heating works because polar molecules (like water) absorb microwave irradiation and convert it into thermal energy.
- When you heat with microwaves, it does not heat from the top or bottom, but all around.
- Depends on material volume
- 2 min – 10 min at 90 WATTS
- Microwave radiation is an efficient method to accelerate the geopolymer reaction.

Mixing and Optimisation

- A high-shear mixer is the best mixer for geopolymer preparation.
- It could be stated that the reaction between slate and alkaline or acid was optimized in the best possible manner.
- Calcium (Calcium Carbonate) needs to be added to get the best geopolymer from slate.

Experimental validation and practical applications



After 7 days of curing, we can conclude that the slate Geopolymer has a promising compressive strength value of at least 27 MPa. This will be a great chance/opportunity to use it in the construction world/sector.

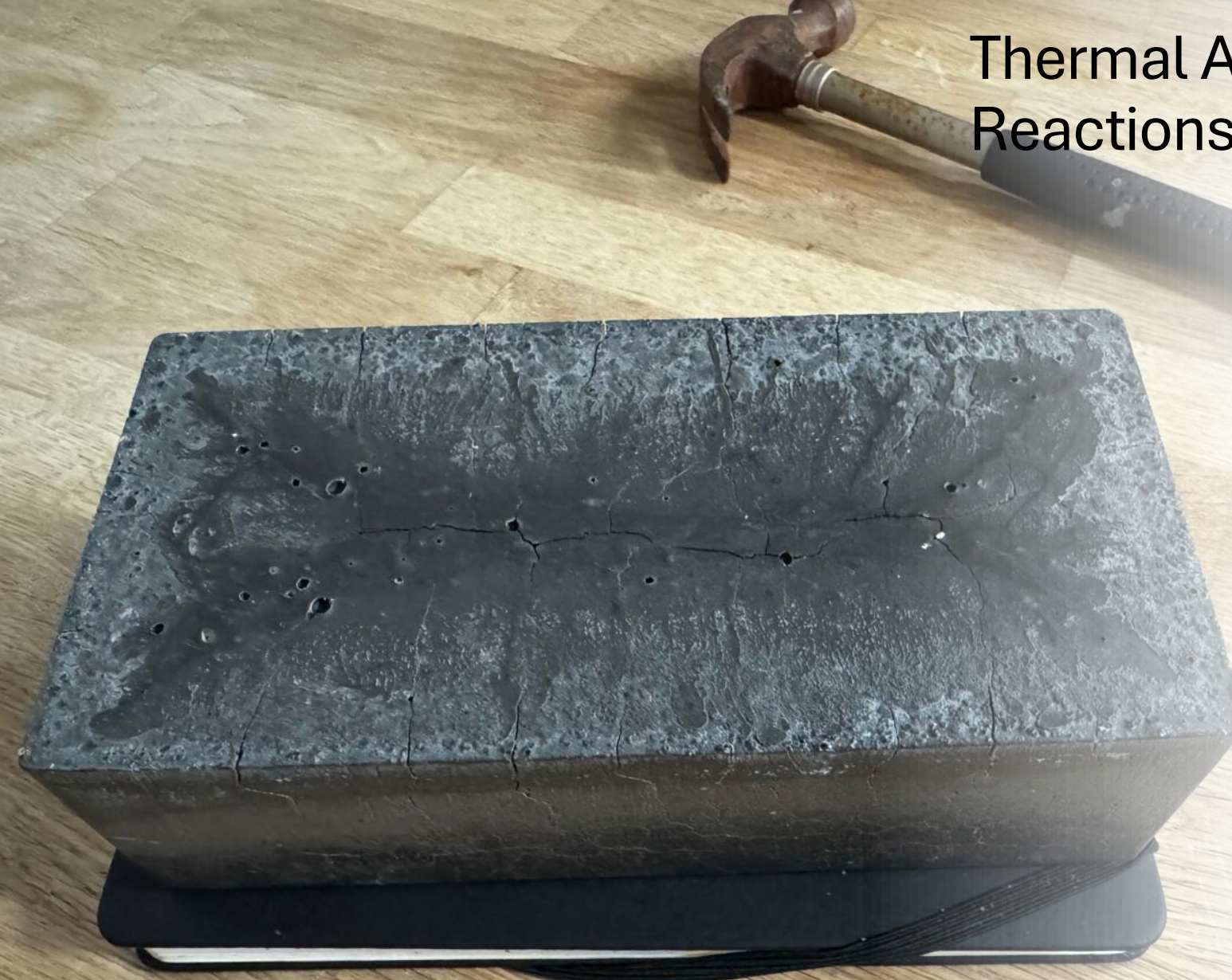


The interaction of a slate powder as an aluminosilicate from the circular economy with an alkali solution gives us a promising sustainable solution for managing otherwise problematic waste product to a new alternative cementitious and encapsulating material.



Thermal Acceleration of Geopolymer Reactions

- Through higher temperatures, the hydration reaction occurs faster, and the concrete sets within hours instead of days.
- Geopolymer curing occurs through a polycondensation reaction.
- Thermal treatment contributes to the faster evaporation of water.
- Heat increases the movement of atoms and molecules, making effective collisions more likely.
- Microwave heating results in less heat loss compared to conventional heating, as there is no heat transfer from an external source (like an oven) to the sample.
- The material heats faster and more effectively through energy conversion from electromagnetic to thermal energy.



Pyroxene and Microwave Effects

- PIROXENE
- The melting of minerals and the closure of original natural cracks mainly lead to the increase in compressive strength of pyroxene after microwave heating.
- With the increase in irradiation temperature, pyroxene fracture gradually transitions from:
 - Wire-like drawing
 - Step-like fracture
 - To water ripple-like fracture
- Pyroxene occurs predominantly in intermediate and mafic (llechi) igneous rocks and in high-grade metamorphic rocks.
- Quadrilateral pyroxenes are fundamental constituents of basalts (llechi as well?).

Exploration and Innovation

- Since this is a recently new material in the geopolymer world there are huge opportunities to experiment and explore.
- ENJOY THE JOURNEY. Especially in the modern technology world such as 3D Printing.
- Pob lwc, best of luck and Bonne chance
- Mwynhewch, Enjoy and Profitez bien.



CONTACT

PANT DU

LLECHI SLATE
EVOLUTION CYF

B BRENNTAG

GRINGO



Y Wern, County Road, Penygroes, Gwynedd,
WALES, UK. LL546HE
+447831 578516
richwyn65@icloud.com